

# Marine redox evolution and organic accumulation in an intrashelf basin, NE Sichuan Basin during the Late Permian

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## ABSTRACT

The Late Permian oceanic redox conditions are suggested to have had a causal relationship with the latest Permian mass extinction (LPME); however, the tempo-spatial variations of marine redox conditions which could have controlled environmental context of the LPME and organic accumulation remain loosely constrained. Here, we selected an organic-rich Lopingian succession deposited in an intrashelf basin, NE Sichuan Basin, SW China. To document the coeval oceanic changes, we present multiple geochemical proxies, including iron speciation, (major and trace) element contents, mercury (Hg) contents, total organic carbon contents (TOC) and carbonate carbon isotopic composition. Four intervals (I-IV) of discrete redox conditions were identified based on Fe–Mo–U–V datasets, revealing the evolutions from oxic to euxinic, and further to ferruginous conditions. Meanwhile, the primary productivity increased, then decreased in these intervals. The euxinic water wedge appeared in parallel with the high primary productivity and enrichment of organic matter, and advanced upslope and retreated downslope dynamically in the pace of sea-level rise and fall. In this case, marine redox variations were related to interactions of eustatic changes, organic decomposition, paleogeographic setting, upwelling currents, terrigenous input and volcanic activities. Compared with previous studies from other continents, it seems that the Paleo-Tethys was more anoxic and stagnant than Neo-Tethys and Panthalassa during the Late Permian, due to paleogeographic contexts and the more restricted oceanic circulation. In general, the prominent marine anoxia prevailed in the moderate water depth, and predated the major pulse of the LPME in several sections, which implies that anoxia may have not acted as the direct “killing mechanism” of this event.

## 1. Introduction

The Late Permian was a turning interval marked by dramatic oceanic, climatic, and geochemical changes in Earth's history, ending up with a mass extinction event accounting for a loss of more than 90% of marine organisms and over 70% of terrestrial vertebrates (Erwin et al., 2002; Sepkoski, 2016). These changes include large-scale perturbation of carbon cycling (Bernier, 2002; Bond and Grasby, 2017), rapid global warming (Huey and Ward, 2005; Joachimski et al., 2012; Benton and Newell, 2014; Chen et al., 2020), ocean acidification and toxicity (Payne

et al., 2010; Hinojosa et al., 2012; Black et al., 2014; Clarkson et al., 2015), and large-scale oceanic anoxia (Isozaki, 1997; Grice et al., 2005; Zhang et al., 2018a), accompanied with intense volcanic activities of the Large Igneous Province in Siberia (LIPS) (Svensen et al., 2009; Burgess et al., 2017) and/or South China (Gao et al., 2013; Zhao et al., 2019).

Over the past decades, numerous studies claimed that the widespread marine anoxia was the dominant killing mechanism for the latest Permian mass extinction (LPME) (Nielsen and Shen, 2004; Grasby and Beauchamp, 2009; Takahashi et al., 2014). Nonetheless, the anoxia was tempo-spatially heterogeneous across different depositional

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environments (or facies) in different continents (Dolenc et al., 2001; Fio et al., 2010; Clarkson et al., 2016; Elrick et al., 2017; Lei et al., 2017; Xiang et al., 2021), so that it was unlikely a direct cause of this event. Most of previous studies have focused mainly on the environmental changes across the Permian-Triassic (P-T) interval; however, less attention has been paid to the marine redox conditions prior to the P-T interval. This impedes the complete understanding of the oceanic context and the co-evolution of environmental-life processes and organic accumulation prior to the severest P-T biotic crisis.

Besides, the deep-water succession of Upper Permian in NE Sichuan Basin is organic-rich, acting as the high-quality source rock providing

abundant hydrocarbons to the gasfields nearby (Cao et al., 2015; Wei et al., 2018). This facies-specific depositional succession archived abundant information associated with oceanic chemical and physical changes of the Late Permian which would be much helpful for understanding the oceanic factors controlling on the anomalous organic accumulation and evolving oceanic context predating the LPME.

To address the oceanic changes over the Late Permian, this study selected an organic-rich intrashelf basinal succession of the Upper Permian (upper Wuchiaping Formation, Dalong Formation and lowermost Feixianguan Formation in ascending order) at Xibeixiang, Guanyuan, NE Sichuan. Here, a series of analyses were carried out, including

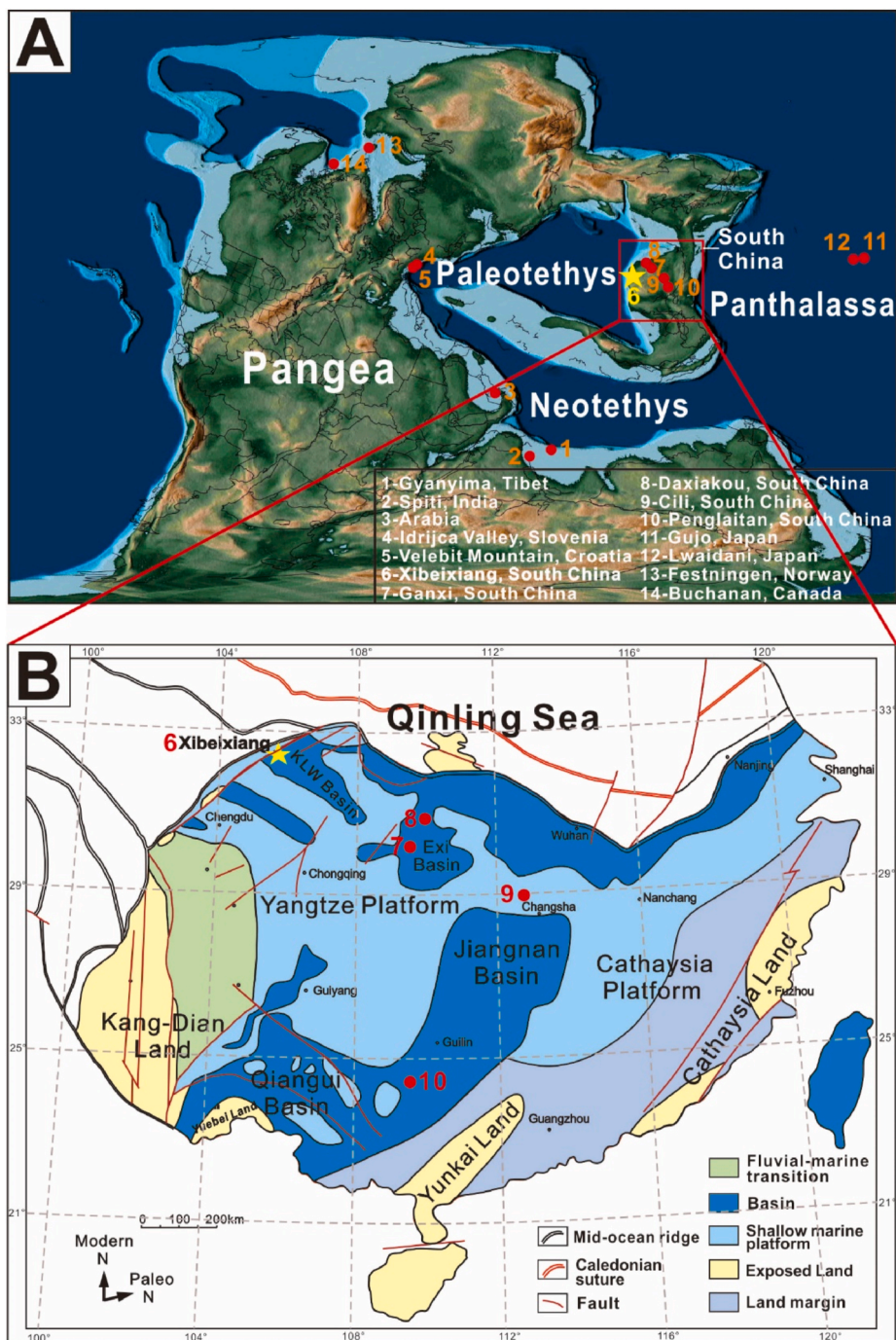


Fig. 1. (A) Paleogeographic map for the latest Permian showing the location of worldwide P-T sections and the present study section (modified from Scotese, 2016), (B) Paleogeographic map of the latest Permian of South China (KLW= Kaijiang-Liangping-Wangcang), showing the location of the Xibeixiang section (Star; modified from Luo et al., 2011).

iron speciation, major and trace element, Hg, total organic carbon contents (TOC), and carbon isotope composition. Based on these data, an integrated sedimentological, and geochemical profile upon the deep-water succession from the Lopingian to the P-T interval is constructed for the first time in the northwestern Yangtze Block. These datasets provide a more complete picture of the paleo-oceanic variations of the entire Late Permian in South China, facilitating better understanding of

the abnormal organic enrichment and co-evolution of environment-life processes during this critical interval prior to the LPME.

## 2. Geological outline of the Lopingian (South China block)

The South China Block, amalgamated by the Yangtze and Cathaysia subblocks, was isolated from other continental blocks and migrated to

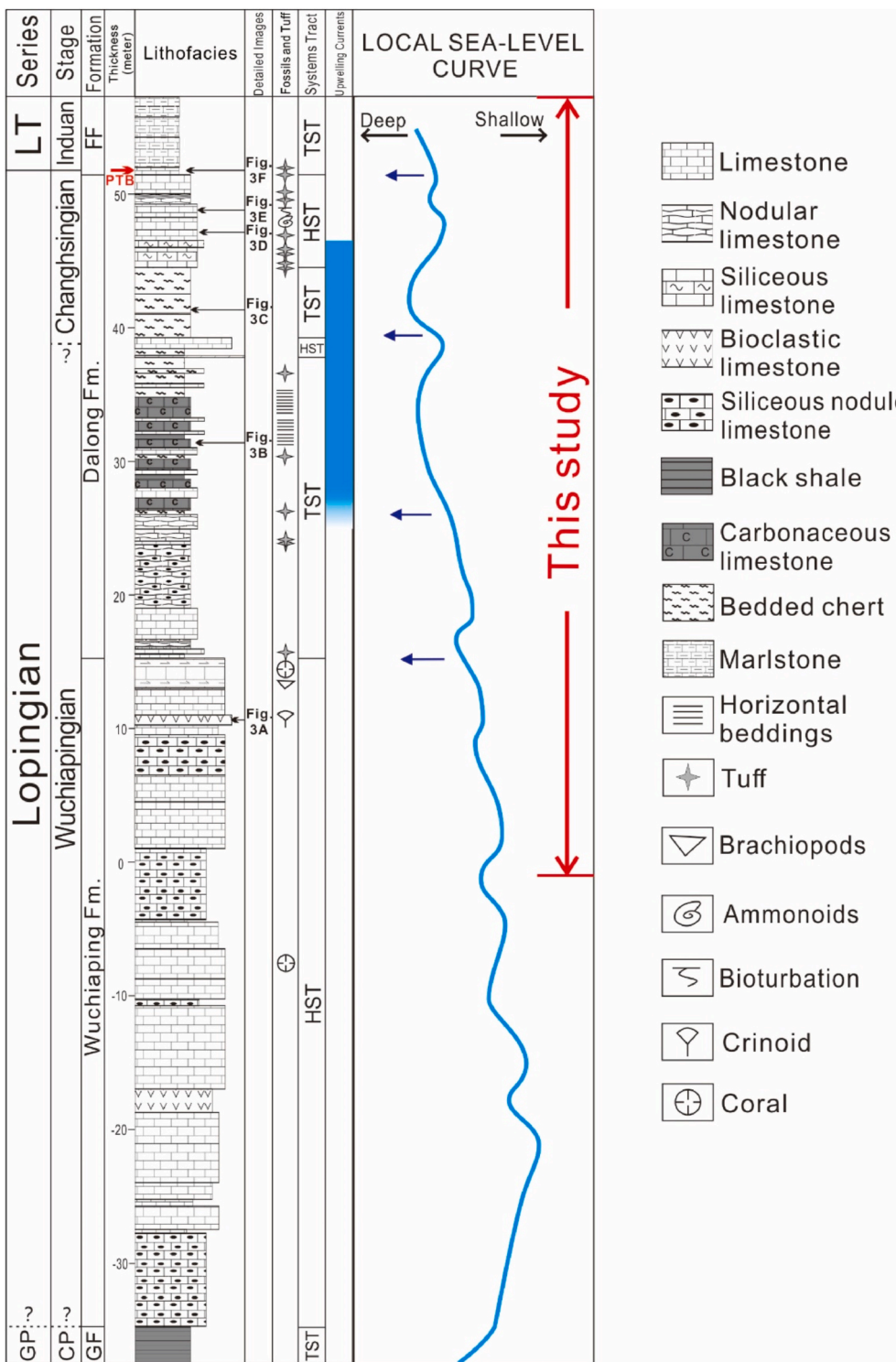
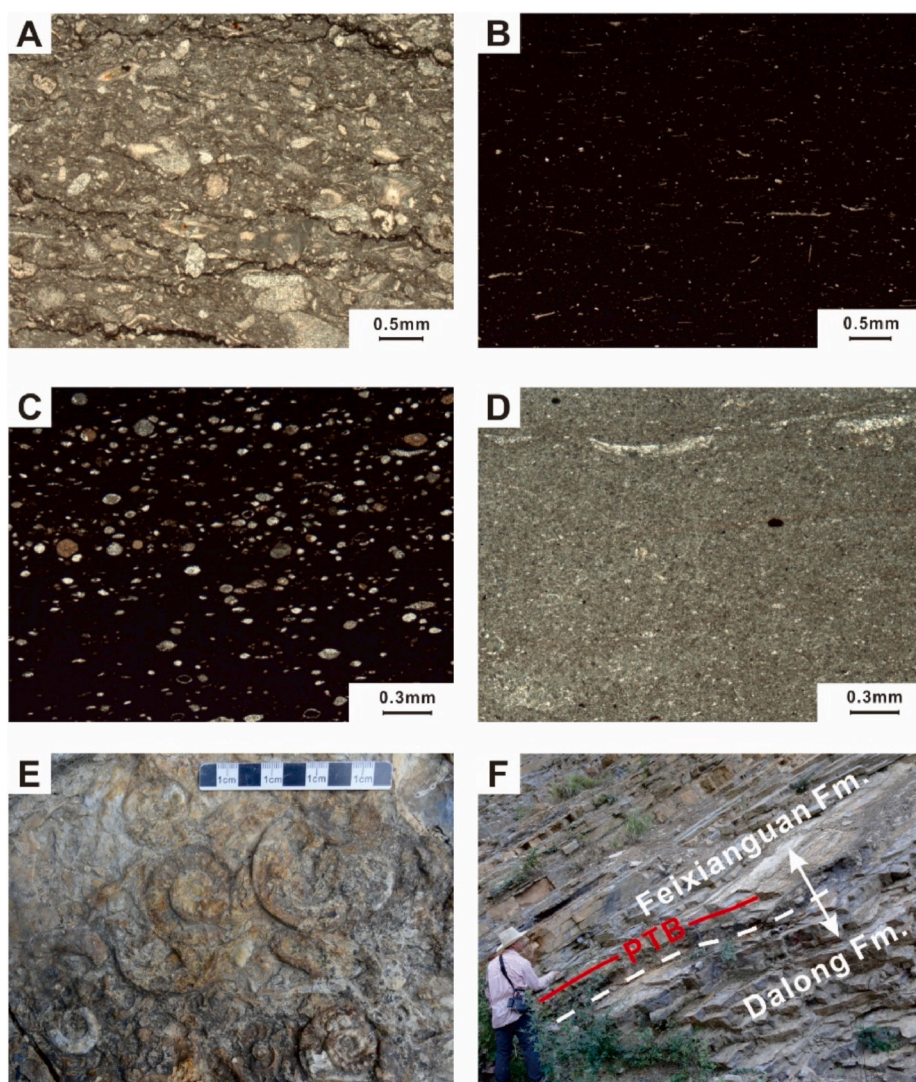


Fig. 2. Lithological column of Xibeixiang Section and sea level changes inferred from the vertical lithofacies changes. Abbreviations: GP-Guadalupian, CP-Capitanian, GF-Gufeng Formation, FF-Feixianguan Formation, LT-Lower Triassic, HST-highstand systems tract, TST-transgressive systems tract. The arrows mark apparent deepening processes (or sea level rise).

the subtropical latitude (approximately 23–25°N) of the Equatorial Warm Water Province (EWWP) during the Lopingian (Late Permian), forming the eastern margin of the Paleo-Tethys Ocean (Scotese and Langford, 1995; Mei and Henderson, 2001, Fig. 1A). During the Late Permian, carbonate platforms predominated over the Yangtze subblock, notably the upper Yangtze region, on which several intraplatform (or intrashelf) basins were developed (Fig. 1B). Paleogeographically, the Xibeixiang section was located near the mouth of the half-opened NW–SE trending Kaijiang-Liangping-Wancang intraplatform basin linked to the Qinling Ocean (or Paleo-Tethys), which was proposed to be caused by the Emei Taphrogenesis (Shellnutt et al., 2012). It is about 35 km far from the Shangsi section northeastward sharing the similar depositional setting, which was once proposed as an auxiliary candidate of GSSP for the P-T boundary (Wignall et al., 1995).

At Xibeixiang, the well-exposed Upper Permian-Lower Triassic succession, overlying the Guadalupian Series, is composed of the Wuchiaping, Dalong and the lower Feixianguan formations in ascending order (Fig. 2). This study focuses on the succession starting from the upper Wuchiaping Formation, which consists of thick-bedded to massive bioclastic wackestones (Fig. 3A) with abundant shallow-water fauna (e. g., echinoid and brachiopod) and cherty (chalcedony) concretions/bands, representing a relative shallow-water environment (late highstand systems tract-late HST). The Dalong Formation characterizes apparent lithological changes. The dominance of deep-water fauna

assemblages (ammonoids and radiolarians) in the Dalong Formation (Fig. 2) is concordant with a deep-water intrashelf basinal environment (Cai et al., 2007; Yan et al., 2008). This formation approximately comprises three parts: the lower, middle and upper ones. The lower part is composed of thin- to medium-bedded lime mudstone/wackestone and thin-bedded siliceous mudstone, indicating a rapid deepening of water depth (onset of transgressive systems tract-TST). The overlying middle part is mainly composed of organic-rich marly limestone, intercalated with black shales/organic-rich limestones (Fig. 3B), indicating a progressive transgression (late TST). The uppermost limestone horizon thus represents a slight sea-level fall of HST. The upper part of the Dalong Formation, approximately corresponding to the Changhsingian Stage, is characterized by bedded cherts intercalated with shale partings in the base (Fig. 3C). It represents the onset of TST of late Lopingian. Upward, it is overlain by the limestone succession composed of thin- to medium-bedded lime mudstone (Fig. 3D) bearing nektonic ammonite fossils (Fig. 3E) and bioturbation, indicating a relative sea level fall (HST). The overlying Feixianguan Formation also marks an apparent lithological transition and is characterized by thin-bedded to platy marlstones or argillaceous lime mudstones intercalated with shaley partings, notably in the basal part (Fig. 3F), reflecting recurring sea-level rise (TST). It is worth to mention that up to 15 layers of bentonites are present, notably around the Permian-Triassic Boundary in Xibeixiang section (Figs. 2 and 3F), showing the direct evidence of volcanic



**Fig. 3.** Lithological features of Upper Permian at Xibeixiang. Photomicrographs (A–D) were taken under plane polarized light (A) Bioclastic wackestone bearing brachiopod and echinoid/criinoid fragments in the upper Wuchiaping Formation, (B) Carbonaceous limestone in the middle Dalong Formation, (C) Bedded chert bearing radiolarians in the upper Dalong Formation and (D) Lime mudstone in the uppermost Dalong Formation. (E) Ammonite fossils in the upper Dalong Formation. (F) Lithological variations across the Permian-Triassic boundary (PTB). Standing person for scale (1.65 m).

activities during deposition.

### 3. Materials and methods

A total of 79 samples were collected from Xibeixiang section, and all potential weathered surfaces, post-depositional veinlets or cavities, and visible nodules or bands were carefully removed from each sample. The samples were then cut into pieces with a hammer and further pulverized into powders of <200 mesh in size using Rocklabs BTRM apparatus. Homogenized powders were taken for subsequent analyses.

#### 3.1. Major and trace elements analyses

For major elements, the lost-on-ignition (LOI) was calculated by the sample's mass difference between prior and after the combustion that can be used later for correction of the elemental abundances. 50 mg of each sample were weighed and dissolved in alkaline solution, followed by dilution of nitric acid and then determined by inductively coupled plasma optical emission spectrometer (ICP-OES) at China University of Geosciences (Beijing). The analytical accuracy monitored by the standards of AGV-2, GSR-1 and GSR-5 was >3%.

For trace elements, about 100 mg of each pulverized sample were dried at 980 °C for 1 h. Thereafter, aliquots of 40 mg of each sample were first treated with mixture solution of 1.5 ml concentrated HNO<sub>3</sub> and 1.5 ml HF, then heated at 195 °C for 48 h within sealed Teflon bombs. 1 ml HNO<sub>3</sub> was added to each sample that was further evaporated to incipient dryness, and then redissolved by 2 ml HNO<sub>3</sub> and 2 ml Milli-Q water and heated at 165 °C for 24 h to ensure complete digestion. Samples were diluted 4000-fold and conducted using inductively coupled plasma-mass spectrometry (ICP-MS) instrument at China University of Geosciences (Beijing). The analytical precision monitored by AGV-2, BHVO-2, W-2, GSR-1 and GSR-3 is generally >5%, except for (P and K) is <15%, and (Cr, Sc, Cu, Zn, Sr, and Ta) is <10%.

To subtract the terrigenous contribution and denote the authigenic component, trace element excess  $X_{XS}$  is calculated as follows:  $X_{XS} = X_{\text{sample}} - Al_{\text{sample}} \times (X/Al)_{\text{PAAS}}$ , where  $X_{\text{sample}}$  and  $Al_{\text{sample}}$  are the concentration of the element X and Al in samples and  $(X/Al)_{\text{PAAS}}$  is the ratio of element X to Al in PAAS (Post-Archean Australian shale) (Taylor and McLennan, 1985; Tribouillard et al., 2006). Likely due to the loss of authigenic components, there are negative  $X_{XS}$  values in some of the samples in this study. We uniformly set these negative values to 0.01 ppm to allow for convenient log<sub>10</sub> plots of trace element excess versus depth.

#### 3.2. Total organic carbon content

TOC analyses were conducted by the acid-wash method as described in (Wang et al., 2012). Aliquots of 200 mg of each sample were first treated twice with 6 N HCl for 24 h to dissolve any carbonate minerals. Samples were then washed with distilled water several times to remove any acid residue until pH > 5.0 and dried overnight (at 50 °C) before being weighed and powdered. Finally, the TOC was determined by Euro-EA3000 elemental analyzer at the Institute of Geology and Geophysics, Chinese Academy of Sciences (IGGCAS).

#### 3.3. Carbon isotopes

Carbonate carbon isotopes were analyzed by Finnigan MAT-253 gas mass spectrometer at the Institute of Geology and Geophysics, Chinese Academy of Sciences (IGGCAS). 0.3–0.45 mg of sample powder was reacted with phosphoric acid at 72 °C and generated CO<sub>2</sub>. Isotopic data were obtained by measuring CO<sub>2</sub>. The results were reported in standard  $\delta$ -notation relative to the Vienna Pee Dee belemnite (V-PDB). Analytical precision is better than 0.15‰.

#### 3.4. Iron speciation

The mass fractions of carbonate-associated ( $Fe_{\text{carb}}$ ), (oxyhydr)oxide ( $Fe_{\text{ox}}$ ), and magnetite-associated ( $Fe_{\text{mag}}$ ) iron phases were analyzed sequentially following the procedure of Poulton and Canfield (2005b). Only samples with total iron ( $Fe_T$ ) > 0.5% were selected for iron speciation analysis (note: all concentrations are weight percent unless otherwise noted). 100 mg of rock powders were used for  $Fe_{\text{carb}}$  extraction by sodium acetate solution buffered to pH = 4.5 with analytical-grade acetic acid in 50 °C water-bath for 48 h. The remaining part of the samples after the first step were treated with sodium dithionite solution buffered to pH = 4.8 by addition of 0.2 M sodium citrate and trace-metal grade acetic acid in a 50 °C water-bath for 2 h to extract  $Fe_{\text{ox}}$ . The remaining residue was further extracted for  $Fe_{\text{mag}}$  using mixed solution of 0.2 M ammonium oxalate and 0.17 M oxalic acid (pH = 3.2, 50 °C, 6 h). After being diluted with 2% HNO<sub>3</sub>, all the extracted fractions were measured for Fe content using atomic absorption spectroscopy (AAS), yielding an RSD of <5% at the State Key Laboratory of Biogeology and Environmental Geology, China University of Geosciences (Wuhan). On the other hand, pyrite iron ( $Fe_{\text{py}}$ ) was extracted following the chromium reduction method (Canfield et al., 1986). Approximately 1–3 g of decarbonated sample powders were reacted with boiling 1 M CrCl<sub>2</sub> solution (200 ml) under N<sub>2</sub> for 2 h. The liberated hydrogen sulfide (H<sub>2</sub>S) gas was trapped in 20 ml of 10 wt% AgNO<sub>3</sub> solution as Ag<sub>2</sub>S precipitates. After filtration, drying, and homogenization of the precipitates, the mass fraction of  $Fe_{\text{py}}$  was calculated stoichiometrically from Ag<sub>2</sub>S. The analysis was carried out at the Institute of Geology and Geophysics, Chinese Academy of Sciences (IGGCAS). Highly reactive iron ( $Fe_{\text{HR}}$ ) was calculated as the sum of  $Fe_{\text{carb}}$ ,  $Fe_{\text{ox}}$ ,  $Fe_{\text{mag}}$ , and  $Fe_{\text{py}}$ . Total iron ( $Fe_T$ ) is obtained from the content of Fe<sub>2</sub>O<sub>3</sub> in major element data.

#### 3.5. Hg content

Prior to Hg analysis, the sample powders were dried at 50 °C for 2 h. Mercury concentration was measured using a Lumex RA-915+ Hg analyzer equipped with a pyrolysis attachment at the Institute of geochemistry, Chinese Academy of Sciences (IGCAS). Standards were measured after every 10 samples to ensure the reliability of the data using GBW07405 (GSS-5) as the Hg standard, and coefficients of variation for triplicate analyses were <6%.

## 4. Results

The main geochemical data from the Xibeixiang section are presented in the supplementary file (Supporting Information), and illustrated in Figs. 4 and 5.

Carbonate carbon isotope data ( $\delta^{13}C_{\text{carb}}$ ) vary from −6.02‰ to 4.57‰, showing three perturbations in the studied succession (Fig. 4): a negative excursion (nadir around −6.02‰) in middle Dalong Formation and a positive anomaly (maximum at 1.62‰) in upper Dalong Formation, which is followed by a rapid negative shift near the P-T boundary. The TOC contents range from 0.03 to 13.68 wt% (mean 2.37 wt%), which are particularly high in the middle of Dalong Formation.

Total Fe ( $Fe_T$ ) concentrations range from 0.06 to 6.20 wt% (average 1.11 wt%), but it is noteworthy that most samples in the upper Wuchiaping Formation and the lower Dalong Formation yield values lower than 0.5 wt%. Their changes generally follow the patterns of TOC variations, except for data in Feixianguan Formation (Fig. 4). For those samples with  $Fe_T$  > 0.5 wt%,  $Fe_{\text{HR}}/Fe_T$  ratios vary from 0.28 to 0.83 (average 0.62), while  $Fe_{\text{py}}/Fe_{\text{HR}}$  ratios vary from 0.01 to 0.94 (average 0.64).

$Mo_{\text{XS}}$  concentrations range from 0.52 to 804.05 ppm (mean 35.85 ppm), and show a similar variation pattern with  $Fe_{\text{py}}/Fe_{\text{HR}}$  ratios (Fig. 4).  $Mo/TOC$  ratios fluctuate between 1.37 and 101.37 ppm/wt% (mean 20.46 ppm/wt%). Moreover,  $[U_{\text{XS}}]$  vary from 0.01 to 81.54 ppm

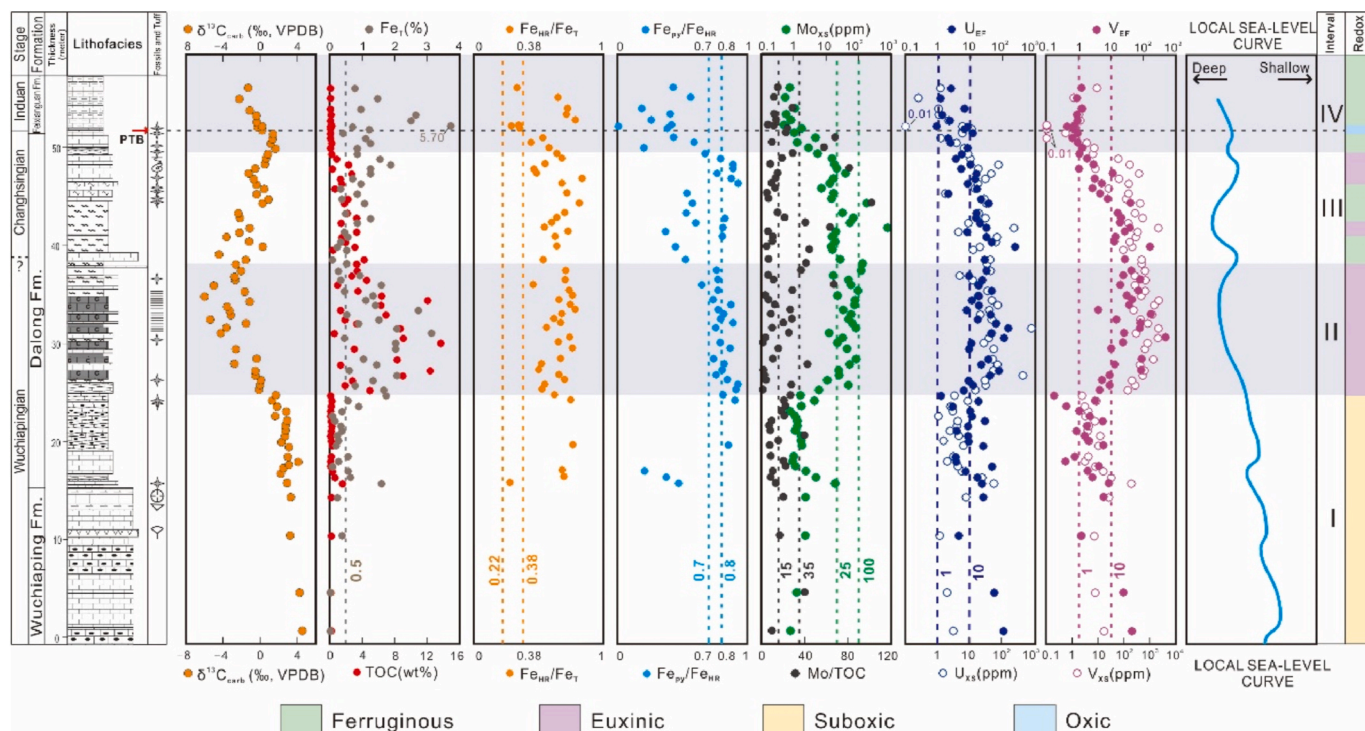


Fig. 4. Stratigraphic variations of  $\delta^{13}C_{carb}$ , TOC, Fe speciation, and other redox proxies in the Xibeixiang section. Vertical dashed lines in each column mark key threshold values of Fe speciation and redox-sensitive trace element proxies documented in the text. See Fig. 2 for lithologies.

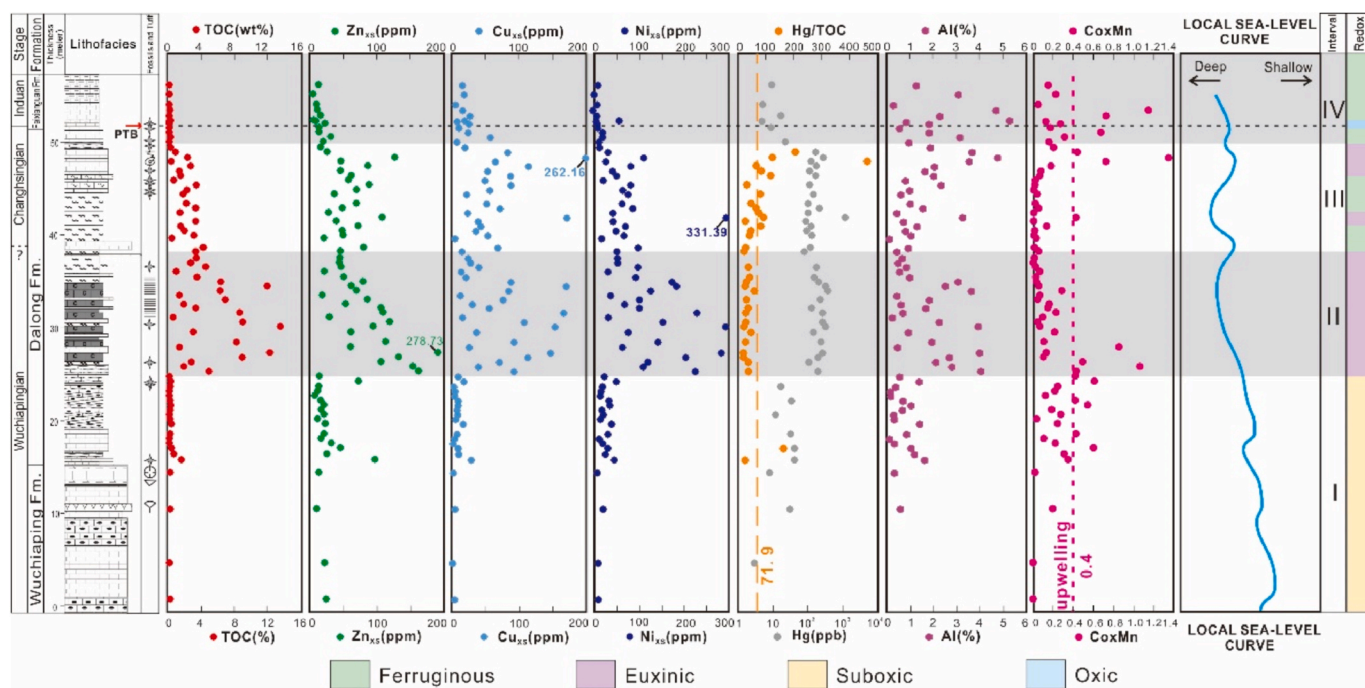


Fig. 5. Vertical variations of proxies of primary productivity (TOC,  $Zn_{XS}$ ,  $Cu_{XS}$ , and  $Ni_{XS}$ ), volcanic activity (Hg abundance and Hg/TOC ratios), terrigenous input (Al contents) and upwelling currents (Co  $\times$  Mn) in the Xibeixiang section. See Fig. 2 for lithologies.

(mean 6.87 ppm),  $U_{EF}$  vary from 0.91 to 347.79 (mean 24.78),  $[V_{XS}]$  vary from 0.01 to 2174.98 ppm (mean 322.39 ppm) and  $V_{EF}$  vary from 0.51 to 200.01 (mean 19.69). All  $U_{XS}$ ,  $V_{XS}$ ,  $U_{EF}$  and  $V_{EF}$  values display similar variation patterns of  $Fe_{Py}/Fe_{HR}$  ratios that stay relatively high in the middle part of Dalong Formation (Fig. 4).

As for nutrient elements (Fig. 5),  $[Zn_{XS}]$ ,  $[Cu_{XS}]$  and  $[Ni_{XS}]$  range from 3.57 to 278.73 ppm, 0.54 to 262.16 ppm and 0.01 to 331.39 ppm,

respectively, following an approximately similar trend maximizing in the middle part of Dalong Formation. Similarly, Hg concentrations vary from 2.92 to 1174.19 ppb (mean 149.34 ppb), reaching relatively high in the mid-upper Dalong Formation. In contrast, Hg/TOC ratios are generally downward-flattening from 21.38 to 463.25 ppb/wt% (mean 101.23 ppb/wt%), although an abrupt increase was created near the P-T boundary. Moreover, Al concentrations vary from 0.04 to 5.62 wt%

(mean 1.62 wt%), maximizing around the P-T boundary. While  $\text{Co} \times \text{Mn}$  values range from 0.0037 to 2.87 ppm-wt% (mean 0.27 ppm-wt%), staying fairly low in the mid-upper Dalong Formation.

## 5. Discussion

### 5.1. Marine redox conditions of Late Permian

According to Tyson and Pearson (1991), paleo-redox states of water media can be differentiated into oxic, suboxic, anoxic, and euxinic conditions. Iron speciation has been widely used to evaluate paleo-oceanic redox conditions under which the fine-grained siliciclastic sediments were deposited. During silicification and dilution of siliciclastic sediments, Fe is a conservative element without significant migration, and remains relatively stable (e.g., Yuan et al., 2014). In this light, iron speciation is applicable for fine-grained siliciclastic rocks, since dilution works consistently onto all iron species.

In modern oxic water columns,  $\text{Fe}_{\text{HR}}$  in sediments makes up less than 38% fraction of  $\text{Fe}_{\text{T}}$ , so that if sediments yield a  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}}$  ratio of  $\geq 0.38$ , an anoxic state is indicated in which as if the  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratio is  $\geq 0.8$ , a euxinic (or sulfidic) condition is further differentiated from the ferruginous-dominant anoxia. Furthermore, an intermittent euxinic state may have been indicated as if the  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios vary between 0.7 and 0.8, notably from deep-time rocks (Poulton and Canfield, 2005a, 2011). These proxies work efficiently only as if  $\text{Fe}_{\text{T}} > 0.5\%$ , even in carbonate rocks (Clarkson et al., 2014).

Several redox-sensitive trace elements (e.g., U, V and Mo) are also commonly used as tracers to reconstruct the paleo-redox history in deep time, because these elements are less soluble under oxygen-depleted marine environments (Calvert and Pedersen, 1993; Algeo and Maynard, 2004; Tribouillard et al., 2006), which allow us to investigate changes in redox conditions in ancient sedimentary archives. Under reducing conditions, U(VI) can be reduced to insoluble U(IV) and precipitate below the Fe(III)–Fe(II) redox boundary (Tribouillard et al., 2006). As for Vanadium, soluble V(V) is reduced to V(IV) under a mildly reducing condition, such as  $\text{VO}(\text{OH})^{3-}$  and insoluble hydroxides  $\text{VO}(\text{OH})_2$ . Under euxinic conditions, the presence of free  $\text{H}_2\text{S}$  can cause V(IV) to be further reduced to V(III), which can be absorbed by the geophyrins complex or precipitated as the solid oxides (Breit and Wanty, 1991; Wanty and Goldhaber, 1992). Besides, molybdate can be converted to particle reactive thiomolybdate in the presence of free  $\text{H}_2\text{S}$  under the anoxic water column, resulting in rapid removal of dissolved Mo from the water into sediments by organic matter or sulfide minerals (Algeo and Lyons, 2006). Studies on modern marine sediments show that Mo concentrations in sediments are commonly higher than 100 ppm under permanent euxinic conditions, and vary between 25 ppm and 100 ppm under intermittently euxinic conditions or permanently Mo-depleted euxinic conditions, and remain lower than 25 ppm under non-euxinic conditions (Scott and Lyons, 2012).

However, it is noteworthy that other factors may have also affected Mo enrichment. Firstly, the so-called “basin reservoir effect” resulting from insufficient Mo renewal in the highly restricted basins could draw down the deep-water aqueous Mo concentrations (Tribouillard et al., 2006). Furthermore, the Mo in water column can be scavenged and trapped by Fe (oxyhydr)oxides and Mn oxides through adsorption in sediments under an oxic-suboxic water column (Scholz et al., 2017; Ho et al., 2018; Scholz, 2018), thus Mo burial fluxes may be higher than expected in weakly anoxic (or sulfidic) environments and lower than anticipated in strongly sulfidic environments. In this light, we thus dominantly rely on the Fe speciation to infer the marine redox conditions in this study, integrated with trace element (U, V and Mo) proxies after careful evaluation of the consistency among different proxies.

Four discrete intervals (I–IV) are identified based on integrated redox proxies (Fig. 4). Interval I (0–23.95 m at height) is located in the lower part of the section, spanning from the upper Wuchiaping Formation to the lower Dalong Formation. In this interval, the relatively depleted  $\text{Fe}_{\text{T}}$

(mostly lower than 0.5 wt%) contents make Fe-specified redox proxies somewhat uncertain, although a few samples with  $\text{Fe}_{\text{T}} > 0.5$  wt% denote an oxygen-deficient environment ( $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}} > 0.38$ ). However, the U, V and Mo are only weakly enriched in this interval (mean  $[\text{U}_{\text{XS}}]$  2.15 ppm, mean  $\text{U}_{\text{EF}}$  19.63, mean  $[\text{V}_{\text{XS}}]$  17.54 ppm, mean  $\text{V}_{\text{EF}}$  4.46, mean  $[\text{Mo}_{\text{XS}}]$  2.72 ppm). Collectively, these redox signals suggest a suboxic water column during deposition.

Interval II (23.95–37.38 m at height) is posited in the middle Dalong Formation. In this interval, almost all samples contain sufficient  $\text{Fe}_{\text{T}}$  ( $>0.5$  wt%), and yield  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}} > 0.38$  (Fig. 4), indicating an anoxic condition, consistent with the enrichment of U (mean  $[\text{U}_{\text{XS}}]$  12.97 ppm, mean  $\text{U}_{\text{EF}}$  42.27) and V (mean  $[\text{V}_{\text{XS}}]$  761.70 ppm, mean  $\text{V}_{\text{EF}}$  43.89). Most of  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios are greater than 0.7, revealing a euxinic-dominant water condition. Moreover,  $[\text{Mo}_{\text{XS}}]$  are mostly between 25 and 100 ppm (mean 55.47 ppm), which may reflect a euxinic-dominant environment in a semi-restricted basin. Therefore, we infer a euxinic condition during this interval.

Upwards to interval III (37.38–48.54 m at height) in the upper Dalong Formation (Fig. 4), samples have  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}} > 0.38$ , suggesting an anoxic watermass; while  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios are fluctuating around 0.7 (0.37–0.94, mean 0.70), implying intermittent incursions of euxinic watermasses into the ferruginous-dominant water column. This scenario generally agrees with relatively high  $[\text{U}_{\text{XS}}]$  (mean 7.53 ppm),  $\text{U}_{\text{EF}}$  (mean 20.69),  $[\text{V}_{\text{XS}}]$  (mean 287.20 ppm),  $\text{V}_{\text{EF}}$  (mean 16.75) and  $[\text{Mo}_{\text{XS}}]$  (mean 67.68 ppm) although moderate declining related to interval II. It is noted that the euxinic water column recurred at the end of this interval concordant with the high  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios ( $>0.8$ ). Therefore, Interval III likely experienced ferruginous-dominant anoxic condition with intermittent incursions of euxinic water.

Further upwards to Interval IV (48.54–55.24 m, the lowermost Feixianguan Formation), samples yield  $\text{Fe}_{\text{T}} > 0.5$  wt% and most of which have  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}}$  ratios  $>0.38$  (0.28–0.78, mean 0.52) with  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios  $<0.7$  (0.01–0.60, mean 0.36) (Fig. 4), indicating a predominance of ferruginous anoxic condition. This scenario generally reconciles the variations in  $[\text{U}_{\text{XS}}]$ ,  $\text{U}_{\text{EF}}$ ,  $[\text{V}_{\text{XS}}]$ ,  $\text{V}_{\text{EF}}$  and  $[\text{Mo}_{\text{XS}}]$  values (Fig. 4). However, samples just at the P-T boundary yield  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{T}}$  ratios less than 0.38, manifesting a short oxic spell. This inference is well in agreement with the drastic drops in  $\text{U}_{\text{XS}}$ ,  $\text{V}_{\text{XS}}$  and  $\text{Mo}_{\text{XS}}$  concentrations (Fig. 4). Thus, a ferruginous state may have predominated over this interval, interrupted by an oxic spell just across the P-T transition.

Furthermore, the weak correlations among  $\text{Mo}_{\text{XS}}$ ,  $\text{U}_{\text{XS}}$ , and  $\text{Fe}_{\text{py}}/\text{Fe}_{\text{HR}}$  ratios versus TOC contents (Fig. 6) suggest that a weak linkage between the redox condition of water column and organic matter accumulation during the Late Permian at Xibeixiang.

### 5.2. Primary productivity

The output of organic carbon onto the seafloor is function of the primary productivity in the surface water to some extent (Pedersen and Calvert, 1990; Canfield, 1994; Tyson, 2005). Several geochemical proxies such as P, Ba, Zn, Cu and Ni abundances in organic matters are somewhat related to the primary bioproductivity of phytoplankton, thereby being used as the tracers to assess the primary productivity in some specific cases (Tribouillard et al., 2006; Piper and Calvert, 2009).

Phosphorus can be deposited in association with decayed microorganisms (Filippelli and Delaney, 1996; Schenau and Lange, 2001; Schenau et al., 2005). Theoretically, oxygenated conditions tend to promote phosphorus retention by adsorption onto the Fe–Mn (oxyhydro-)oxides within sediments such that  $\text{P}_{\text{XS}}$  contents tend to be proportional to the primary productivity in an oxic environment (Cappellen and Ingall, 1994; Schoepfer et al., 2015). In contrast, P can be recycled from organic matter into the reducing water column, resulting in lower  $\text{P}_{\text{XS}}$  contents (or P depletion) in sediments even with a high bioproductivity (Schenau and Lange, 2001; Algeo and Ingall, 2007). In turn, a permanently sulfidic bottom water could impede the precipitation of Fe-oxyhydroxides there, reducing the potential for their

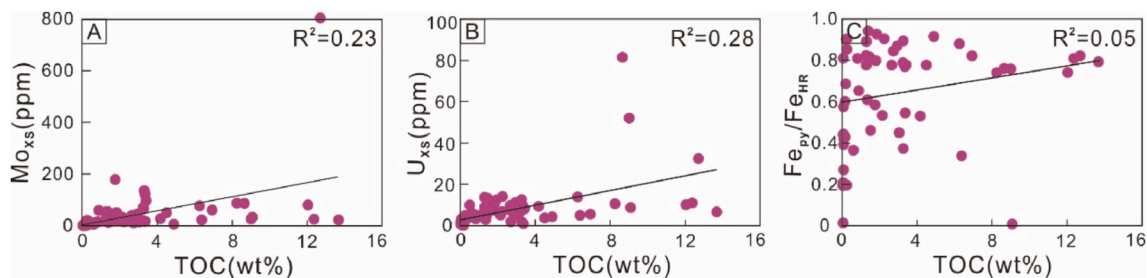


Fig. 6. Crossplots of  $Mo_{XS}$ ,  $U_{XS}$ , and  $Fe_{Py}/Fe_{HR}$  contents versus total organic carbon (TOC) contents in the Upper Permian succession at Xibeixiang section.

adsorption and complexation of the re-mineralized organic P (Tribovillard et al., 2006), limiting direct usage of P abundance as a productivity proxy in organic-rich sediments.

Similarly, although the  $Ba_{XS}$  flux, the indicator of the amount of microcrystal barite rain fallout in the water column, highly depends on the primary productivity (Bishop, 1988; Schmitz, 1987), the redox states of bottom water or porefluid in sediments, notably the anoxic water-mass, would favour the bacterial sulfate reduction (barite dissolution), drawing down the Ba contents in sediments. Therefore, both P and Ba have lower preservation efficiency in anoxic conditions, limiting direct usage of P and Ba abundances as productivity proxies (Dehairs et al., 1980, 1987, 1992; Dymond et al., 1992).

Zinc, copper and nickel are micronutrient-limiting elements and their authigenous fractions are proportional to primary productivity (Algeo and Maynard, 2004; Piper and Perkins, 2004). They are absorbed by micro-organisms through photosynthesis of primary producers before deposition. Subsequently, as the organic matter is partly or completely decomposed, Zn, Cu and Ni could be released to the overlying watermass and potentially to form sulfides in an  $H_2S$ -containing water column, thereby preserving in sediments (Algeo and Maynard, 2004; Piper and Perkins, 2004). As discussed above in Section 5.1, an oxygen-deficient even euxinic condition dominated the intrashelf basin in the Late Permian (Fig. 4). Therefore, only  $Zn_{XS}$ ,  $Cu_{XS}$  and  $Ni_{XS}$  can be used as the proxies for paleo-productivity.

It is well demonstrated that  $[Zn_{XS}]$ ,  $[Cu_{XS}]$ , and  $[Ni_{XS}]$  coherently vary with TOC changes in the Upper Permian succession at Xibeixiang (Figs. 5 and 7), pointing to a deterministic contribution of increased bioproductivity to the organic output and accumulation such that they are efficient indicators of bioproductivity. More specifically, tied to the redox intervals as documented above (Section 5.1), these proxies ( $[Zn_{XS}]$ ,  $[Cu_{XS}]$ , and  $[Ni_{XS}]$ ) stay fairly low in interval I, and culminate in interval II, which are then followed by an episodic decrease in interval III, and remain lowstand in interval IV (Fig. 5). Accordingly, a low productivity rate is indicated in interval I and increasing productivity in interval II, which then is followed by episodic decreases in interval III and remain persistent low in interval IV.

### 5.3. Influence of volcanic activities

Intense volcanic activities may cause major perturbations to

terrestrial and marine ecosystems, and global biogeochemical cycles, due to inputs of both toxic and nutrient substances, and releasing of greenhouse gases ( $CO_2$ ,  $CH_4$  and  $SO_2$ ) into Earth surface system (Grasby et al., 2017). The enhanced volcanic activity was widely suggested as the trigger for inducing the oceanic anoxic events (OAEs) in geological history (Percival et al., 2015; Scaife et al., 2017; Charbonnier et al., 2018). In recent years, mercury (Hg) abundance and its isotope systematics have been increasingly used as a new alternative proxy to trace the volcanic activities in deep geological times (Grasby et al., 2013, 2017; Percival et al., 2015; Sial et al., 2016), providing the linkage of environmental-life coevolution on Earth surface to the Earth's internal process (Sanei et al., 2011; Thibodeau et al., 2016; Percival et al., 2017).

The volcanic flux of Hg is thought to be the largest source of non-anthropogenic Hg input (Pirrone et al., 2010). Hg was washed out from the atmosphere and/or terrestrial riverine runoff and then entered aqueous media (Gill and Fitzgerald, 1988; Holmes et al., 2010; Amos et al., 2014). Subsequently, Hg was precipitated mainly in organic matter, sulfides and clay minerals through forming organic-Hg complexes,  $HgS$ , or adsorption, resulting in a high affinity of Hg to TOC, total sulfur (TS) and Al (Benoit et al., 1999; Ravichandran, 2004; Bower et al., 2008; Bouffard and Amyot, 2009; Selin, 2009; Han et al., 2014; Duan et al., 2016). In order to precisely evaluate Hg anomalies, it is necessary to determine the main occurrence forms and controlling factors of Hg.

Our data from Xibeixiang section show that Hg contents are proportional to TOC contents in most samples, but are poorly correlated with Al and TS (Fig. 8), indicating that Hg resides mainly in the form of organics-Hg complexes in sediments. To buffer the overwhelming influence of TOC contents on the Hg anomalies, the Hg contents are normalized to the TOC contents ( $Hg/TOC$ ) which average 71.9 ppb/wt. % in sediments (Grasby et al., 2019). However, if TOC values are lower than the analytical uncertainty (0.2 wt%), the Hg anomalies should be questioned (Grasby et al., 2013).

In intervals I and IV, most of TOC contents are lower than 0.2 wt%, hence  $Hg/TOC$  ratios are not adopted. Although Hg concentrations remain high in both intervals II and III,  $Hg/TOC$  ratios are flattened, except a few positive anomalies on the top of interval III (Fig. 5), indicating intensified volcanic activities that time. This scenario agrees with concentrated occurrences of tuffaceous (bentolite) layers across the boundary of Dalong and Feixianguan formations (Fig. 3A). In addition, this  $Hg/TOC$  anomaly has been reported from more than 20 localities

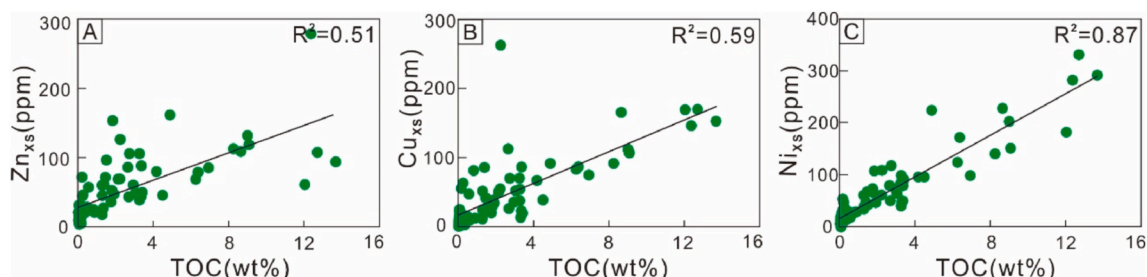


Fig. 7. Crossplots of  $Zn_{XS}$ ,  $Cu_{XS}$ , and  $Ni_{XS}$  contents versus total organic carbon (TOC) contents in the Upper Permian succession at Xibeixiang section.

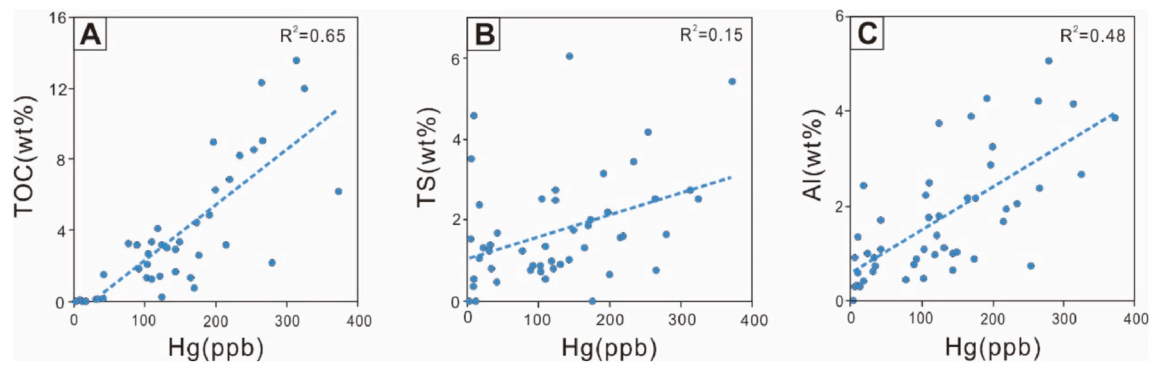


Fig. 8. Cross-plots of (A) Hg vs. TOC, (B) Hg vs. TS, (C) Hg vs. Al in studied interval.

over the world in the time-equivalent horizons of the uppermost Permian, indicating a widespread volcanic activity near the end-Permian (Sanei et al., 2011; Grasby et al., 2013, 2017; Wang et al., 2018a, 2019; Shen et al., 2019a, 2019b). Whether this volcanic signature recorded on South China block was derived from the arc volcanisms around South China block or the Siberia Large Igneous Provinces (SLIPS) still remains controversial (Gao et al., 2013; Burgess et al., 2017; Zhao et al., 2019). Nevertheless, more and more studies suggested that these volcanic materials were probably from the subduction-zone magmatic arcs along the margins of the South China Craton, as evidenced by the crystal fragments, age spectrum of zircons, Cu concentration and isotope anomalies, Hg isotopes and the spatial distribution of particle sizes (Gao et al., 2013; Wang et al., 2018b; Zhao et al., 2019; Zhang et al., 2021).

#### 5.4. Paleogeographic and hydrographic environment

The paleogeographic and hydrographic backgrounds control the topography of sea-floor, terrestrial input, and water circulation in marine systems, which would further act on the primary productivity, marine stratification and (bio)geochemical cycles. In a restricted or semi-restricted basin, the stagnant water may obstruct the replenishment of major and trace elements from the open sea and contribute to redox stratification (Algeo and Rowe, 2012). Therefore, reconstruction of the Lopingian paleogeographic and hydrographic conditions could help constrain oceanic variations on both physical and chemical aspects.

From the late Middle Permian (or Guadalupian), a couple of NW-SE trending intraplateform (or intrashelf) rift subbasins, notably the Kaijiang-Liangping-Wancang (KLW) subbasins, were initiated and generated in the northwestern margin of the Upper Yangtze Platform (Fig. 1B). Due to the temporal synchronism with the Emeishan LIP or mantle plume activity (Shellnutt et al., 2012), they were considered as having been induced by the lateral extensional process of the upper crust surrounding the Emeishan LIP in the course of mantle plume upwelling (He et al., 2003). This half-opened appendix-shaped basin configuration could have restricted the water circulation with the open ocean (Qinling or Paleo-Tethys ocean) to some extent, notably the inner part that was distal away from the outlet (Fig. 1B). The studied Xibeixiang section was located to the mouth of the KLW subbasin, where the depositional environment may be less restricted relative to the inner part.

Mo/TOC ratio is commonly used as a proxy to assess the deep-water paleo-hydrographic conditions of specific basins/shelves (Algeo and Lyons, 2006). Previous data from one upwelling zone (i.e., the Namibian Shelf) and four anoxic silled basins (i.e., the Black Sea, Framvaren Fjord, Cariaco Basin, and Saanich Inlet) revealed that Mo/TOC ratios are higher than  $35 \times 10^{-4}$  ppm/wt% in weakly restricted basins, and vary between  $15 \times 10^{-4}$  to  $35 \times 10^{-4}$  ppm/wt% in moderately restricted basins, and lower than  $15 \times 10^{-4}$  ppm/wt% in strongly restricted basins (Algeo and Lyons, 2006). Mo/TOC values from Xibeixiang Upper Permian section range from 1.37 to 101.37 ppm/wt% (mean 20.46 ppm/wt%) (Fig. 4), showing a moderately restricted environment,

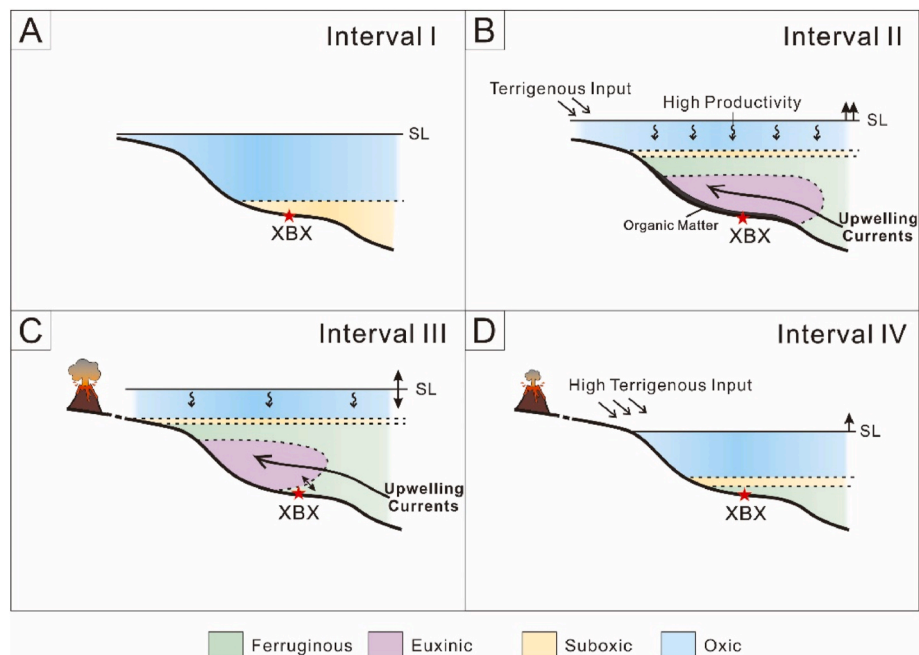
generally agreeing with the paleogeographic setting (Fig. 1B). Liao et al. (2019) also reported a similar Mo/TOC variation pattern in the Upper Permian in the Lower Yangtze region, pointing to a similar hydrographic setting.

Moreover, Co (ppm)  $\times$  Mn (wt%) have also been used recently as the proxy for upwelling currents in modern and ancient coastal systems (Sweere et al., 2016; Zhang et al., 2018b), where the sediments deposited generally yield a low Co  $\times$  Mn value ( $<0.4$  ppm-wt%) (e.g., Peruvian Margin; Böning et al., 2004) probably in response to increasing Mn–Co removing of the upwelling Mn–Co conveyor belt (Sweere et al., 2016). Co  $\times$  Mn values from Xibeixiang section are relatively high in the intervals I and IV (mean 0.27 and 0.58 ppm-wt%, respectively) (Fig. 5), indicating weak upwelling (or restricted) environments. In contrast, they turn to drop in most of intervals II and III (mean 0.20 and 0.19 ppm-wt%, respectively), likely showing enhanced upwelling currents during the two intervals. This inference are consistent with the transgression (Fig. 2) and the deposition of radiolarian-bearing siliceous rocks (Fig. 3C) in this interval.

#### 5.5. Controls on the redox variations during the Late Permian

As documented above (Section 5.1), four intervals (or phases) are discriminated in terms of redox changes through the Late Permian to the earliest Triassic (Fig. 4). Interval I was characterized by a suboxic condition during deposition. The relative lowstand of sea level, as indicated by thick-bedded shallow water limestones bearing abundant benthic faunas (e.g., coral, brachiopod and crinoids) (Figs. 2 and 3), enabled the carbonate platform (ramp) to stand within the euphotic zone of sea surface water in which the excessive  $O_2$  concentration and carbonate production of carbonate factory could have inhibited the organic accumulation/preservation due to the fast rate of organic decomposition and sediment dilution (Fig. 9A). On the other hand, the oligotrophic clear surface water above the carbonate platform, as reflected by the benthic oligotrophic-dominant faunal assemblages (Figs. 2 and 3) and extremely low nutrient-limiting elements (i.e., Fe, Zn, Ni, Cu) (Figs. 4 and 5), could have substantially limited the primary productivity. All these factors together may have maintained the oxygenated state of the shallow surface water (Fig. 9A).

Interval II was characterized by a euxinic state of bottom water at Xibeixiang, as indicated by the Fe speciation and RSE abundances (Fig. 4). During this interval, the relative sea level rose rapidly as evidenced by the occurrence of organic-rich marly (or argillaceous) limestones bearing pelagic fauna (Figs. 2 and 3), so that the carbonate factory was pushed down below the optimal production depth and subsequently drown in which the carbonate production was plummeted and suffocated, leading to few deposition of carbonate sediments. Meanwhile, rapid increases in the nutrient-limiting elements (i.e., Fe, Zn, Ni, Cu) into the basin (Figs. 4 and 5), probably induced by enhanced oceanic upwelling of nutrient-rich deep watermasses in the context of transgression as evidenced by the simultaneous negative shift of  $\delta^{13}C_{carb}$



**Fig. 9.** Schematic diagrams showing the redox variations and controlling factors on the redox conditions in Kaijiang-Liangping-Wangcang basin during deposition of Intervals I (A), Interval II (B), Interval III (C) and Interval IV (D). This schematic illustration is not scaled. See text for detailed explanation. Abbreviations: XBX-Xibeixiang section.

(Fig. 4) and low  $\text{Co} \times \text{Mn}$  values (Fig. 5), and moderate terrigenous input indicated by high Al contents (Fig. 5), could have induced the phytoplankton bloom, increasing the primary productivity and organic fluxes (Piper and Perkins, 2004). This scenario, in turn, would have increased the oxygen consumption through organic decaying, and aggravated the oxygen depletion (anoxicity) in the water column, reconciling the coeval RSE enrichment in stratigraphic column (Fig. 4). Under this circumstance, the increasing organic output/accumulation and sufficient sulfate concentration of the sea further enhanced bacterial sulfate reduction (BSR) and excessive  $\text{H}_2\text{S}$  releasing into the overlying watermass (Habicht et al., 2002; Rimmer et al., 2004). As a result, sulfidic (euxinic) watermass was created coincident with the high production zone (Fig. 9B) as seen in the stratigraphic records. This condition, on the other hand, would further favour organic preservation in sediments, at least partly contributing to the high TOC contents recorded in the stratigraphic column (Fig. 4). Similar patterns of coupled high production and anoxic (even euxinic) zones were widely reported on modern upwelling shelves and deep-time stratigraphic records (Rullkötter, 2006; Huang et al., 2019; Zhang et al., 2020b).

Interval III was dominated by ferruginous anoxic condition with intermittent incursions of euxinic water. During this interval, the relative sea level remained still high with fluctuations during the early stage and tended to drop later on as indicated by the transition from bedded cherts to thin-to medium-bedded limestones bearing ammonites (Figs. 2, 3 and 9C). On the other hand, apparent decreases in nutrient-limiting element fluxes (Fe, Zn, Ni, and Cu) to the waters (Figs. 4 and 5) could have reduced the primary productivity significantly likely driven by the slowdown of terrigenous input. Meanwhile, the decreasing organic fallouts onto the seafloor would in turn slow down the oxygen consumption and subsequent BSR, lessening the anoxicity of the water column, thus accounting for the dominance of ferruginous anoxia with intermittent euxinic spells, notably in the early stage (with a highstand sea level). In contrast, the recurrence of euxinic water column at the end of this interval might result from the downdip retreat of euxinic zone (or wedge) driven by the relative sea level drop, or volcanic-induced increases in both nutrient and sulfur fluxes, as indicated by increased occurrence of bentonite layers (Fig. 2) and sharp increases in  $\text{Hg}/\text{TOC}$

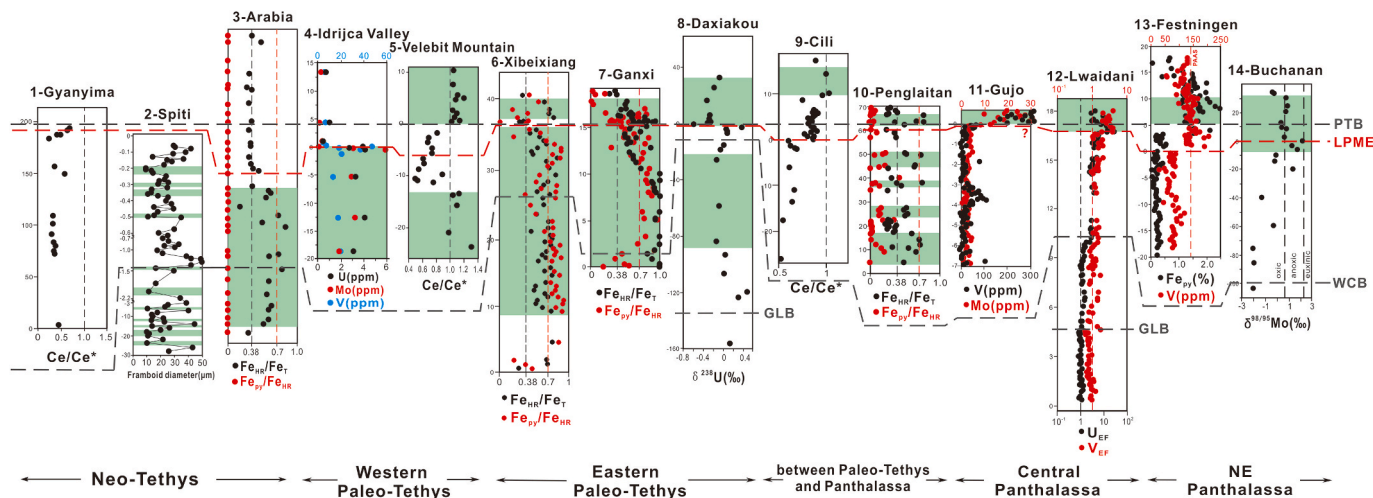
ratio (Fig. 5).

Interval IV was dominated by ferruginous anoxic state with an oxic spell just across the P-T transition, as indicated by the Fe speciation and RSE abundances (Fig. 4). During this interval, the relative sea level rose apparently after the end-interval III drop as evidenced by the occurrence of thin-bedded to platy lime mudstones intercalated with shale layers (Figs. 2 and 9D). On the other hand, the low primary productivity, as indicated by extremely low nutrient-limiting (i.e., Fe, Zn, Ni, Cu) (Figs. 4 and 5) fluxes, was likely driven by the waning of upwelling currents, agreeing with relatively high  $\text{Co} \times \text{Mn}$  values (Fig. 5). These factors together could have induced the decreasing organic matter output, which ultimately reduce the oxygen consumption, and the anoxicity of watermass. Meanwhile, the abrupt high terrigenous fluxes, notably across the P-T boundary, as evidenced by high Al contents in sediments (Fig. 5), could result in the sporadic sediment dilution, thus a short oxic episode during this interval.

#### 5.6. Comparison of redox changes during the Late Permian

In previous studies, the severest oceanic anoxia (i.e., “superanoxia”), as indicated by shallowing of the chemocline and the expansion of sulfidic (euxinic) watermasses to the surface euphotic zone, was hypothesized to be the leading cause driving the destruction of the main habitable niches and the subsequent mass extinction at the end-Permian (Riccardi et al., 2006; Shen et al., 2011). Nevertheless, numerous works demonstrated that anoxic or euxinic conditions had even occurred, dominantly during the Late Permian long before the LPME (Dolenec et al., 2001; Fio et al., 2010; Clarkson et al., 2016; Elrick et al., 2017; Lei et al., 2017; Xiang et al., 2021), only making up the prelude chapter of this severest crisis. To test the spatiotemporal extents/changes of the marine anoxia during the Late Permian, and the links to the LPME, we collect the redox data of Late Permian from different paleogeographic and paleobathymetrical settings around the world. These datasets, however, showed a high spatial heterogeneity and temporal diachroneity of the marine anoxic processes during the Late Permian, and a loose temporal link to the LPME (Fig. 10).

In Neo-Tethys regime,  $\text{Ce}/\text{Ce}^*$  data from the Upper Permian



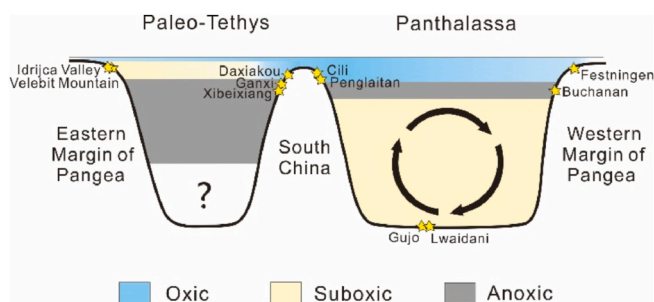
**Fig. 10.** Stratigraphic and paleo-redox conditions of different sections in variable backgrounds of paleogeography and paleobathymetry. Note that the green area represents anoxic conditions, and the locations of section are marked in Fig. 1A. Data sources: Gyanyima, Garbelli et al. (2016); Spiti, Stebbins et al. (2019); Arabia, Clarkson et al. (2016); Idrijca Valley, Dolenc et al. (2001); Velebit Mountain, Fio et al. (2010); Ganxi, Lei et al. (2017); Daxiakou, Elrick et al. (2017); Cili, Loope et al. (2013); Penglaitan, Xiang et al. (2021); Gujo, Algeo et al. (2011); Lwaidani, Onoue et al. (2021); Festningen, Grasby et al. (2015); Buchanan, Proemse et al. (2013). Abbreviations: PTB–Permian-Triassic boundary, LPME–Latest Permian Mass Extinction, WCB–Wuchiapingian–Changhsingian boundary, GLB–Guadalupian-Lopingian boundary. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

succession deposited in a carbonate platform setting on the seamount at Gyanyima, Tibet (Fig. 1 for location), indicate an oxic prevailing marine environment without signal of anoxia during the Late Permian (Garbelli et al., 2016). The Spiti section from India, located in southern Neo-Tethys (Fig. 1), comprises carbonates with black shale intercalations, representing a deeper water sedimentary environment relative to Gyanyima section. Size statistics of framboid pyrite diameter in sediments suggested only sporadic anoxia rather than a persistent anoxic condition from the late Wuchiapingian to the early-middle Changhsingian (Stebbins et al., 2019). Furthermore, iron speciation data from an Arabian section deposited from continental margin to pelagic basin in the western part of Neo-Tethys revealed that anoxia only occurred from the late Wuchiapingian to the early-middle Changhsingian, but did not occur during the LPME (Clarkson et al., 2016); whereas, data from a mid-slope section showed that the anoxic environment did not appear until the extinction interval. In contrast, data from a platform section illustrated an oxic condition during almost the entire Late Permian (Clarkson et al., 2016). Collectively, the oxic environment prevailed in the shallow seas of Neo-Tethys regime and intermittent anoxia only appeared in the mid-depth water column below the mid-slope, whilst euxinic condition did not take place completely.

However, in the western Paleo-Tethys, redox-sensitive element proxies from black carbonates at Idrijca Valley in Slovenia suggested that oceanic anoxia dominated the Late Permian, but was shifted to an oxygenated condition during the P-T interval (Dolenc et al., 2001). In contrast, Ce anomalies in dolomites of the Velebit Mountains in Croatia demonstrated an oxic condition during the Late Permian prior to the P-T transition, and the anoxia only occurred after the extinction event (Fio et al., 2010). In the eastern Paleo-Tethys, iron speciation data from both the Xibeixiang and Ganxi sections of deep-water-dominated facies show an overall anoxic condition from the late Wuchiapingian to the Early Triassic (this study; Lei et al., 2017). While U isotope data from the Daxiakou section (deposited on a carbonate ramp) proposed that the anoxic condition emerged during the mid-late Wuchiapingian and the LPME (Elrick et al., 2017). It is worth noting that a short-term oxic spell occurred across the PTB in Xibeixiang, Shangsi and Daxiakou sections (this study; Xiang et al., 2016; Elrick et al., 2017), which possibly resulted from accelerating increases in terrestrial flux and dilution due to the loss of terrestrial vegetation during the LPME (Zhang et al., 2020a). Collectively, it seems that the oceanic anoxia was more

prevalent in Paleo-Tethys regime than that in the Neo-Tethys during the Late Permian and the degree of anoxia in deep-water settings was more serious than that in shallow marines (Fig. 11). Nonetheless, redox variations in pelagic basins in the middle of Paleo-Tethys remain uncertain in the absence of data.

In the Panthalassa regime (Fig. 10), in the shallow saddle portion between Paleo-Tethys and Panthalassa, Ce anomalies data from the Cili section in South China show that the watermass was oxic until the Early Triassic (Loope et al., 2013). Similarly, iron speciation data from Penglaitan, South China demonstrate sporadic oxygen deficiency during the Late Permian and predominance of oxic-suboxic in absence of euxinic watermass during the LPME (Xiang et al., 2021). Nevertheless, data from sections in Japan in central Panthalassa show different results. No anoxic watermass was generated in the Late Permian until the Early Triassic at Gujo section (Algeo et al., 2011; Fujisaki et al., 2019). In addition, redox-sensitive element (RSE) proxies from the Lwaidani section in Japan show predominance of a suboxic condition on the pelagic seafloor over the most of the Changhsingian, although aggravating near the LPME (Onoue et al., 2021). Moving to the northeastern Panthalassa (Fig. 10), Fe speciation and RSE data from Festningen section in Norway, however, show that marine anoxia only appeared during the Early Triassic (Grasby et al., 2015). With the facies deepening to the distal deep-water slope setting (Buchanan Lake, Canada), a large positive shift in  $\delta^{98/95}\text{Mo}$  values was discovered at the extinction horizon, consistent with the onset of anoxic conditions (Proemse et al., 2013). In contrast, Mo isotopic data from a storm-influenced shelf (West Blind Fjord,



**Fig. 11.** Schematic W-E cross section showing the redox conditions across the Paleo-Tethys and Panthalassa during the Late Permian.

Canada) show little changes in composition, indicating a persistent oxic condition across the LPME interval (Proemse et al., 2013). In summary, as documented above, suboxic conditions predominated over pelagic basin floor in the Panthalassa regime through the Late Permian and the onsets of anoxia were diachronous depending on latitudes (Takahashi et al., 2021), while anoxic or even euxinic conditions could have impinged onto the mid-depth continental slopes, evolving updip to a complete oxic condition on the continental shelf even throughout the LPME interval.

Multiple lines of evidence have revealed apparent redox heterogeneities among the Neo-Tethys, Paleo-Tethys and Panthalassa regimes during the Late Permian (Fig. 11). Compared with the Neo-Tethys and the Panthalassa, the degree of anoxicity in the Paleo-Tethys regime was more serious, which may result from the paleogeographic contexts of the three co-existing ocean regimes during the Late Permian which could have resulted in the differences in oceanic circulation among these oceans (Fig. 1A).

The Paleo-Tethys regime was bounded by Pangea on the west, the Gondwanaland on the south, the Siberian Plate on the north and Asiatic Hunic terranes along the eastern and southern flanks of the ocean, which included the Cimmerian, South China, and North China blocks (Fig. 1A). This loop-like continent/terrane chains around the Paleo-Tethys Ocean thus could have prevented its seawater circulation with oceans outboard through which the water exchange with Panthalassa and Neo-Tethys could only take place through the narrow channels/saddles between blocks/terrane near the equator. Thus, the ocean was relatively restricted in the mid-latitudes. The magnesium isotope data of dolomite also confirm a restricted oceanic environment in the Paleo-Tethys (Hu et al., 2021). Meanwhile, without contact with the polar ice caps on both sides, the decreasing temperature difference between surface water and bottom water in the Paleo-Tethys as a result of the P-T thermal maximum (Huey and Ward, 2005; Joachimski et al., 2012; Benton and Newell, 2014; Chen et al., 2020) could have also played an important role in driving the oceanic stagnation and stratification.

From the perspective of paleogeographic contexts, the Panthalassa extended from the equator to polar regions within which the sea surface seawater could approach the ice rafts in high latitudes, causing inversions in temperature and density gradient of water column (Fig. 11), thereby enhancing the ocean circulation in the Panthalassa regime. The water exchange between the Neo-Tethys and Panthalassa could freely occur without blocking (Fig. 1A), which facilitated increasing oxygen contents in the Neo-Tethys, so that oxic conditions overwhelmingly dominated the shallow carbonate platforms/shelves from the Late Permian to the Early Triassic.

It is obvious that the onset of anoxia was diachronous in some sections. However, the timing of the mass extinction in non-hypoxic areas such as at Gyanyima in Tibet was approximately synchronous with that in other anoxic environments (Shen et al., 2010; Garbelli et al., 2016). These facts indicate that the ocean anoxia may have not acted as the exclusive killing mechanism for the organism, while the aerobic areas also did not become a refuge for organisms. Therefore, the anoxia possibly was an interactive outcome of other "precursive causes" of environmental perturbations, such as volcanic eruptions and warming (Penn et al., 2018), and then in turn exacerbated the ecosystem through interactions of other environmental forces. On the other hand, the prolonged anoxia in the Early Triassic may have delayed the restoration of the ecosystem and biotic recovery (Lau et al., 2016; Penn et al., 2018).

## 6. Conclusions

Detailed analyses of Fe speciation, major and trace element concentrations and mercury contents during the Late Permian are reported from Xibeixiang section, which was deposited in an intrashelf basin at the northern margin of Yangtze Block, SW China. Our data distinguish four distinct paleo-redox intervals (I-IV) based on Fe–Mo–U–V geochemical datasets. Of these, interval I, spanning from the upper

Wuchiaping Formation to the basal Dalong Formation, was characterized by mainly an oxic-suboxic condition during deposition. Interval II (middle Dalong Formation) was dominated by euxinic conditions. Interval III, accommodating in the upper Dalong Formation, recorded a ferruginous-dominant anoxic condition interrupted by euxinic spells. Ultimately, the interval IV, spanning from the uppermost Permian to the lowermost Triassic, was dominated by a ferruginous condition with occasional incursion of oxic episode just across the P-T boundary. Moreover, the primary productivity stayed fairly low in interval I, and culminated in interval II, followed by an apparent decline in interval III, and remained lowstand in interval IV. Coincidentally, the high primary productivity was temporally synchronous with the euxinic interval, so that the productivity likely played an important role in formation of the water redox states. It is noted that intensive volcanic activities as shown by concentrated tuffaceous layers and Hg/TOC anomalies occurred slightly earlier than P-T boundary, which can be well correlated with those reported at more than 20 localities over the world. Furthermore, by comparison the redox data of the Late Permian from different paleogeographic and paleobathymetrical settings around the world, we found that the paleogeographic contexts likely played a critical role in controlling the spatial heterogeneity and temporal diachroneity of the marine redox states during the Late Permian. The Paleo-Tethys was relatively restricted in oceanic circulation, thus subject to more severely anoxic condition than Neo-Tethys and Panthalassa.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.marpetgeo.2022.105633>.

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