

Green Hydrogen

In the framework of the postgraduate course "Renewable Energy"



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Introduction

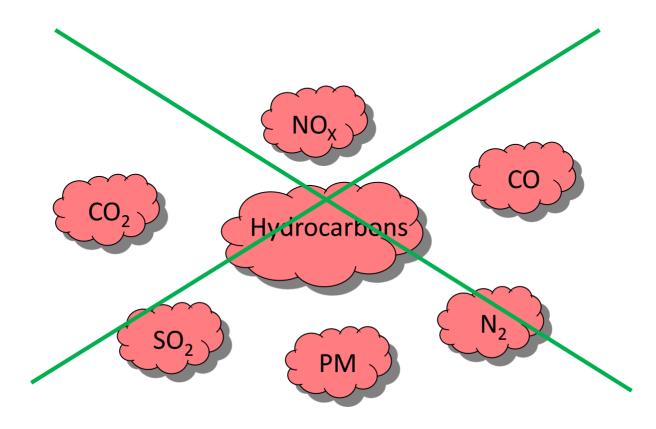


Environmental Impact



The combustion of hydrogen with oxygen produces water as its only product:

$$2H_2 + O_2 \rightarrow 2H_2O$$





Green Hydrogen Systems (GHS)

Several countries have set for themselves the target of reducing CO₂ emissions

To achieve this goal with the greatest possible efficiency, more key technologies are needed in addition to the expansion of renewable energy generation and the electricity infrastructure

The conversion of electricity generated by renewable energies into hydrogen (H₂)—power-to-gas—is such a key to a successful energy transition

Source: GET H₂ initiative

The key concept:

- 1. Electricity from renewable energies (wind and sun) is converted into H₂
- 2. H₂ is transported via a separate infrastructure, for whose development the existing gas infrastructure is used
- 3. In the industry, transport and heating sectors, green H_2 is used as a CO_2 -free raw material, fuel and energy source
- 4. In industry, it replaces, among other things, fossil generated H₂
- 5. Green H₂ is used as the basis for the production of e-fuels, especially for aircraft and rail vehicles
- 6. H₂ not directly used is efficiently stored in existing underground salt caverns
- 7. H₂ can be used for re-electrification, e.g. via a gas turbine

Source: GET H₂ initiative



Hydrogen Fuel:

- 1. Hydrogen fuel is a zero-emission fuel burned with oxygen.
- 2. It can be used in fuel cells or internal combustion engines.
- 3. H₂ is usually considered an energy carrier as it must be produced from a primary energy source such as solar energy, biomass, electricity, or hydrocarbons (e.g. natural gas or coal).
- 4. Recently, the majority of H_2 (~95%) is produced from fossil fuels by steam reforming or partial oxidation of methane and coal gasification with only a small quantity by alternative routes such as biomass gasification or electrolysis of water or solar thermochemistry, a solar fuel with no carbon emissions.





Disadvantages:

- 1. Hydrogen fuel is hazardous because of the low ignition energy and high combustion energy of hydrogen, and because it tends to leak easily from tanks.
- 2. Explosions at hydrogen filling stations have been reported.
- 3. Hydrogen fuelling stations generally receive deliveries of hydrogen by truck from hydrogen suppliers. An interruption at a hydrogen supply facility can shut down multiple hydrogen fuelling stations.

Internal combustion engine conversions to hydrogen:

- 1. Combustion engines in commercial vehicles have been converted to run on a hydrogen-diesel mix, where up to 70% of emissions have been reduced during normal driving conditions.
- 2. Minor modifications are needed to the engines, as well as the addition of hydrogen tanks at a compression of 350 bars.
- 3. Trials are now underway to test the efficiency of the 100% conversion of a Volvo FH16 heavy-duty truck to use only hydrogen. The range is expected to be 300 km/ 17 kg; which means an efficiency better than a standard diesel engine (where the embodied energy of 1 gallon of gasoline is equal to 1 kilogram of hydrogen).

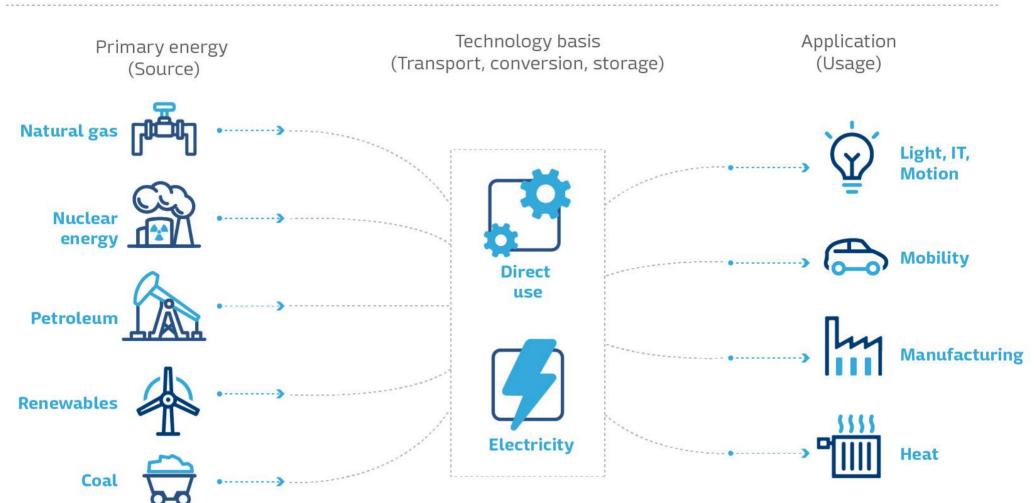
(Dalagan, 2018; Martensson, 2018)





Energy system today

Despite the expansion of renewable energies, the **energy system** in **Germany** continues to feed itself largely from non-renewable sources.



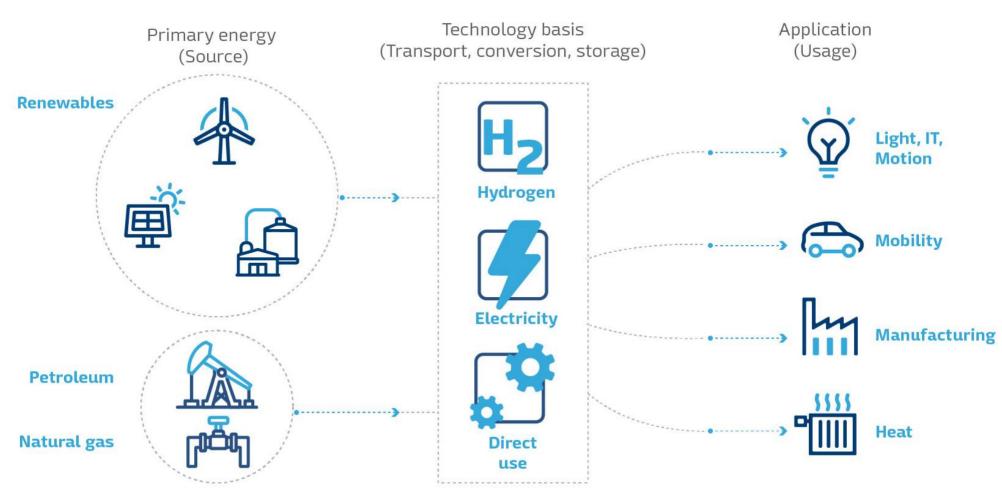
Source: GET H₂ initiative





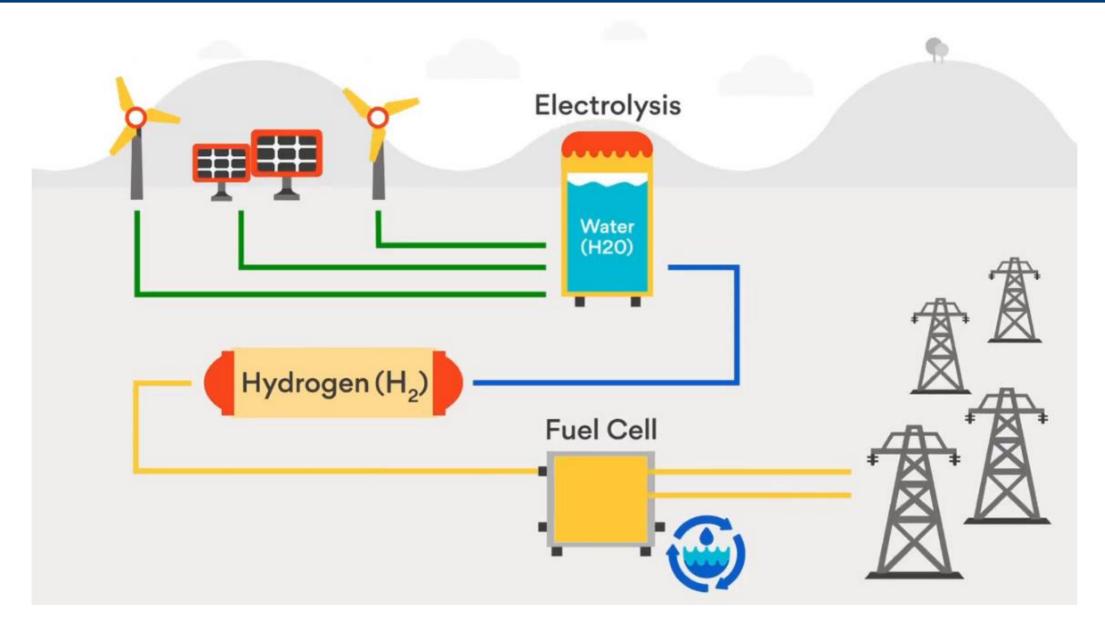
Energy system tomorrow

Our vision is to establish **hydrogen** as the **second central technology basis**. Most of the energy is generated from renewable sources.



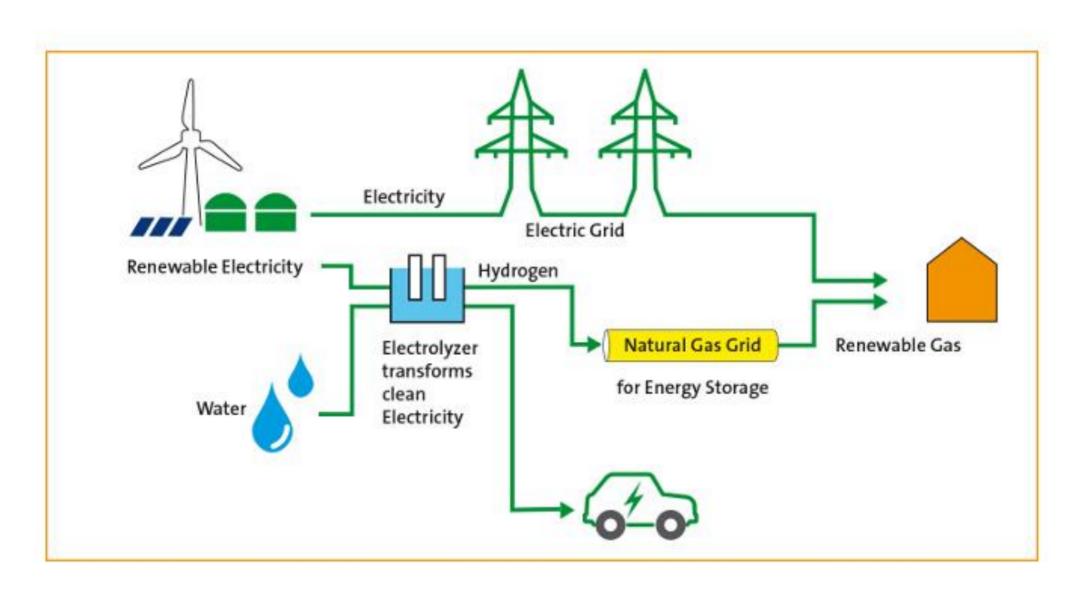
Source: GET H₂ initiative





Key Concept

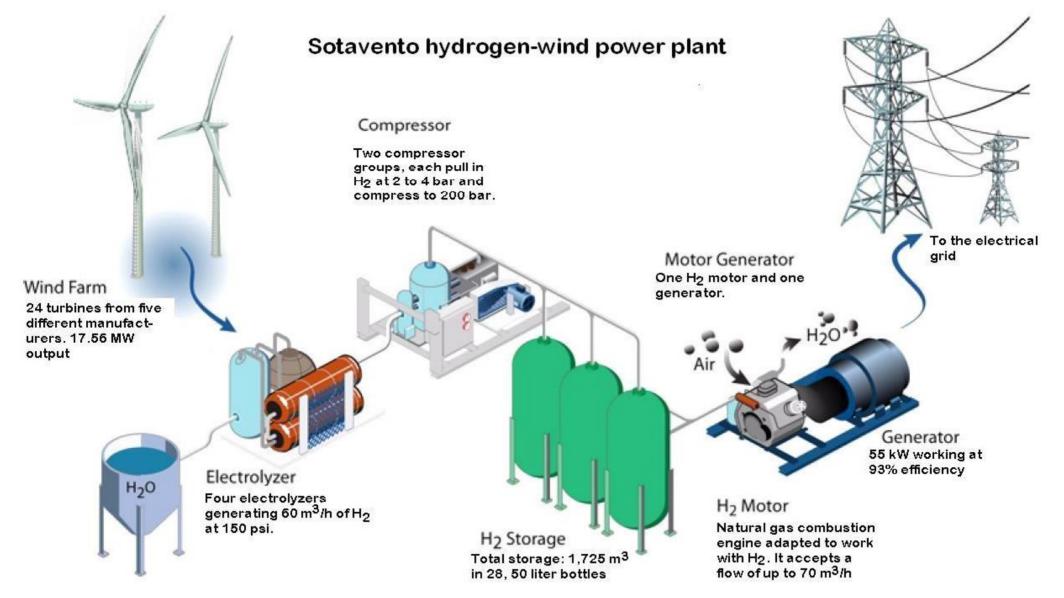
Source: Origin Energy Limited



Key Concept







Key Concept

Air separation unit



Hydrogen refueling station Electrolyzer Hydrogen Dissociation Hydrogen compression Power Distribution Nitrogen Hydrogen Ammonia Wind/solar Carbon-free Ammonia hydrogen production

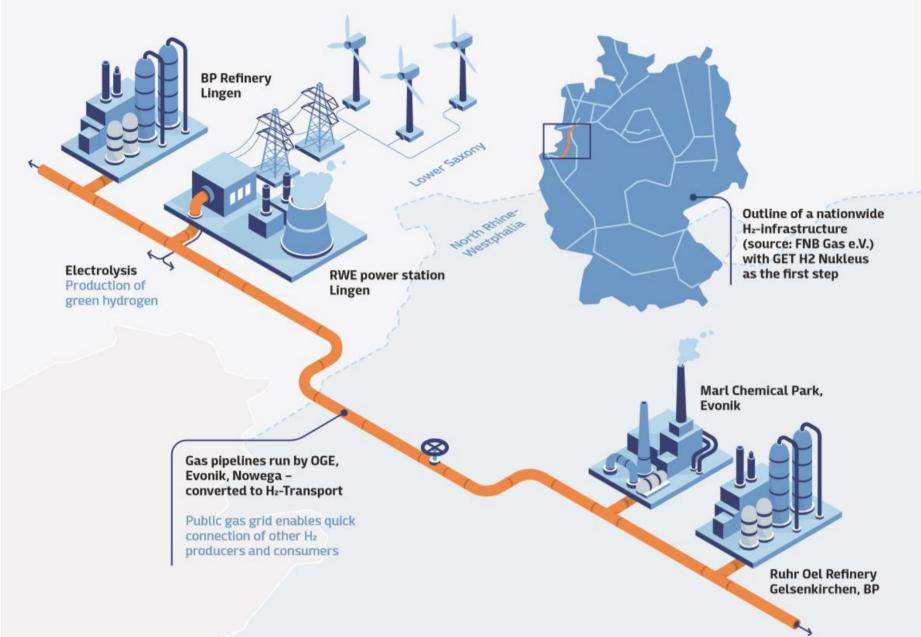
Key Concept

Source: American Chemical Society



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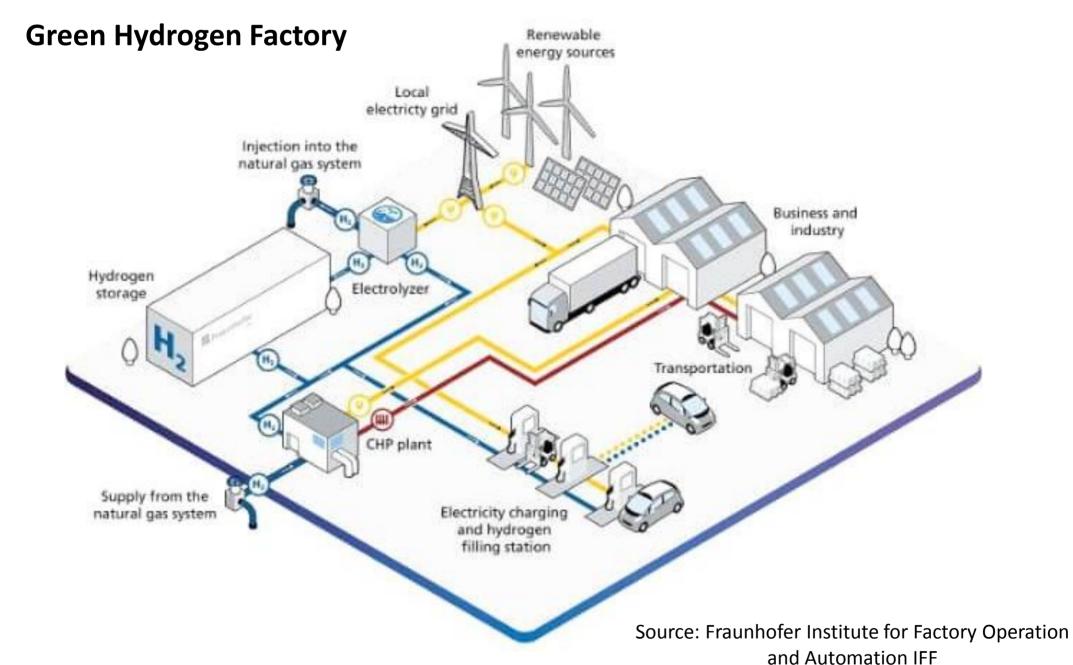




Source: GET H₂ initiative

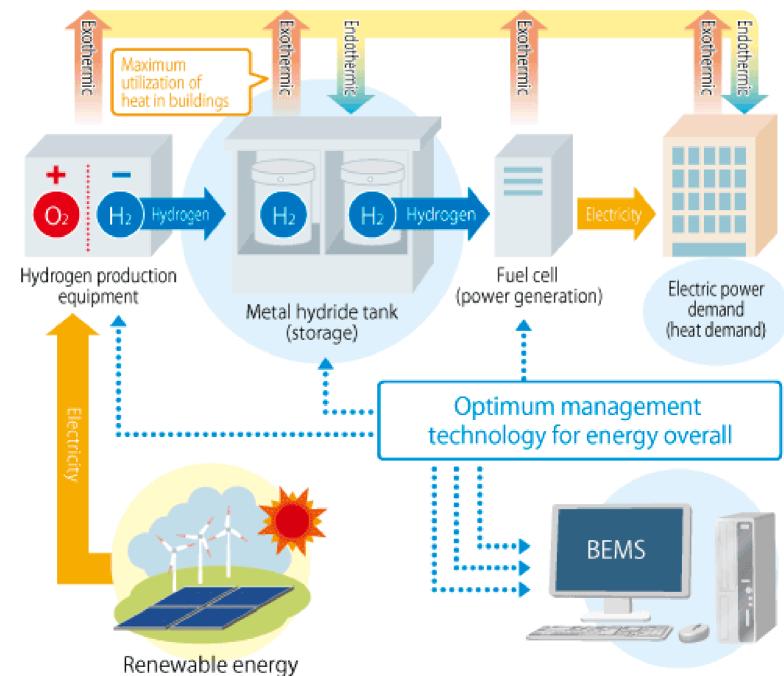
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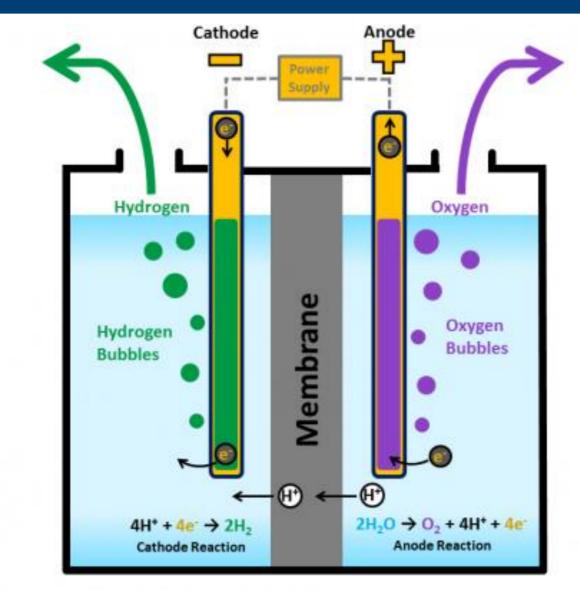






Source: National
Institute of Advanced
Industrial Science
and Technology
(AIST)

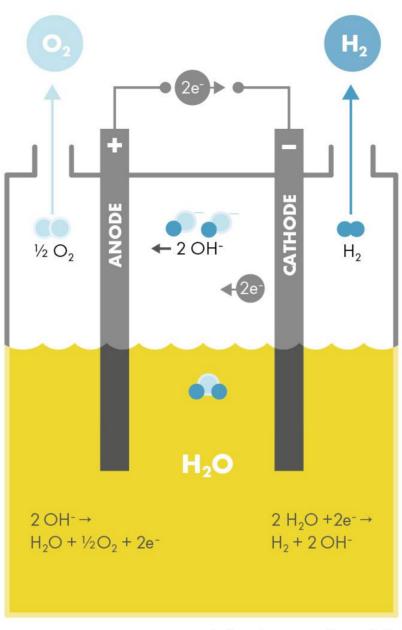
Electrolysis



Proton exchange membrane (PEM) electrolyzer







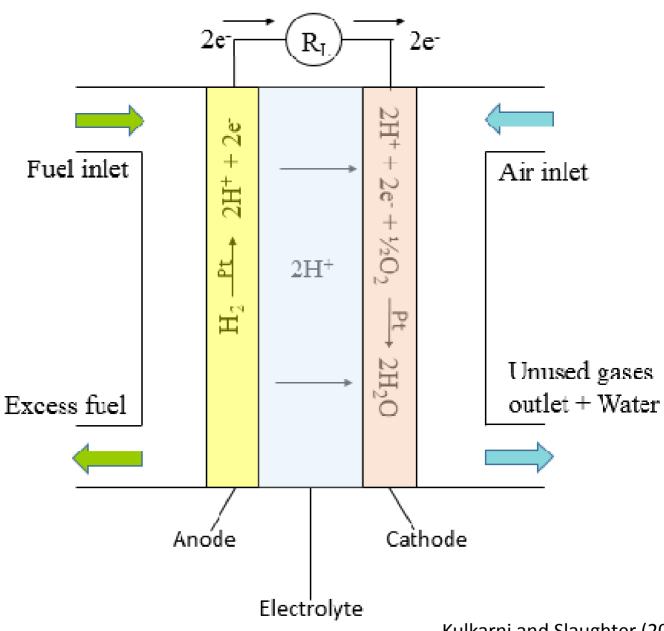
Shell Hydrogen Study © Shell

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Using a fuel cell to power an electric motor is 2 to 3 times more efficient than using a combustion engine.

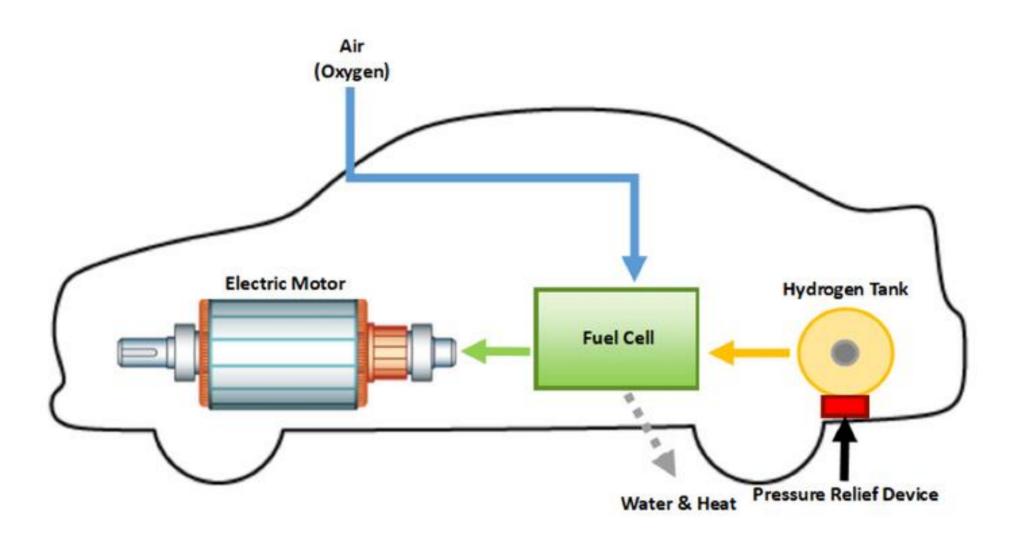
This means that much greater fuel economy is available using hydrogen in a fuel cell.



Hydrogen Fuel Cell

Kulkarni and Slaughter (2015)



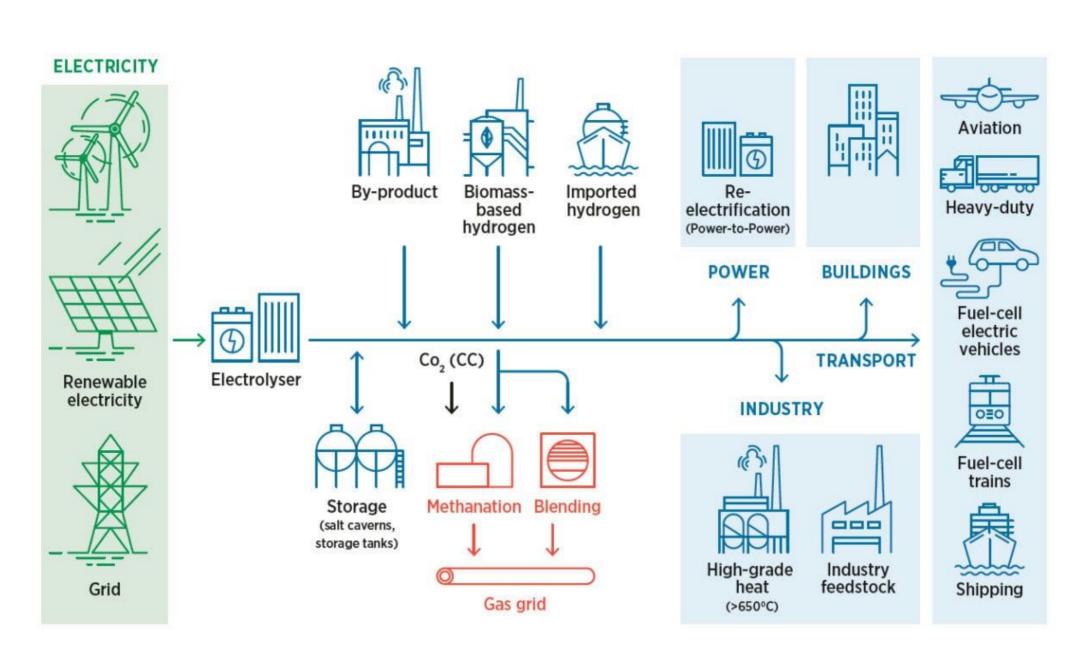


Hydrogen Fuel Cell Vehicles

Source: Firehouse





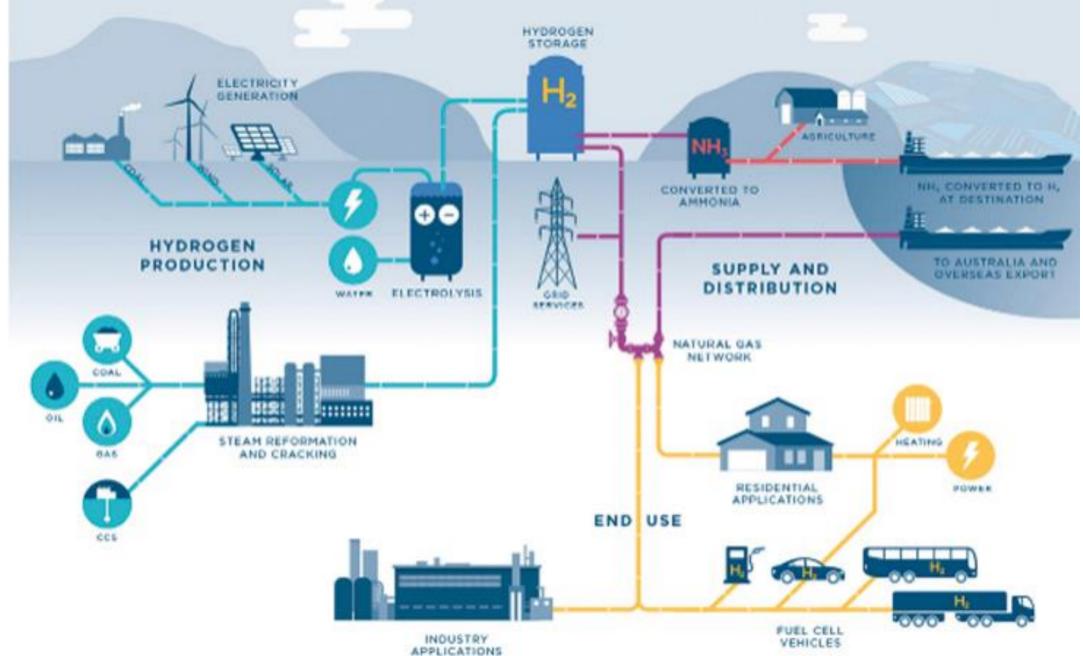


Source: POWER

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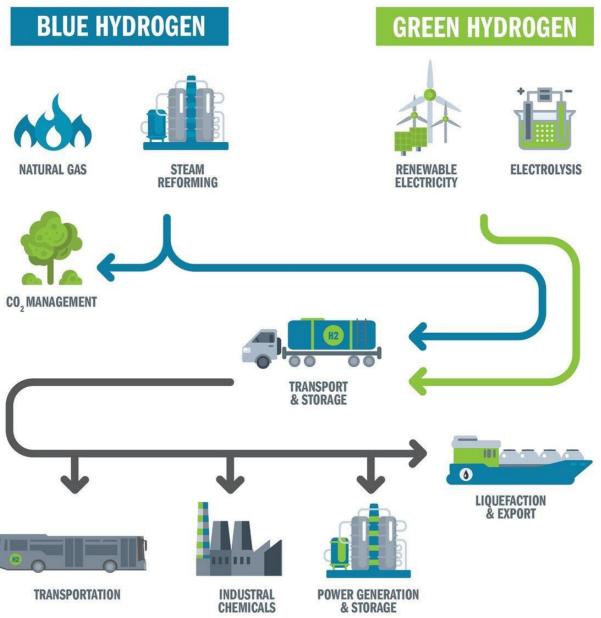
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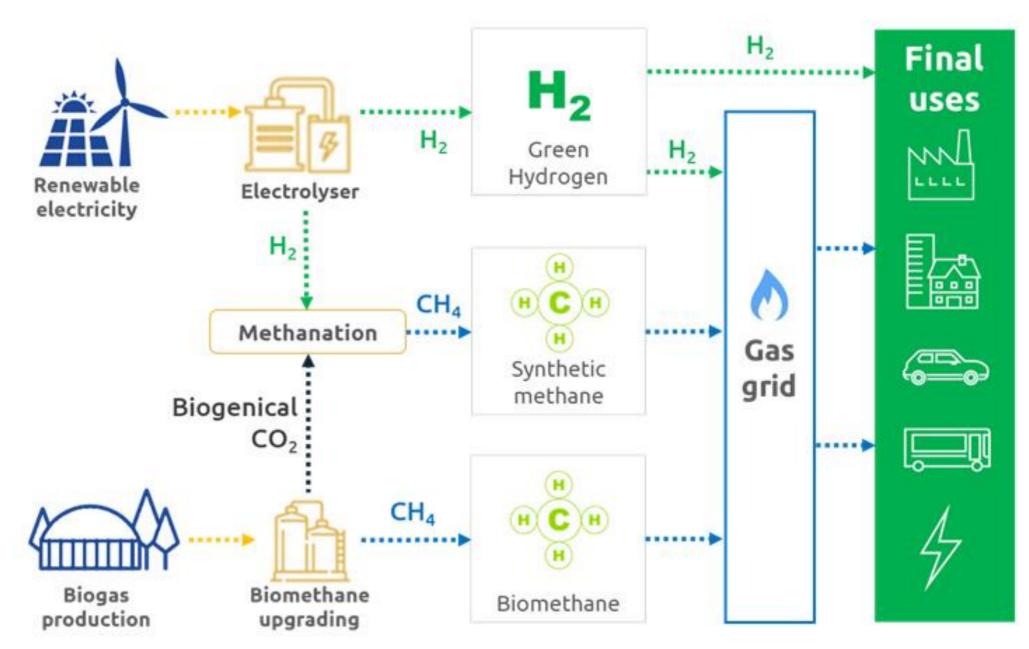


Source: Oman Observer





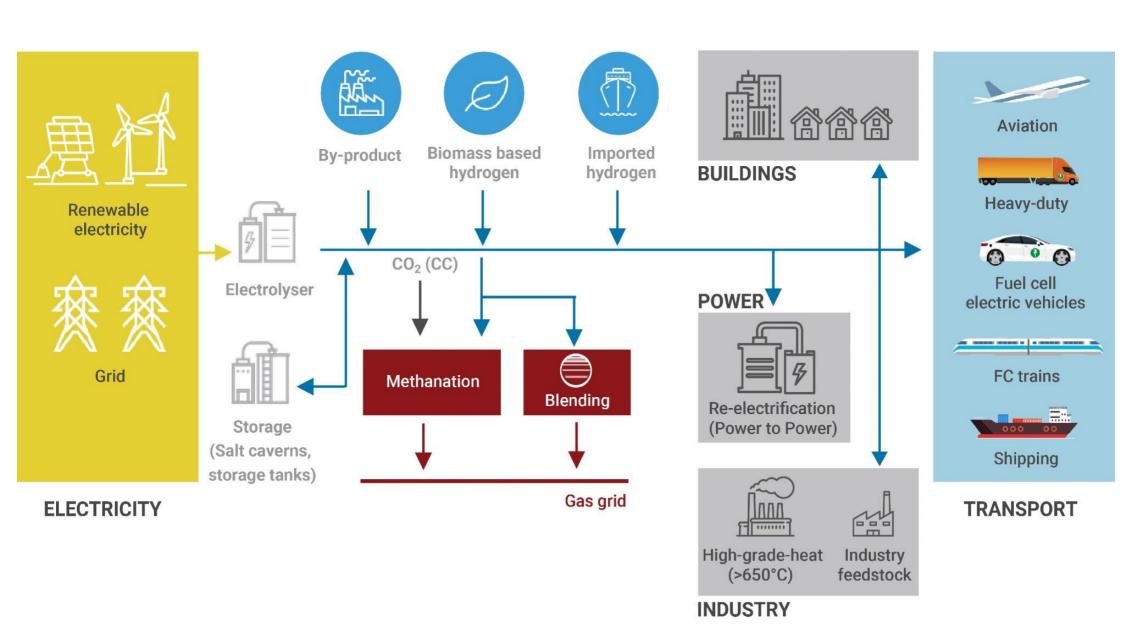
Source: Ballard Power Systems



Source: Snam SPA





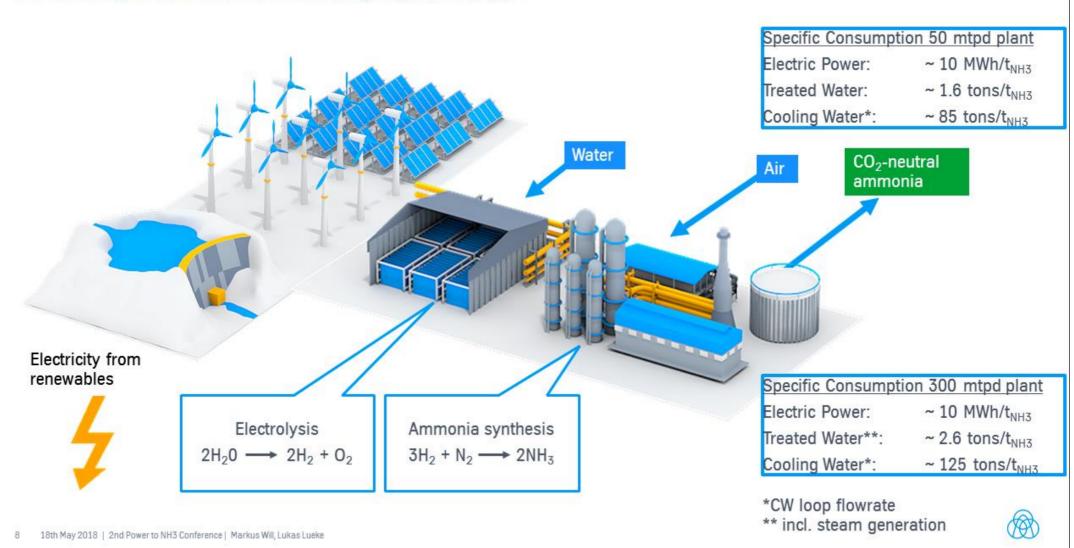


Source: International Renewable Energy Agency (IRENA)

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Introducing renewable ammonia by thyssenkrupp



Green Hydrogen and Renewable Ammonia

Source: Ammonia Energy Association



Example of a Hydrogen Production Facility

The world's largest facility for producing hydrogen fuel is claimed to be the Fukushima Hydrogen Energy Research Field (FH2R), a 10 MW-class hydrogen production unit, inaugurated on 7 March 2020, in Namie, Fukushima Prefecture, Japan.

The site occupies 180,000 m² of land, much of which is occupied by a solar array; but power from the grid is also used to conduct electrolysis of water to produce hydrogen fuel.

Source: Toshiba Energy Systems & Solutions Corporation



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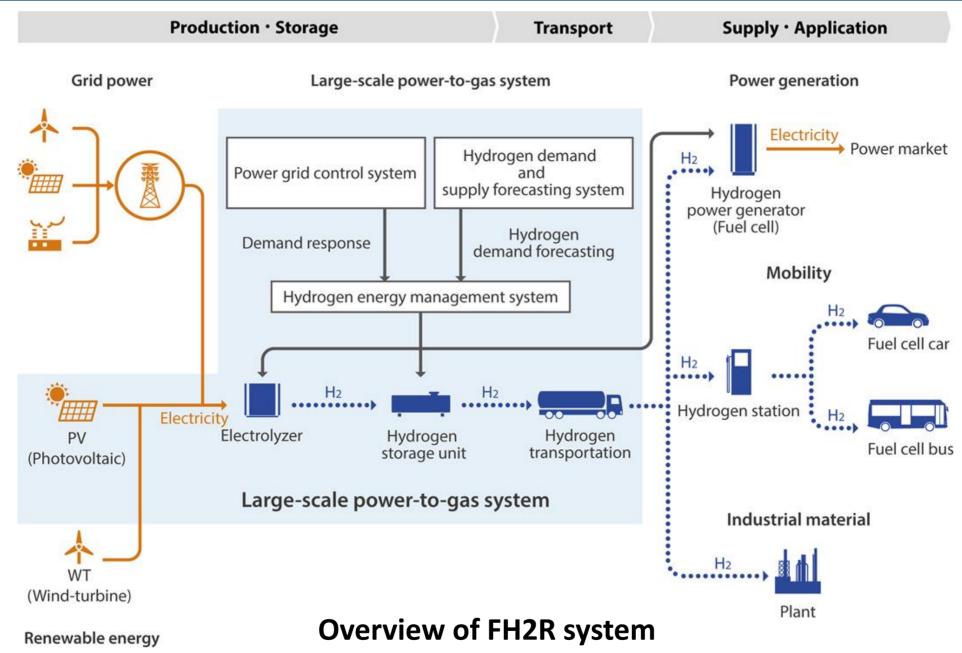


Fukushima Hydrogen Energy Research Field (FH2R)

Source: Toshiba Energy Systems & Solutions Corporation

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Source: Toshiba Energy Systems & Solutions Corporation



Videos

Thank You!

