

Utilization of Thiazolylacetonitriles in the Synthesis of Thiophene, Thiazole, Pyrazolo[1,5-*a*]pyrimidine and Pyrazolo[5,1-*c*]triazine Derivatives

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ABSTRACT: *Thiazolylcyanothioacetanilides react with α -haloketones and haloesters to give the corresponding thiophene or thiazole derivatives according to the reaction conditions. Pyrazolo[1,5-*a*]pyrimidines and pyrazolo[5,1-*c*]triazines were synthesized by reaction of 3-amino-4-(4'-arylthiazol-2'-yl)-5-phenylaminopyrazole with different reagents. Structures of the new compounds were confirmed by elemental analyses, spectral data, and alternative methods of synthesis whenever possible. © 1999 John Wiley & Sons, Inc. Heteroatom Chem 10: 508–516, 1999*

INTRODUCTION

The interesting pharmacological properties of thiophene, thiazole, pyrazolo[1,5-*a*]pyrimidine and pyrazolo[5,1-*c*]triazine derivatives [1] in relation to the various changes in the structures of these compounds are important in the synthesis of some less toxic and more potent drugs. The present investigation deals with the synthesis of some such types of compounds, in continuation of our studies in the chemistry of these heterocycles [2–6]. The syntheses of several new heterocyclic derivatives are described.

RESULTS AND DISCUSSION

Treatment of 2-(4'-phenyl)thiazolylacetonitrile (1a), phenyl isothiocyanate, and potassium hydrox-

ide in *N,N*-dimethylformamide with ω -bromoacetophenone afforded a product with the molecular formula $C_{26}H_{17}N_3S_2$. The IR (cm^{-1}) spectrum revealed a band at 2175 that was attributed to the presence of a cyano group, and there was no band between 3100 and 3500 or 1800 and 1650 because of the absence of each NH and CO groups [7]. Its 1H NMR spectrum showed only a signal at $\delta = 7.21$ – 7.88 (m, ArHs, and thiazole H-5), and the mass spectrum revealed a peak at $m/z = 435$. Based on these facts, the product was assigned as: 1-[2'-(3',4'-diphenyl)]thiazoline-2-cyano-2-(4'-phenyl)thiazolyethene (3a). However, Gewald et al. [8] reported that ω -bromoacetophenone reacted with 1a and phenyl isothiocyanate in the presence of potassium ethoxide to give 3-amino-2-benzoyl-5-phenylamino-4-(4'-phenyl)thiazol-2'-ylthiophene (4a). From the above data, product formation may be dependent on the reaction conditions (cf. Scheme 1). To clarify this situation, treatment of 2b with ω -bromoacetophenone in ethanol at room temperature gave product 5. The 1H NMR spectrum of 5 showed signals at $\delta = 2.35$ (s, 3H, 4- $CH_3C_6H_4$), 3.39 (s, 2H, SCH_2), 7.25–8.05 (m, 15H, ArHs, and thiazole H-5) and 11.94 (s, br, 1H, NH). Its IR (cm^{-1}) spectrum revealed bands at 3441 (NH), 2209 (CN), 1666 (CO), and 1624 (C=N). Compound 5 was converted to the thiophene 4b by boiling in ethanol containing triethylamine and to the thiazole 3b by treatment with polyphosphoric acid (Scheme 1).

Similarly, ω -bromoacetophenone, chloroacetone, and ethyl chloroacetate reacted with the ap-

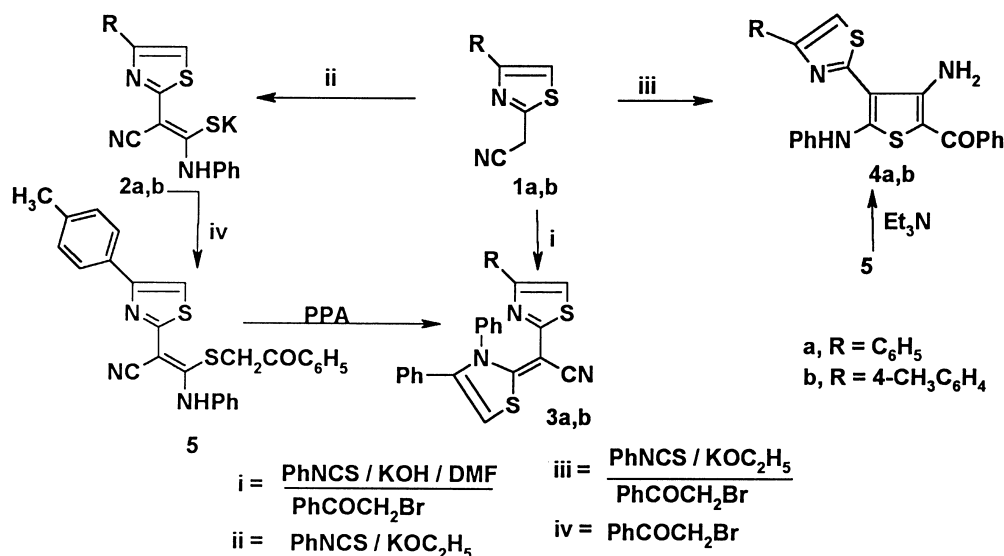
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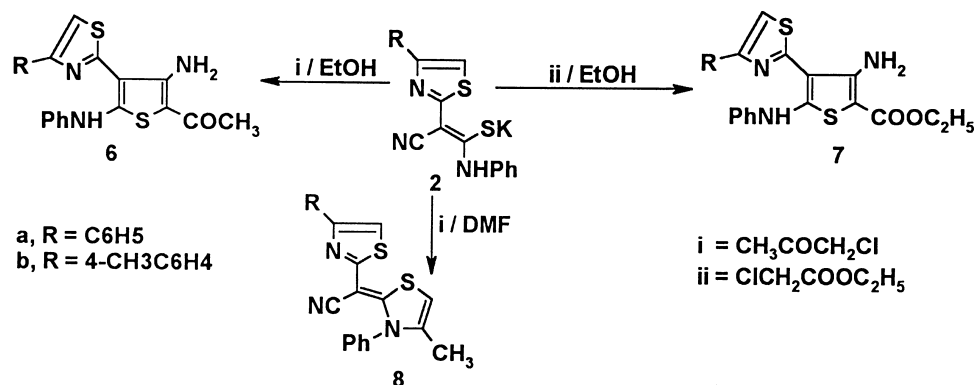
appropriate potassium 2-(4'-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates **2a,b** in ethanol to afford 3-amino-2-substituted-5-phenylamino-4-[4'-(aryl)thiazol-2'-yl]thiophenes **4**, **6**, and **7**, respectively, and *o*-bromoacetophenone and chloroacetone reacted with the appropriate **2a,b** in *N,N*-dimethylformamide to give the [(3,4-disubstituted)thiazolidene-2-yl](4-arylthiazol-2-yl)acetonitriles **3** and **8**, respectively (cf. Scheme 2).

On the other hand, the appropriate potassium 2-(4'-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates **2a,b** reacted with 3-chloropentan-2,4-dione in *N,N*-dimethylformamide solution to afford two products (cf. Scheme 3). The first product was identical in all respects (m.p., mixed m.p., and spectra) with the corresponding thiophene **6** and the second product was formulated as 2-[2'-(5'-acetyl-4'-methyl-3-phenyl)thiazolinyl]-2-[2'-(4'-substi-

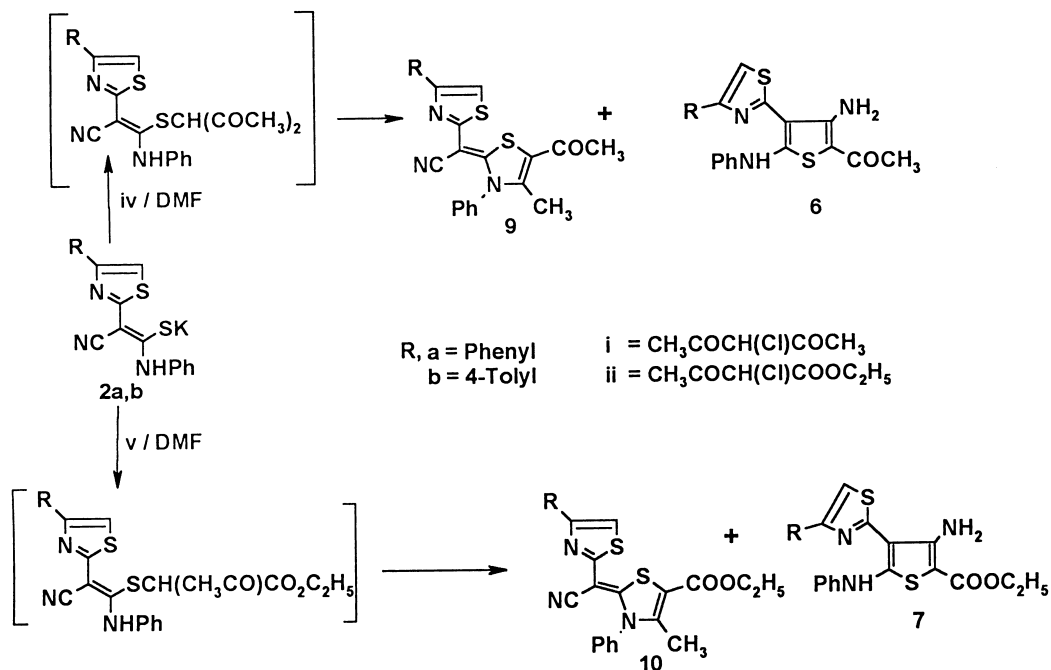
tuted)thiazolylcyanoethene **9**. The structure of **9** was confirmed on the basis of elemental analysis and spectral data. Thus, the ¹H NMR spectrum of **9a** showed signals at $\delta = 2.29$ (s, 3H, CH₃CO), 2.58 (s, 3H, thiazole CH₃), 7.26–7.70 (m, 10H, ArHs) and 8.20 (s, 1H, thiazole H-5). Its IR (cm⁻¹) spectrum revealed bands at 2191 (CN) and 1635 (CO conjugated), and no band was apparent near 3500–3100 because of the absence of the NH group. The formation of these products involves initial attack by one molecule of 3-chloropentan-2,4-dione on a molecule of the appropriate **2a,b** to give an intermediate, which cyclized to the final products, the thiophene **6** and the thiazole **9**. Similarly, ethyl 2-chloro-3-oxobutanoate reacted with the appropriate potassium 2-(4'-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates **2a,b** in *N,N*-dimethylformamide to give, in each case, the corresponding thiophenes **7** and



SCHEME 1



SCHEME 2



SCHEME 3

thiazoles **10**. Structures **7** and **10** were elucidated on the basis of elemental analyses, spectral data, and alternative methods of synthesis (cf. Experimental).

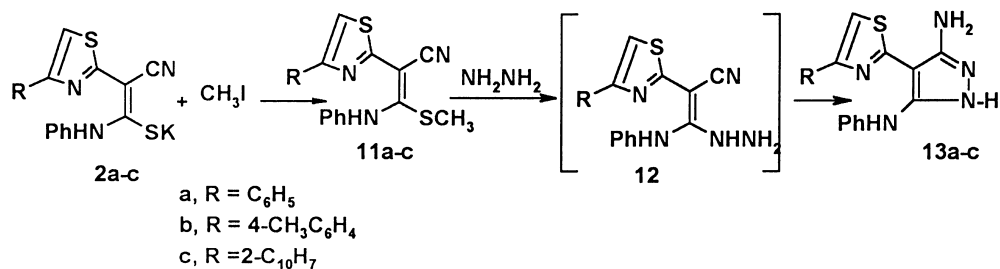
Also, treatment of the appropriate 2-(4-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates **2a-c** reacted with methyl iodide to give the corresponding *S*-methyl derivative **11a-c**. The IR (cm^{-1}) spectrum of **11** revealed bands near 3350 (NH), 2200 (CN), and 1610 (C=N). The ^1H NMR spectrum of **11a** showed signals at $\delta = 2.7$ (s, 3H, SCH_3), 7.36–8.22 (m, 11H, ArHs, and thiazole H-5) and 11.21 (s, br., 1H, NH). More information on the structure **11** came from its reaction with hydrazine hydrate in ethanol that was accompanied by the evolution of methanethiol and the conversion to the aminopyrazoles. The IR spectra of the products showed bands attributable to the NH_2 group and the absence of any absorption bands due to the nitrile group. Based on the above data, the products can be formulated as 3-amino-4-[2'-(4'aryl)thiazolyl]-5-(phenylamino)pyrazoles **13**. The formation of **13** proceeded most likely via the intermediacy of the corresponding 2-hydrazino derivatives **12**, which cyclized via an intramolecular addition of the hydrazine group to the nitrile function to afford the final product **13** (cf. Scheme 4).

Treatment of the appropriate **13a-c** with pentane-2,4-dione in boiling glacial acetic acid gave the corresponding 2,4-dimethylpyrazolo[1,5-*a*]pyri-

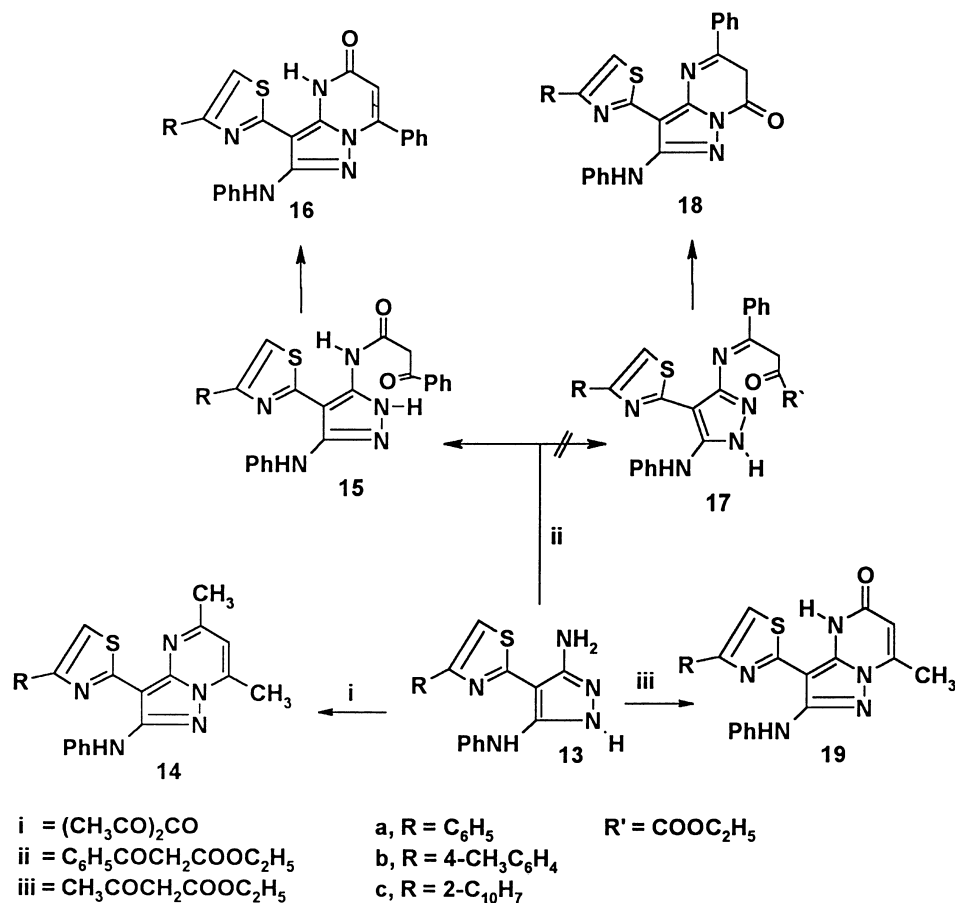
midines **14a-c**, respectively (cf. Scheme 5). The structure of each **14** was elucidated on the basis of elemental analyses and spectral data. Thus, the IR (cm^{-1}) spectra of each compound **14a-c** revealed bands near 3265–3275 (NH) and 1640–1630 (C=N). The ^1H NMR spectrum of **14a** showed signals at $\delta = 2.44$ (s, 3H, CH_3), 2.67 (s, 3H, CH_3), 6.61 (s, 1H, pyrimidine H-5), 7.31–7.82 (m, 10H, ArHs), 8.31 (s, 1H, thiazole H-5), and 11.32 (s, br., 1H, NH).

Unequivocal support for each structure **13** came from its reaction with β -ketoesters and β -ketoanilides in boiling acetic acid. Thus, the reaction of ethyl benzoylacetate or benzoylacetanilide with the appropriate aminopyrazole **13** afforded the identical product in all respects (m.p., mixed m.p., and spectra). The structure of the product could have been one of structures **15–18**. On the basis of the ^1H NMR spectrum, the structure **17** was eliminated because no signals attributable to the ethoxy group were evident. Thus, the reaction took place through the elimination of ethanol (or aniline) to give **15**, which cyclized to the pyrazolo[1,5-*a*]pyrimidine **16** by its treatment with concentrated sulfuric acid or by boiling with ethanolic piperidine solution (cf. Scheme 5).

By analogy, the reaction of the appropriate of 3-amino-4-(4'-arylthiazol-2'-yl)-5-(phenylamino)pyrazoles **13a-c** with ethyl 3-oxobutanoate (or acetacetanilide) in boiling acetic acid produces **19a-c**.



SCHEME 4



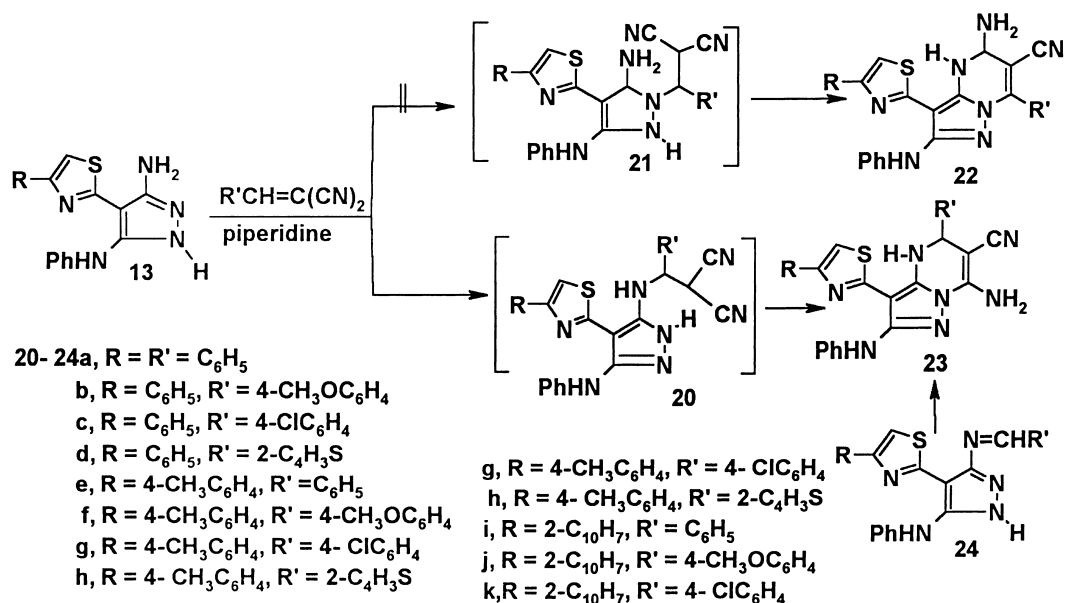
SCHEME 5

The structure 19 was elucidated on the basis of the elemental analysis and spectral data. Thus, the ¹H NMR spectrum of 19a showed signals at δ = 2.67 (s, 3H, CH₃), 6.61 (s, 1H, pyrimidine H-5), 7.31–7.82 (m, 10H, ArHs), 8.32 (s, 1H, thiazole H-5) and 11.82 (s, br., 2H, 2NH). Its IR (cm⁻¹) spectrum revealed bands at 3419 (NH), 1668 (CO), and 1620 (C–N).

Also, the appropriate 13a–c reacted with the appropriate 1-cyano-2-substituted acrylonitriles in ethanol containing piperidine as a catalyst to afford a

single product, in each case, according to thin-layer chromatography (TLC). The structure of each product was confirmed on the basis of elemental analyses, spectral data, and the alternative method of synthesis by reaction of the Schiff's base 24 with malononitrile (cf. Scheme 6).

Meanwhile, the appropriate diazonium chlorides 25a,b were coupled with active methylene compounds such as acetylacetone, malononitrile, ethyl cyanoacetate, and ethyl 3-oxobutanoate in



SCHEME 6

ethanolic sodium acetate solution to afford pyrazolo[5,1-*c*][1,2,4]triazines 26–29, respectively (cf. Scheme 7). The structures of 26–29 were established on the basis of elemental analysis and spectral data. Thus, the ¹H NMR spectrum of 26a showed signals at δ = 2.25 (s, 3H, CH₃), 2.67 (s, 3H, CH₃), 7.31–8.31 (m, 11H, ArHs and thiazole H-5), and 9.52 (s, br., 1H, NH). Its mass spectrum revealed peaks at m/z = 426 (100%), 316 (12.9%), 134 (54.7%) and 77 (27%).

Similarly, the appropriate diazonium chlorides 25a,b coupled with 3-chloropentan-2,4-dione and ethyl 2-chloro-3-oxobutanoate in cold ethanolic sodium acetate solution to give products 30a,b and 31a,b, respectively. Structures 30 and 31 were confirmed on the basis of spectral data and elemental analyses. The mass spectrum of 30a, for example, revealed peaks at m/z = 420 (M-H₂O), 37.7%, and 418 (M-H₂O), 66.9%.

EXPERIMENTAL

All melting points were determined on an electrothermal apparatus and are uncorrected. The IR spectra were recorded (KBr discs) on a Shimadzu FT-IR 8201 PC spectrophotometer. The ¹H NMR spectra were recorded in CDCl₃ and (CD₃)₂SO on a Varian Gemini 200 MHz spectrometer, and chemical shifts were expressed in δ units using TMS as an internal reference. The MS spectra were recorded on a GC-MS QP1000 EX Shimadzu. Elemental analyses were carried out at the Microanalytical Center of the Uni-

versity of Cairo, Giza, Egypt. The 2-(4'-aryl)thiazolylacetonitrile 1a,c and potassium 2-(4'-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates 2a,2c were prepared as previously reported [6,9].

2-(4'-*P*-tolyl)thiazolylacetonitrile (1b)

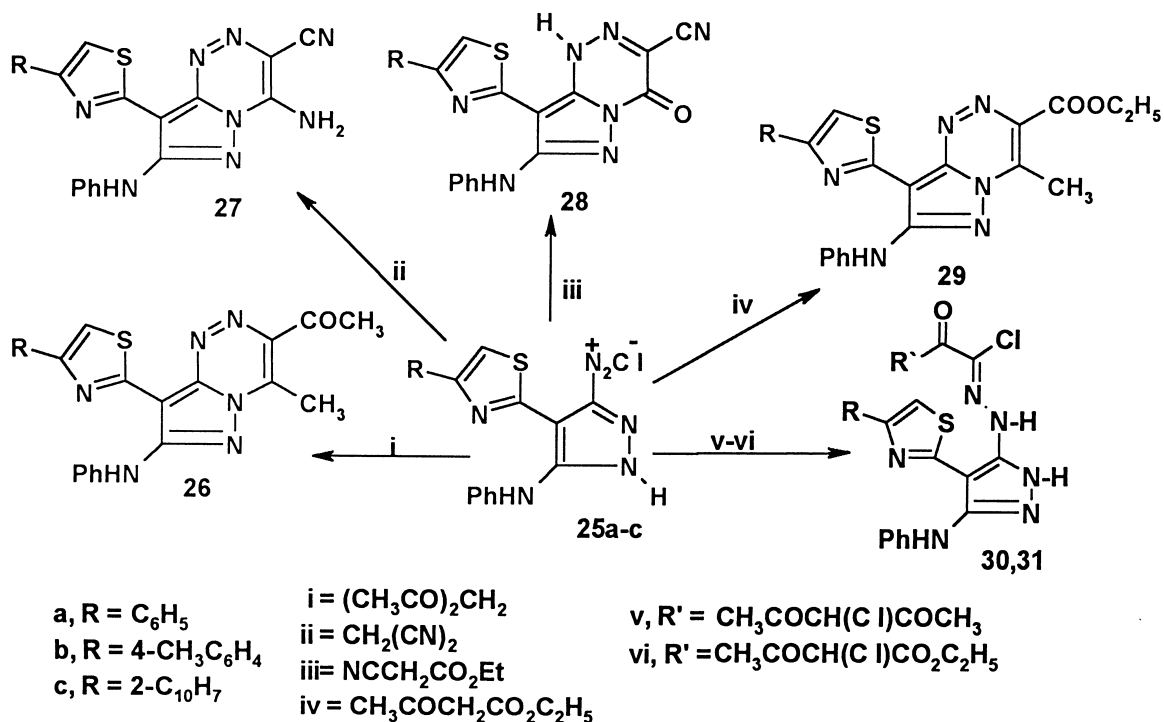
Equimolar amounts of ω -bromo-*p*-methylacetophenone (21.3 g, 0.01 mol) and cyanothioacetamide (10 g, 0.1 mol) in ethanol (50 mL) were refluxed for 30 minutes. The reaction mixture was cooled and poured onto ice cold water containing two drops of ammonium hydroxide (100 mL). The resulting solid was collected and crystallized from ethanol to give thiazole 1b, in a 65% yield (cf. Table 1).

Potassium 2-(4'-*p*-tolylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates 2b

Potassium ethoxide solution, which was prepared via reaction of potassium metal (1 g) in absolute ethanol (10 mL), was added to the mixture of the thiazole 1b (2.15 g, 0.01 mol) and phenyl isothiocyanate (1.35 g, 0.01 mol) in ethanol (20 mL) with stirring. The resulting solid was collected and washed with diethyl ether.

1-[2'-(3',4'-Disubstituted)]thiazoline-2-cyano-2-(4'-aryl)thiazolyethene 3a,b and 8a,b

A mixture of each appropriate 2-cyanomethylthiazole 1a,b, phenyl isothiocyanate, and potassium hydroxide (0.05 mol, each) in *N,N*-dimethylformamide



SCHEME 7

was stirred for 4 hours at room temperature. The ω -bromoacetophenone or chloroacetone (0.05 mol) was added, and stirring was continued for 2 hours. The resulting solid, after dilution with water, was collected and crystallized from dimethylformamide to give the corresponding thiazoles **3a,b** and **8a,b**, respectively, in a 62–64% yield (cf. Tables 1 and 2).

3-Amino-5-phenylamino-2-substituted-4-[4'-(aryl)thiazol-2'-yl]thiophenes **4a,b** and **6a,b**

The appropriate ω -bromoacetophenone, chloroacetone, or ethyl chloroacetate (0.05 mol) was added to the appropriate potassium salts **2a,b** (0.05 mol) in ethanol (20 mL) with stirring. The solid that formed after 4 hours was collected and then crystallized from a proper solvent to give the thiophenes **4a,b** and **6a,b**, in 60–62% yields, respectively (cf. Tables 1 and 2).

Reaction of **2a,b** with 3-Chloropentane-2,4-dione and Ethyl 2-chloro-3-oxobutanoate

3-Chloropentane-2,4-dione or ethyl 2-chloro-3-oxobutanoate was added to the appropriate thiazoles **2a,b** (5 mmol each) in *N,N*-dimethylformamide (20 mL) with stirring at room temperature for 4 hours and left overnight. The resulting solid was collected

by filtration, and fractional crystallization gave [3',4',5'-tri-substituted]thiazolidene-2-yl[(4-arylthiazol-2-yl)cyanoethene] **9a,b** or **10a,b** and thiophenes **6a,b** or **7a,b** in 30–50% yields (cf. Tables 1 and 2).

Synthesis of **5**

A mixture of ω -bromo-4-methylphenacyl bromide (1.01 g, 0.005 mol) and potassium 2-(4'-*p*-tolylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolate (**2b**) (1.92 g, 0.005 mol) in ethanol (20 mL) was stirred for 1 hour at room temperature. The resulting solid was collected and crystallized from benzene to give compound **5** in a 68% yield (cf. Tables 1).

Cyclization of **5**: Formation of 3-Amino-2-benzoyl-5-phenylamino-4-(4'-*p*-tolyl)thiazol-2'-ylthiophene (**4b**)

To a solution of **5** (1 g) in ethanol (20 mL), triethylamine (0.5 mL) was added, and the reaction mixture was refluxed for 30 minutes. The solid, so formed, was collected and crystallized from benzene to give the corresponding thiophene **4b** (cf. Tables 1 and 2).

Cyclization of **5**: 1-[2'-(3',4'-Diphenyl)thiazoline-2-cyano-2-(4'-*p*-tolyl)thiazolyethene (**3b**)

Compound **5** (1 g) was mixed with polyphosphoric acid, which was prepared by dissolving P₂O₅ (1 g) in

TABLE 1 Characterization Data of the Newly Synthesized Compounds

Compd No.	M.P. (°C) color	Compd No.	M.P. (°), color	Compd No.	M.P. (°) color	Compd No.	M.P. (°C) color
1b	103–105 Brown	11b	156–158 Yellow	23c	>320 Yellow	24h	258–260 Yellow
3a	28–283 Yellow	11c	215–217 [9] Yellow	23d	>320 Orange	24i	248–250 Brown
3b	255–257 Yellow	13a	220–222 White	23e	>320 Yellow	24j	253–255 Yellow
4a	198–199 [8] Yellow	13b	216–218 White	23f	>320 Yellow	24k	289–290 Yellow
4b	204–206 Yellow	13c	226–228 [9] White	23g	>320 Yellow	26a	241–243 Black
5	152–153 Yellow	14a	233–235 Yellow	23h	244–246 Red	26b	176–178 Black
6a	205–207 White	14b	255–257 Yellow	23i	>320 Yellow	27a	>320 Dark green
6b	206–207 White	14c	282–284 Yellow	23j	>320 Yellow	27b	246–248 Black
7a	176–178 White	15c	148–150 Green	23k	>320 Yellow	28a	255–257 Dark green
7b	175–176 White	16a	238–240 Green	24a	230–231 Yellow	28b	258–260 Black
8a	245–247 Yellow	16b	263–265 Green	24b	212–214 Yellow	29a	308–310 Red
8b	263–265 Yellow	16c	265–267 Green	24c	263–265 Yellow	29b	240–243 Black
9a	271–273 Yellow	19a	332–334 White	24d	225–227 Brown	30a	182–184 Black
9b	>320 Yellow	19b	>325 White	24e	243–245 Orange	30b	238–240 Black
10a	272–274 Orange	19c	>325 White	24f	253–255 Yellow	31a	172–174 Dark red
10b	280–282 Orange	23a	>320 Yellow	24g	265–266 Yellow	31b	213–215 Red
11a	196–198 Yellow	23b	>320 Yellow				

Crystallization solvents: a = acetic acid; b = benzene; c = *N,N*-dimethylformamide; d = dioxin; e = ethanol. Microanalytical data are satisfactory: $\pm 0.2\%$.

ortho-phosphoric acid (3 mL; 85%), and heated at 110°C for 1 hour. The reaction mixture was poured onto ice-cold water (30 mL), and the resulting solid was collected and crystallized to give the corresponding thiazole **3b** (cf. Tables 1 and 2).

1-Cyano-1-(4'-substituted)thiazol-2'-yl-2-phenylamino-2-thiomethylethene 11a–c

Methyl iodide (0.71 g, 0.005 mol) was added to the appropriate 2-(4-arylthiazol-2'-yl)-2-cyano-1-(phenylamino)ethenylthiolates **2a–c** (0.005 mol) in *N,N*-dimethylformamide (20 mL) with stirring. The reaction mixture was stirred for 1 hour, and the resulting solid was collected and crystallized from acetic acid to give products **11a–c** in 72–75% yields, respectively (cf. Tables 1 and 2).

3-Amino-4-(4'-substituted)thiazol-2'-yl-5-phenylaminopyrazoles 13a–c

A mixture of the appropriate **11a–c** (0.01 mol) and hydrazine hydrate (5 mL, 0.02 mol) in ethanol (20 mL) was refluxed for 6 hours. The resulting solid was collected and crystallized from ethanol (or dioxane) to give the corresponding aminopyrazoles **13a–c** in 66–68% yields, respectively (cf. Tables 1 and 2).

*3-(4'-Aryl)thiazol-2'-5,7-disubstituted-2-phenylaminopyrazolo[1,5-*a*]pyrimidines 14a–c, 16a–c, and 19a–c*

A mixture of the appropriate 3-aminopyrazoles **13a–c** (5 mmol) and the appropriate pentane-2,4-dione or

ethyl 3-oxo-4-phenylpropanoate (or benzoylacetanilide) or ethyl 3-oxobutanoate (or acetoacetanilide) (0.005 mol) in acetic acid (20 mL) was refluxed for 3 hours. The solid was collected and crystallized from the proper solvent to give the corresponding pyrazolo[1,5-*a*]pyrimidine **14a–c**, **16a–c**, and **19a–c** in 70–75% yields, respectively (cf. Tables 1 and 2). In the reaction of **13c** with 3-oxo-4-phenylpropanoate, the filtrate was diluted with water and **15c** was isolated.

Synthesis of Schiff's Bases 24a–k

General Procedure. A mixture of the appropriate aminopyrazoles **13a–c** and the appropriate aldehyde (0.005 mol each) in ethanol (20 mL) containing 3 drops of piperidine was refluxed for 4 hours. The resulting solid was collected and crystallized from the proper solvent to give the products **26a–k** in 70–75% yields, respectively (cf. Tables 1 and 2).

*7-Amino-3-(4-aryl)thiazol-2'-yl-6-cyano-5-substituted-2-phenylaminopyrazolo[1,5-*a*]pyrimidines 23*

General Procedure. Method A: Equimolar amounts of the appropriate aminopyrazoles **13a–c**, 1-cyano-1-substituted acrylonitrile (0.005 mol each), and 3 drops of piperidine in ethanol (20 mL) were refluxed for 4 hours. The resulting solid was collected and crystallized from the proper solvent to give 7-amino-3-(4'-aryl)thiazol-2'-yl-6-cyano-5-substituted-2-phenylaminopyrazolo[1,5-*a*]pyrimidines

TABLE 2 IR and ¹H NMR Spectra of the Newly Synthesised Compounds

Compd.	IR (cm ⁻¹)	¹ H NMR (δ)
3b	2175 (CN)	2.44 (s, 3H, 4-CH ₃ C ₆ H ₄) and 7.22–7.92 (m, 16H, ArHs, and thiazole)
4b	3433 (NH ₂), 1650 (CO) and 1600 (C=C)	2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 7.22–7.92 (m, 17H, ArHs, thiazole H-5, and NH ₂) and 11.94 (s, 1H, NH)
6a	3247 (NH), 1635 (CO)	2.28 (s, 3H, CH ₃), 7.12–7.19 (m, 13H, ArHs), thiazole H-5, and NH ₂) and 11.94 (s, 1H, NH)
6b	3247 (NH), 1640 (CO)	2.28 (s, 3H, CH ₃), 2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 7.12–7.19 (m, 12H, ArHs, thiazole H-5, and NH ₂) and 11.94 (s, 1H, NH).
7a	3408, 3311 (NH ₂), 1715 (CO).	1.13 (t, 3H, CH ₂ CH ₃), 4.22 (q, 2H, CH ₂ CH ₃), 7.22–7.72 (m, 13H, ArHs, thiazole H-5, and NH ₂) and 11.85 (s, 1H, NH).
7b	3396, 3319 (NH ₂) and 1733 (CO).	1.13 (t, 3H, CH ₂ CH ₃), 2.45 (s, 3H, CH ₃) 4.22 (q, 2H, CH ₂ CH ₃), 7.22–7.72 (m, 12H, ArHs, thiazole H-5, and NH ₂) and 11.85 (s, 1H, NH)
8a	2177 (CN)	2.57 (s, 3H, CH ₃), 7.21–7.82 (m, 12H, ArHs, and thiazole).
8b	2171 (CN)	2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 2.57 (s, 3H, CH ₃) and 7.22–7.86 (m, 11H, ArHs and thiazole)
9b	2191 (CN) and 1635 (CO).	2.29 (s, 3H, CH ₃), 2.42 (s, 3H, 4-CH ₃ C ₆ H ₄), 2.59 (s, 3H, CH ₃), 7.26–7.70 (m, 9H, ArHs), and 8.64 (s, 1H, thiazole)
10a	2189 (CN), 1703 (CO), and 1606 (C=N)	1.44 (t, 3H, CH ₂ CH ₃), 2.21 (s, 3H, CH ₃), 4.35 (q, 2H, CH ₂ CH ₃), and 7.27–7.39 (m, 11H, ArHs)
10b	2183 (CN), 1708 (CO), and 1608 (C=N)	1.44 (t, 3H, CH ₂ CH ₃), 2.21 (s, 3H, CH ₃), 2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 4.35 (q, 2H, CH ₂ CH ₃), and 7.27–7.39 (m, 10H, ArHs)
11b	3350 (NH), 2230 (CN), and 1610 (C=N)	2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 2.7 (s, 3H, SCH ₃), 7.36–8.22 (m, 10H, ArHs, and thiazole H-5) and 11.21 (s, br., 1H, NH).
13a	3199 3122 (NH ₂) and 1618 (C=N)	5.62 (s, 2H, NH ₂), 7.21–7.772 (m, 11H, ArHs), 8.22 (s, br., 1H, NH), and 8.45 (s, br., 1H, NH)
13b	3199, 3122 (NH ₂) and 1618 (C=N)	2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 5.62 (s, 2H, NH ₂), 7.21–7.77 (m, 10H, ArHs), 8.22 (s, br., 1H, NH), and 8.45 (s, br., 1H, NH).
14b	3276 (NH) and 1626 (C=N)	2.44 (s, 3H, CH ₃), 2.45 (s, 3H, 4-CH ₃ C ₆ H ₄), 2.67 (s, 3H, CH ₃), 6.61 (s, 1H, pyrimidine H-4), 7.22–7.77 (m, 10H, AHS, and thiazole), and 8.34 (s, br., 1H, NH)
14c	3276 (NH) and 1626 (C=N)	2.44 (s, 3H, CH ₃), 2.67 (s, 3H, CH ₃), 6.61 (s, 1H, pyrimidine H-4), 7.22–7.77 (m, 13H, AHS, and thiazole) and 8.34 (s, br., 1H, NH)
19b	3417 (NH), 3186 (NH), 1666 (CO), and 160 (C=N).	2.44 (s, 3H, 4-CH ₃ C ₆ H ₄), 2.67 (s, 3H, CH ₃), 6.61 (s, 1H, pyrimidine H-5), 7.31–7.82 (m, 9H, ArHs), 8.32 (s, 1H, thiazole H-5), and 11.82 (s, br., 2H, 2NH).
19c	3417 (NH), 3186 (NH), 1666 (CO), and 1604 (C=N).	2.67 (s, 3H, CH ₃), 6.6 (s, 1H, pyrimidine H-5), 7.31–7.82 (m, 12H, ArHs), 8.32 (s, 1H, thiazole H-5), and 11.82 (s, br., 2H, 2NH)
23b	3448, 3292, 3199 (NH, NH ₂), 2216 (CN).	2.74 (s, 1H, CH), 2.89 (s, 1H, NH), 3.87 (s, 3H, OCH ₃), 7.00–8.05 (m, 15H, ArHs, and thiazole), 8.98 (s, 2H, NH ₂), and 10.14 (s, 1H, NH)
23f	3448, 3292, 3199 (NH, NH ₂), 2216 (CN)	2.45 (s, 3H, 4-CH ₃ C ₆ H ₅), 2.74 (s, 1H, CH), 2.89 (s, 1H, NH), 3.80 (s, 3H, CH ₃ OC ₆ H ₄), 7.00–8.05 (m, 14H, ArHs and thiazole), 8.98 (s, 2H, NH ₂), and 10.14 (s, 1H, NH)
24b	3261 (NH).	3.85 (s, 3H, OCH ₃), 7.21–8.03 (m, 17H, ArHs thiazole and NH), and 8.52 (s, 1H, CH=N)
24e	3273 (NH).	2.42 (s, 3H, CH ₃), 7.21–8.03 (m, 17H, ArHs, thiazole, and NH) and 8.52 (s, 1H, CH=N)
24f	3265 (NH).	2.42 (s, 3H, CH ₃), 3.85 (s, 3H, OCH ₃), 7.21–8.03 (m, 16H, ArHs, thiazole, and NH) and 8.52 (s, 1H, CH=N)
29a	3350 (NH), 1720 (CO), and 1611 (C=N)	1.03 (t, 3H, CH ₂ CH ₃), 1.95 (s, 3H, CH ₃), 4.22 (q, 2H, CH ₂ CH ₃), 7.01–7.75 (m, 10H, ArHs), 8.31 (s, 1H, thiazole H-5), and 9.57 (s, br., 1H, NH)

23a–k in 73–75% yields, respectively (cf. Tables 1 and 2).

Method B. Equimolar quantities of each the appropriate aminopyrazoles **13a–c** and the appropriate **24a–j** (5 mmol each) in ethanol (20 mL) containing 3 drops of piperidine as a catalyst was refluxed for 4 hours. The resulting solid was collected and crystallized to give products identical in all respects (m.p., mixed m.p., and spectra) with those corresponding in method A.

3,4-Disubstituted-8(4'-aryl)thiazol-2'-yl-7-phenylaminopyrazolo[5,1-c]-1,2,4-triazines 26–29, and hydrazonoyl chlorides 31a,b and 32a,b

The appropriate aminopyrazolediazonium chlorides, **25a,b** (ca. 0.01 mol) which were prepared by adding concentrated hydrochloric acid (3 mL, 12 M) to a cold solution of the appropriate aminopyrazole **13a,b** (0.01 mol) in acetic acid (2 mL) followed by treatment with a cold solution of sodium nitrite (0.7 g, 0.01 mol) in water (5 mL), was added dropwise with stirring at 0–5°C to a cold solution of each of acetylacetone or malononitrile or ethyl cyanoacetate or ethyl 3-oxobutanoate or 3-chloropentane-2,4-di-

one or ethyl 2-chloro-3-oxobutanoate (0.01 mol) in ethanol (50 mL) containing sodium acetate trihydrate (1.3 g, 0.01 mol). The reaction mixture was stirred for 3 hours, and the precipitated was filtered off, washed with water, dried, and crystallized from acetic acid (or DMF) to give **29,31–32** in 63–65% yields, respectively (cf. Tables 1 and 2).

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