

Synthesis of Some Novel Pyrazolo[1,5-a]quinazolines and Their Fused Derivatives

Eman Ali Ragab¹, Nadia Hanafy Metwally¹, Mona Said Mohamed¹

¹Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

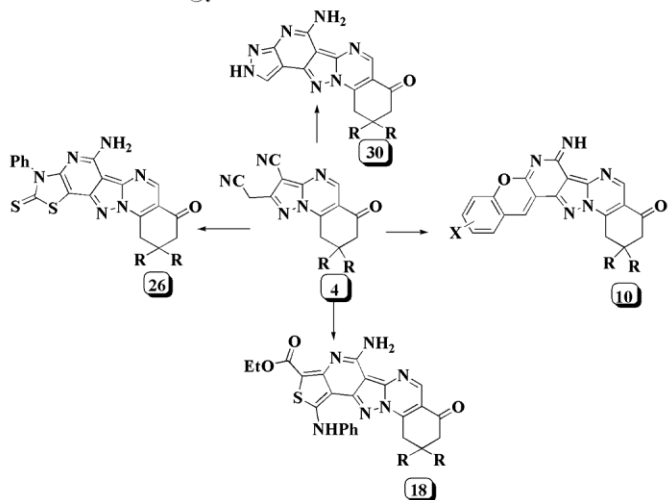
Corresponding author Mona Said Mohamed E-mail: nhmmohamed@yahoo.com

Abstract

New pyrazolo[1,5-a]quinazoline-3-carbonitriles **4a,b** were obtained via cyclocondensation of 5-amino-3-cyanomethyl-1H-pyrazole-4-carbonitrile (**1**) with enaminones of 1,3-cyclohexanedione derivatives **2a,b** in refluxing glacial acetic acid. The condensation of compounds **4a,b** with various aromatic aldehydes furnished the corresponding arylidene derivatives **6a-j**. On the other hand, condensation of **4a,b** with o-hydroxybenzaldehydes yielded the polyheterocyclic compounds **10a-h**. Coupling of compounds **4a,b** with aryldiazonium chlorides led to formation of 2-arylhydrazono derivatives **12a-h**. Also, reaction of compounds **4a,b** with phenyl isothiocyanate, followed by addition of ethyl chloroacetate and chloroacetonitrile afforded the polyheterocyclic compounds based on pyrazolo[1,5-a]quinazoline core. The reaction of compounds **4a,b** with phenyl isothiocyanate and elemental sulfur gave the thiazole-2-thione derivatives **25a,b**. The reaction of enamines of compounds **4a,b** with each of hydrazine hydrate and guanidine hydrochloride afforded pyrazolo[4'',3''':5',6']pyrido[4',3':3,4]pyrazolo[1,5-a]quinazolin-8-ones **30a,b** and pyrimido[5'',4''':5',6']pyrido[4',3':3,4]pyrazolo[1,5-a]quinazolin-9(10H)-ones **33a,b**, respectively. The structure of all the newly synthesized compounds was elucidated by elemental analyses and spectral data. The plausible mechanisms have been postulated to account for their formation.

Graphical Abstract

Synthesis of Some Novel Pyrazolo[1,5-*a*]quinazolines and Their Fused Derivatives
Eman Ali Ragb, Nadia Hanafy Metwally* and Mona Said Mohamed
Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt
E-mail: nhmmohamed@yahoo.com



KEYWORDS: 2-(Cyanomethyl)-6-oxo-pyrazolo[1,5-*a*]quinazolines; pyrazolo[4'',3'':5',6']pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazolinones; pyrimido[5'',4'':5',6']pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazolinones

INTRODUCTION

Pyrazolo[1,5-*a*]quinazoline derivatives have a broad spectrum of biological activity such as anticancer,^[1-3] anti-inflammatory^[4] and antimicrobial^[5]. Also, they are effective as antisecretory agents^[6] and as poly(ADP-ribose) polymerase-1 (PARP-1) inhibitors^[7].

Therefore, based on the previous reports and in continuation of our interest in the synthesis of new bioactive heterocyclic compounds,^[8-17] we report herein the preparation of some new polyheterocyclic compounds containing pyrazolo[1,5-*a*]quinazoline moiety starting with the unreported hitherto 2-(cyanomethyl)-6-oxo-pyrazolo[1,5-*a*]quinazolin-3-carbonitriles **4a,b** as key intermediates for our objectives.

RESULTS AND DISCUSSION

Chemistry

The reaction of 5-amino-3-cyanomethyl-1*H*-pyrazole-4-carbonitrile (**1**) with enaminone of 1,3-cyclohexanedione derivatives **2a,b** in refluxing glacial acetic acid, afforded the colored products **4a,b** (Scheme 1). The structure of the isolated products was determined by their elemental analyses and spectral data (IR, ¹H NMR, ¹³C NMR and MS). The IR spectrum of the isolated product **4b** showed absorption bands at ν_{\max} 2227 and 1687 cm⁻¹ corresponding to CN and CO groups, respectively. The ¹H NMR spectrum of compound **4b** revealed a singlet signal at $\delta = 4.47$ ppm assigned to the methylene protons (CH₂CN) and a singlet signal at $\delta = 9.02$ ppm attributable to the pyrimidine proton. Also, ¹³C NMR spectrum of **4b** showed characteristic three signals at $\delta = 17.3$, 154.5 and 194.3 ppm assigned for methylene (CH₂CN), pyrimidine and carbonyl (CO) carbons, respectively. Also, elemental analysis with mass spectrum of **4b** showed a correct molecular ion peak at $m/z = 279$ (M⁺). Consistent with the formula C₁₅H₁₃N₅O. Thus, the reaction between aminopyrazole **1** and enaminone **2b** proceeded through an initial *Michael* addition of the exocyclic amino group in the aminopyrazole **1** to the α,β -unsaturated moiety in the enaminone **2b** yielded the corresponding acyclic non-isolable intermediates which undergo dehydrative cyclization and aromatization to give the final product **4b** rather than the isomeric structure **3b** (Scheme 1).

Condensation of compounds **4a,b** with various aromatic aldehydes **5a-e** in absolute ethanol in the presence of few drops of piperidine afforded the corresponding arylidene

derivatives of pyrazolo[1,5-*a*]quinazolines **6a-j** (Scheme 2). The IR spectrum of the isolated product **6b** taken as a typical example of the prepared series showed absorption bands at ν_{\max} 2226 and 1688 cm^{-1} corresponding to CN and CO groups, respectively. The ^1H NMR spectrum of **6b** revealed the presence of a singlet signal at $\delta = 3.88$ ppm assigned to the methoxy protons, in addition to a singlet signal at $\delta = 8.34$ ppm attributable to vinylic proton and a singlet signal at $\delta = 9.11$ ppm assigned for pyrimidine proton. Also, the elemental analyses are in agreement with the structure **6b** (Scheme 2).

On the other hand, refluxing compounds **4a,b** with *o*-hydroxybenzaldehydes **7a,b** in dry ethanol containing few drops of piperidine produced, in each case, yellowish orange solid of melting point above 300°C (Scheme 3). The IR spectrum of compound **9a** showed absorption bands at ν_{\max} 3372, 2228 and 1683 cm^{-1} corresponding to NH, CN and CO groups, respectively. The ^1H NMR spectrum of **9a** revealed the presence of a D_2O -exchangeable signal at $\delta = 9.13$ ppm assigned to the NH proton. The elemental analyses are in agreement with the structures **9a-d** (Scheme 3). Refluxing the latter compounds in sodium ethoxide solution afforded products **10a-d**. The structure of the isolated products **10a-d** was inferred from correct analytical and spectral data. As typical example, the IR spectrum of **10a** showed disappearance of an absorption band corresponding to nitrile group near $\nu_{\max} \sim 2200$ cm^{-1} and instead revealed an absorption band at ν_{\max} 3371 cm^{-1} due to the NH group. The ^1H NMR spectrum of **10a** revealed a D_2O -exchangeable signal at $\delta = 9.17$ ppm attributable to NH proton.

Subsequently, we extended our study in synthesizing new azodye derivatives. Thus, compounds **4a,b** were coupled with the diazotized aromatic amines **11a-d** in *N,N*-dimethylformamide (DMF) in the presence of anhydrous sodium acetate at 0-5°C leading to yellow colored compounds **12a-h** (Scheme 4). The IR spectrum of the isolated compound **12a** which was taken as a typical example of the prepared series, showed absorption bands at ν_{\max} 3238, 2224 and 1687 cm^{-1} corresponding to the NH, CN and CO groups, respectively. The ^1H NMR spectrum of the isolated product **12a** revealed a singlet signal at $\delta = 8.99$ ppm due to the pyrimidine proton, in addition, a D_2O -exchangeable signal appeared at $\delta = 12.12$ ppm assigned to the NH proton, beside the other expected signals for aryl protons. ^{13}C NMR spectrum of **12a** showed characteristic signals at $\delta = 155.6$ and 194.0 ppm assigned for pyrimidine and carbonyl carbons, respectively. The mass spectrum of **12a** showed correct molecular ion peak at $m/z = 285$ (M^+). However, our attempts to cyclize these hydrazone derivatives **12a-h to give iminopyridazino[4,5':3,4]pyrazolo[1,5-*a*]quinazoline derivatives 13a-h** failed under all the different conditions reported for such cyclization^[18,19] (Scheme 4).

The electronic absorption spectra of the products **12a-h** in DMF revealed, in each case, two absorption bands in the regions λ_{\max} 413-390 and 314-303 nm. This absorption pattern seems to indicate that the studied compounds **12a-h** exist predominantly in solution as hydrazone form **A**. The reason is that such an absorption pattern is similar to that of typical hydrazones^[20,21]. ^1H NMR spectra provide additional evidence that they have the hydrazone form **A** rather than the azo-forms **B**. For instance, the ^1H NMR spectra of compounds **12** in $\text{DMSO-}d_6$ exhibit, in each case, one singlet signal near 12.0 ppm

due to the =NNH-proton and this is substantiated by the literature data which indicate that the chemical shift (δ) of hydrazone NH resonance is usually observed near 13.0 ppm^[22,23] (see Table 1).

The reaction of **4a,b** with phenyl isothiocyanate in *N,N*-dimethylformamide (DMF) in the presence of potassium hydroxide, followed by addition of ethyl chloroacetate **15** at room temperature, afforded in each case a single product (**17a** or **17b**, as evidenced by TLC) (Scheme 6). The IR spectrum of **17a** showed absorption bands at ν_{\max} 3453, 3065, 2225, 1710 and 1691 cm^{-1} due to the NH_2 , NH, CN and two CO groups, respectively. The ^1H NMR spectrum of compound **17a** revealed a triplet signal at $\delta = 1.12$ ppm with $J = 6.9$ Hz assigned to the methyl (CH_3) protons and a quartet signal at $\delta = 4.13$ ppm with $J = 6.9$ Hz assigned to methylene (CH_2) protons of ester group. In addition, two D_2O exchangeable signals at $\delta = 6.92$ and 10.21 ppm attributable to NH_2 and NH protons, beside the other expected signals for pyrimidine and aromatic protons. Elemental analyses together with the spectral data are in agreement with structure **17**. Refluxing compounds **17a,b** in sodium ethoxide solution afforded compounds **18a,b** (Scheme 5). The IR spectrum of **18b** showed absorption bands at ν_{\max} 3435, 3073, 1712 and 1693 cm^{-1} due to the NH_2 , NH and two CO groups, respectively. The ^1H NMR spectrum of compound **18b** revealed signals at $\delta = 1.17$ and 4.14 ppm with $J = 6.9$ ppm assigned to the methyl (CH_3) and methylene (CH_2) protons for ester group, respectively, in addition to, two D_2O exchangeable signals at $\delta = 6.89$ and 10.12 ppm assigned to NH_2 and NH protons, beside the other expected signals. Also, elemental analyses are in agreement with structure **18b**. The formation of compounds **18a,b** assumed to proceed through

formation of thiophene derivatives **17** *via* **non-isolable intermediate 16**, followed by **addition** of amino function to cyano group to generate thieno[3'',4'':5',6']pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazoline derivatives **18a,b** (Scheme 5).

In the same manner, the reaction of the non-isolable potassium salt of pyrazolo[1,5-*a*]quinazoline derivative **14** with chloroacetonitrile **19** in DMF at room temperature, afforded a single product identified as thiophene derivatives **21a,b** *via* non-isolable intermediate *S*-alkyl derivative **20** (Scheme 6). Refluxing compounds **21a,b** in a solution of sodium ethoxide afforded products identified as thieno[3'',4'':5',6']pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazoline derivatives **22a,b** (Scheme 6).

The reaction of compounds **4a,b** with phenyl isothiocyanate and elemental sulfur in DMF in the presence of triethylamine as catalyst, gave isolated products **25a,b** (Scheme 7). The IR spectrum of product **25b** showed absorption bands at ν_{\max} 3293, 2203 and 1682 cm^{-1} due to the NH_2 , CN and CO groups, respectively. The ^1H NMR spectrum of compound **25b** revealed a multiplet signal at $\delta = 7.29\text{-}7.95$ ppm assigned to the NH_2 and aromatic protons. Mass spectrum showed correct molecular ion peak at m/z 446 (M^+). Furthermore, elemental analyses are in agreement with the proposed structure **25b**. However, trials to cyclize the compounds **25a,b** into 5-amino-3-phenyl-2-thioxothiazolo[5'',4'':5',6']pyrido-[4',3':3,4]pyrazolo[1,5-*a*]quinazolin-8(9*H*)-one derivatives **26a,b** failed under reflux in a solution of sodium ethoxide (Scheme 7).

Enamines are interesting class of organic compounds; the special value of these compounds is due to their utility as valuable intermediate for the synthesis of several interesting compounds^[24-29]. Based on this, we prepared the enamine derivatives of compounds **4a,b**. Thus, the reaction of compounds **4a,b** with dimethylformamide-dimethylacetal (DMF-DMA) in dry dioxane afforded the corresponding enamine derivatives **27a,b** (Scheme 8). The IR spectrum of the isolated product **27b** showed absorption bands at ν_{\max} 2217, 2192 and 1692 cm^{-1} corresponding to the two CN and CO groups, respectively. The ^1H NMR spectrum of **27b** revealed the presence of two singlet signals at $\delta = 3.28$ and 3.33 ppm, corresponding to the methyl protons [-N(Me)₂] and a singlet signal at $\delta = 7.89$ ppm attributable to the olefinic proton. The ^{13}C NMR spectrum of the isolated product **27b** showed three characteristic signals at $\delta = 153.4$, 156.6 and 194.2 ppm assigned to pyrimidine, olefinic [=CHN(Me)₂] and carbonyl carbons, respectively. Treatment of the enamine derivatives **27a,b** with hydrazine hydrate in DMF in the presence of few a drops of piperidine afforded brown solid products **30a,b** (Scheme 8). The IR spectrum of compound **30b** showed absorption bands at ν_{\max} 3287, 3071 and 1692 cm^{-1} due to the NH₂, NH and CO groups, respectively. Its ^1H NMR spectrum revealed two *D*₂*O* exchangeable singlet signals at $\delta = 7.63$ and 13.13 ppm attributable to the NH₂ and NH protons, beside the other expected signals assigned to aromatic and pyrimidine protons. Elemental analyses and mass spectra are in consistent with structure **30a,b** (see Exp.).

The reaction of enamines **27a,b** with guanidine hydrochloride in DMF in the presence of anhydrous sodium acetate furnished the fused pyrimidine derivatives **33a,b** (Scheme 9).

EXPERIMENTAL

Materials And Methods

Melting points were determined on an Electrothermal (9100) apparatus and are uncorrected. The IR spectra were recorded as KBr pellets on a Perkin Elmer 1430 spectrophotometer. ¹H NMR and ¹³C NMR spectra were recorded in deuterated dimethylsulfoxide at 300 and 75 MHz, respectively, on a Varian Gemini NMR spectrometer using tetramethylsilane as internal reference and the results are expected as δ value. Mass spectra were taken on a Shimadzu GCMS-QP 1000 Ex mass spectrometer at 70 eV. UV spectra and elemental analyses were carried out at the Microanalyses Center of Cairo University, Giza, Egypt.

Synthesis Of 2-(Cyanomethyl)-6-Oxo-6,7,8,9-Tetrahydropyrazolo[1,5-A]Quinazoline-3-Carbonitriles (**4a,B**).

A mixture of 5-amino-2-cyanomethyl-1*H*-pyrazole-4-carbonitrile **1** (0.01 mol) and 2-(dimethylaminomethylene)-1,3-cyclohexanedione derivatives **2a,b** (0.01 mol) in glacial acetic acid (10 ml) was heated under reflux for 6 h, then allowed to cool. The solid was collected by filtration and recrystallized from an ethanol-dioxane mixture to give compounds **4a,b**.

2-(Cyanomethyl)-6-Oxo-6,7,8,9-Tetrahydropyrazolo[1,5-A]Quinazoline-3-

Carbonitrile (4a).

Brown crystals, yield 60%, m.p = 205°C, $\nu_{\max} / \text{cm}^{-1}$ (KBr) 2225 (2CN), 1691 (CO); ^1H NMR (DMSO) δ = 2.24-2.28 (m, 2H, CH₂), 2.68 (t, 2H, J = 6.6 Hz, CH₂), 3.43 (t, 2H, J = 6 Hz, CH₂), 4.65 (s, 2H, CH₂), 9.09 (s, 1H, pyrimidine-H); ^{13}C NMR (DMSO) δ = 17.3, 19.7, 23.7, 36.6, 82.4, 111.8, 115.9, 116.8, 151.1, 151.4, 152.8, 156.4, 194.4; Anal. Calcd for C₁₃H₉N₅O: C, 62.15; H, 3.61; N, 27.87. Found: C, 62.32; H, 3.42; N, 27.62 %.

Synthesis Of Compounds 6a-J

General Procedure

A solution of the appropriate compound **4a,b** (0.01 mol) in ethanol (20 ml) containing a few drops of piperidine was added to the appropriate aldehyde **5a-e** (0.01 mol). The mixture was heated under reflux for 5 h and the reaction mixture was followed by TLC. The solid product that precipitated by cooling was filtered off and recrystallized from *N,N*-dimethylformamide to give the respective products **6a-j**. The physical constants and spectral data are shown below:

2-(1-Cyano-2-Phenylvinyl)-6-Oxo-6,7,8,9-Tetrahydropyrazolo[1,5-A]Quinazoline-3-

Carbonitrile (6a).

Reddish brown crystals, yield 65%, m.p = 240°C, $\nu_{\max} / \text{cm}^{-1}$ (KBr) 2225 (2CN), 1691 (CO); ^1H NMR (DMSO) δ = 2.22-2.26 (m, 2H, CH₂), 2.61 (t, 2H, J = 6.3 Hz, CH₂), 3.39 (t, 2H, J = 6 Hz, CH₂), 7.49-7.65 (m, 5H, Ar), 8.51 (s, 1H, CH), 9.09 (s, 1H, pyrimidine-

H); Anal. Calcd for C₂₀H₁₃N₅O: C, 70.79; H, 3.86; N, 20.64. Found: C, 70.59; H, 3.68; N, 20.92 %.

Synthesis Of Compounds 9a-D

General Procedure:

To a solution of each of **4a,b** (0.01 mol) was added the appropriate *o*-hydroxybenzaldehydes **7a,b** (0.01 mol) in absolute ethanol (20 ml) containing few drops of piperidine under reflux for 3 h. The solid product so formed was collected by filtration, washed with ethanol and recrystallized from an ethanol-dioxane mixture to give the respective products **9a-d**.

2-(2-Imino-2*H*-Chromen-3-yl)-6-Oxo-6,7,8,9-Tetrahydropyrazolo[1,5-*A*]Quinazoline-3-Carbonitrile (**9a**).

Brownish yellow crystals, yield 94%, m.p >300°C, ν_{\max} /cm⁻¹ (KBr) 3372 (NH), 2228 (CN), 1683 (CO); ¹H NMR (DMSO) δ = 2.23-2.26 (m, 2H, CH₂), 2.46 (t, 2H, *J* = 6.6 Hz, CH₂), 3.14 (t, 2H, *J* = 6 Hz, CH₂), 6.52-7.44 (m, 4H, Ar), 7.97 (s, 1H, CH), 8.96 (s, 1H, pyrimidine-H), 9.13 (s, 1H, NH); Anal. Calcd for C₂₀H₁₃N₅O₂: C, 67.60; H, 3.69; N, 19.71. Found: C, 67.43; H, 3.51; N, 19.45 %.

CONCLUSION

In conclusion, we have described an efficient synthesis of some novel pyrazolo[1,5-*a*]quinazolin-6-ones and their fused derivatives *via* the reaction of readily accessible starting material, 2-cyanomethylpyrazolo[1,5-*a*]quinazolinones with various aromatic

aldehydes furnished the corresponding arylidene derivatives. Also, formation of polyheterocyclic compounds based on pyrazolo[1,5-*a*]quinazoline core. Also, the reaction of enamines of 2-cyanomethylpyrazolo[1,5-*a*]quinazolinones with each of hydrazine hydrate and guanidine hydrochloride afforded pyrazolo[4'',3''':5',6']pyrido[4',3':3,4]pyrazolo[1,5-*a*]8-ones and pyrimido[5'',4''':5',6']-pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazolin-9(10*H*)-ones.

REFERENCES

1. Taliani, S.; Pugliesi, I.; Barresi, E.; Salerno, S.; Marchand, C.; Agama, K.; Simorini F.; Motta, C. L.; Marini, A. M.; Leva, F. S. D.; Marinelli, L.; Cosconati, S.; Novellino, E.; Pommier, Y.; SantoR. D.; Settimo, F. D. Phenylpyrazolo[1,5-*a*]-quinazolin-5(4*H*)-one: A suitable scaffold for the development of noncamptothecin topoisomerase I (Top1) inhibitors. *J. Med. Chem.* **2013**, 56, 7458-7462.
2. Chimichi, S.; Boccalini, M.; Selleri, S.; Costaqi, C.; Guerrini, G.; Viola, G. On the reactivity of 6-acetyl-7-(2-dimethylaminovinyl)pyrazolo[1,5-*a*]pyrimidines with 1,3- and 1,4-bisnucleophiles. *Org. Biomol. Chem.* **2008**, 6, 739-744.
3. Shekarrao, K.; Kaishap, P. P.; Saddanapu, V.; Addlagatta, A.; Gogoi, S.; Boruah, R. C. Microwave-assisted palladium mediated efficient synthesis of pyrazolo[3,4-*b*]pyridines, pyrazolo[3,4-*b*]quinolines, pyrazolo[1,5-*a*]pyrimidines and pyrazolo-[1,5-*a*]quinazolines. *R. S. C. Adv.* **2014**, 4, 24001-24006.
4. Hussein, M. A. Synthesis, anti-inflammatory, and structure antioxidant activity relationship of novel 4-quinazoline. *J. Med. Chem. Res.* **2013**, 22, 4641-4653.

5. Alexander, E. J. U. S., Patent, 4,5-Dihydro-5-oxopyrazolo[1,5-*a*]quinazoline-3-carboxylic acid derivatives. 4105766 (1978).
6. Nazeer, A.; Perveen, N.; Khan, M. A.; Munawar, M. A.; Lin, W. O. Synthesis and antibacterial activity of pyrazolo[1,5-*a*]quinazoline-3-carbonitriles. *Asian J. of Chem.* **2013**, *25*, 7705-7709.
7. Niu, M.; Gu, Y. An in silico protocol for identifying potential poly(ADP-ribose) polymerase-1 (PARP-1) inhibitors from chemical database. *New J. Chem.* **2015**, *39*, 1060-1066.
8. Elgemeie, G. E. H.; Mansour O. A.; Metwally, N. H. Synthesis and anti-HIV activity of different noval nonclassical nucleosides. *Nucleosides and Nucleotides* **1999**, *18*, 113-123.
9. Elgemeie, G. E. H.; Metwally, N. H. Synthesis of structurally related purines: benzimidazo[1,2-*a*]pyridines, benzimidazo-[1,2-*c*]pyrimidines, and pyrazolo[1,5-*a*]pyrimidines. *Monatsh Chem.* **2000**, *131*, 779-785.
10. Abdelrazek, F. M.; Metwally, N. H. Some reactions with ω -bromoacetophenone: A novel synthesis of some polysubstituted pyridine, pyrimidine, pyrazolo[1,2-*a*]pyrimidine derivatives. *A finidad* **2003**, *60*, 554-557.
11. Metwally, N. H.; Abdelrazek, F. M. Reaction of anthranilo nitrile with some active methylene reagents, synthesis of some new quinoline and quinazoline derivatives. *Synth. Comm.* **2005**, *35*, 2481-2487.
12. Abdelrazek, F. M.; Metwally, N. H. Synthesis of some new *N*-substituted pyrroles, pyrrolo[1,2-*a*]quinazoline, and diaza-as-in dacene derivatives. *Synth. Comm.* **2006**, *36*, 83-89.

13. Abdelrazek, F. M.; Metwally N. H. Novel synthesis of N-arylpyrrole, pyrrolo[1,2-*a*]quinazoline, and pyrrolo[3,4-*d*]pyridazine derivatives. *Synth. Comm.* **2009**, 39, 4088-4099.
14. Metwally, N. H.; Abdallah, M. A.; Mosselhi, M. A.; El-Desoky, E. A. Synthesis and antimicrobial activity of some new *N*-glycosides of 2-thioxo-4-thiazolidinone derivatives. *Carbohydr. Res.* **2010**, 345, 1135-1141.
15. Metwally, N. H. 3-Aryl-2-sulfanyl propenoic acids as precursors for some novel (*Z*)-5-substituted-2-trichloromethyl-4-thiazolidinones. *Arkivoc* **2011**, x, 254-265.
16. Metwally, N. H. 2-Keto-3-mercaptocinchronic acids as precursors for novel thiazino[6,5-*c*]quinoline-1,5-dione derivatives. *Synth. Comm.* **2013**, 43, 398-405.
17. Metwally, N. H.; Abdallah, M. A.; Almabrook, S. A. Pyrazolo[1,5-*a*]pyrimidine derivatives as precursor for some novel pyrazolo[1,5-*a*]pyrimidines and tetraheterocyclic compounds. *J. Heterocycl. Chem.* **2016**, in press.
18. Hafez, T. S.; Osman, S.; Yosef, H. A. A.; Abdel-All, A.; Hassan, A.; El-Sawy, A. A.; Abdallah, M. M.; Youns, M. Synthesis, Structural elucidation, and in vitro antitumor activities of some pyrazolopyrimidines and Schiff bases derived from 5-amino-3-(arylamino)-1*H*-pyrazole-4-carboxamides. *Sci Pharm.* **2013**, 81, 339-357.
19. Krishnaiah, A.; Narsaiah, B. A facile method for the synthesis of novel fluorinated pyrido[2',3':3,4]pyrazolo[1,5-*a*]pyrimidines. *J. Fluorine Chem.* **2001**, 109, 183-187.
20. Shawali, A. S.; Harb, N. M. S.; Badhdah, K. O. A study of tautomerism in diazonium coupling products of 4-hydroxycoumarin. *J. Heterocycl. Chem.* **1985**, 22, 1397-1403.
21. Joneo, R.; Ryan, A. J.; Stemhell, S.; Wrifgt, S. E. The structures of some-5-pyrazolone and derived 4-arylazo-5-pyrazolones. *Tetrahedron* **1963**, 19, 1497-1507.

22. G. E. Wright; *J. Heterocycl. Chem.*, **20**, 1037 (1989).
23. Shawali, S. A.; Albar, H. A. Kinetics and mechanism of dehydrochlorination of N-aryl-Cethoxycarbonylformohydrazidoyl chlorides. *Can. J. Chem.* **1986**, 64, 871-875.
24. Al-Zaydi, K. M. Microwave assisted synthesis, Part 1: Rapid solventless synthesis of 3-substituted coumarins and benzocoumarins by microwave irradiation of the corresponding enamines. *Molecules* **2003**, 8, 541-555.
25. Al-Saleh, B.; El-Asery, M. A.; Abdelaziz, R. S.; Elnagdi, M. H. Enaminones in heterocyclic synthesis: Synthesis and chemical reactivity of 3-anilino-1-substituted-2-propene-1-one. *J. Heterocycl. Chem.* **2005**, 42, 563-566.
26. Rossignol, E.; Youssef, A.; Moreau, P.; Prudhomme, M.; Anizon, F. Synthesis of aminopyrimidylindoles structurally related to meridianins. *Tetrahedron* **2007**, 63, 10169-10179.
27. Echalié, A.; Bettayeb, K.; Frandin, Y.; Lozach, O.; Clément, M.; Valette, A.; Liger, F.; Marquet, B.; Morris, J. C.; Endicott, J. A.; Joseph B.; Meijer, L. Meriolins (3-(pyrimidin-4-yl)-7-azaindoles): synthesis, kinase inhibitory activity, cellular effects, and structure of a CDK2/cyclin A/meriolin complex. *J. Med. Chem.* **2008**, 51, 737-751.
28. Vishwakama, J. N.; Dutta, M. C.; Chanda, K.; Das, B.; Laskar, M. A.; Nongkhaw, R. L. Synthesis and antimicrobial activities of novel 5-isonicotinoyl-1,2,3,4-tetrahydropyrimidines and bis-(5-isonicotinoyl-1,2,3,4-tetrahydro-pyrimidine). *Arkivoc* **2009**, xii, 131-141.
29. Al-Mousaw, S. M.; El-Asery, M. A.; Elnagdi, M. H. Enaminones in heterocyclic synthesis: a novel route to tetrahydropyrimidines, dihydropyridines, triacylbenzenes and naphthofurans under microwave irradiation. *Arkivoc*, **2010**, 15, 58-67.

30. Carboni, R. A.; Coffman, D. D.; Howard, E. G. Cyanocarbon chemistry. XI.1 malonitrile dimer. *J. Am. Chem. Soc.* **1958**, 80, 2838-2840.

31. Claramunt, R.M.; Lopez, C.; Medina, C. P.; Pinilla, E.; Torres, M. R.; Elguero, J. Synthesis and structural study of tetrahydroindazolones. *Tetrahedron* **2006**, 62, 11704-11713.

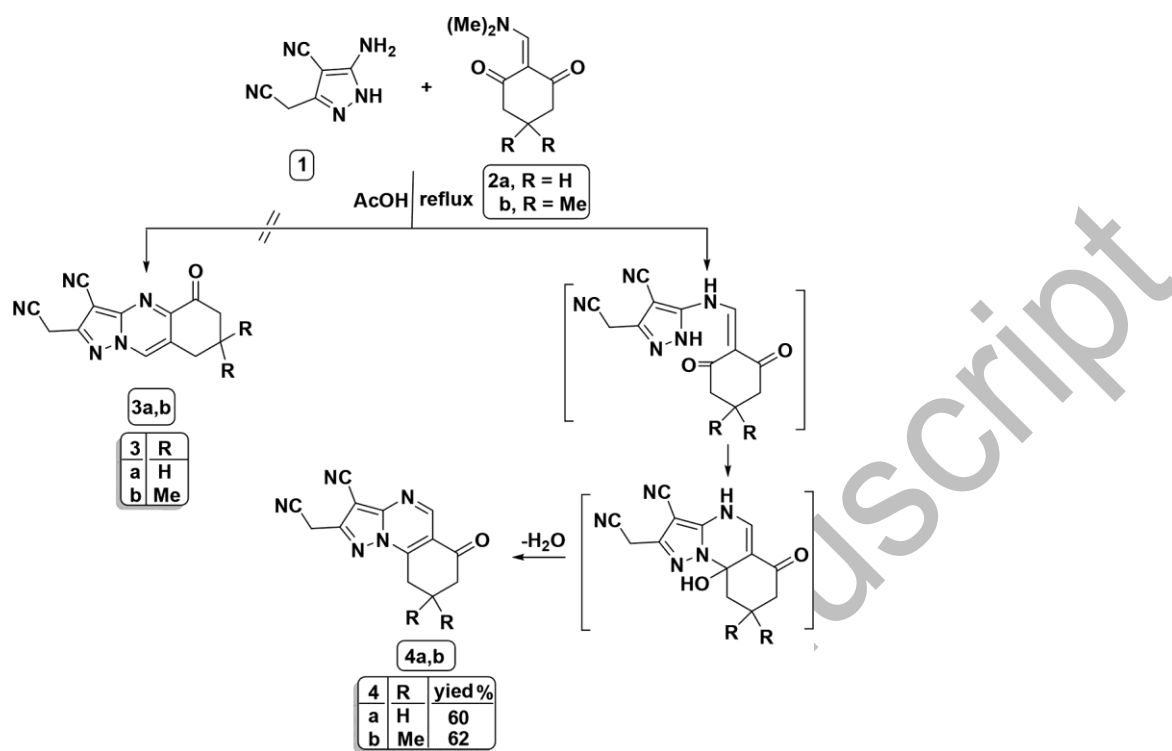
Accepted Manuscript

Table 1. Electronic absorption spectral data of compounds 12a-h in *N,N*-dimethylformamide (DMF)

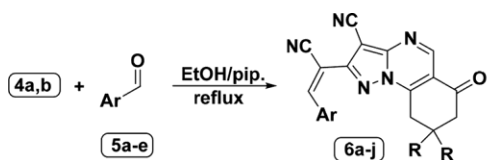
Compound No.	λ_{max} (nm) DMF	Compound No.	λ_{max} (nm) DMF
12a	390, 313	12e	394, 312
12b	392, 310	12f	400, 307
12c	413, 303	12g	400, 314
12d	398, 313	12h	395, 308

Accepted Manuscript

Scheme 1. Synthetic route to 2-cyanomethyl-6-oxo-pyrazolo[1.5-*a*]quinazolines **4a,b**



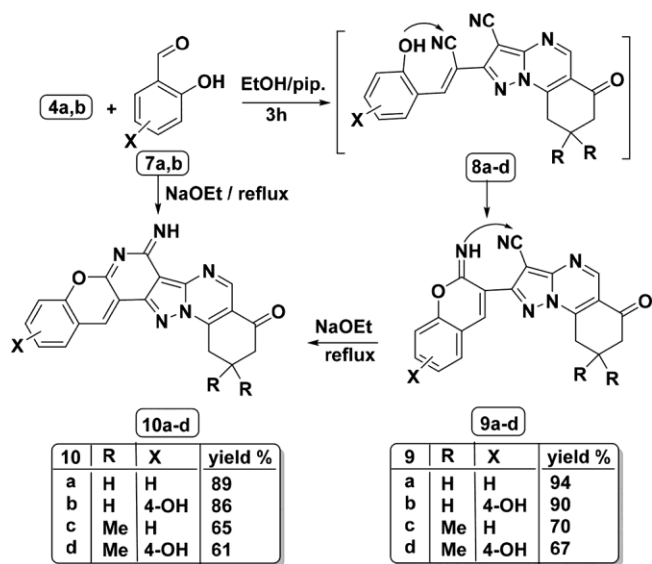
Scheme 2. Synthetic route to 2-(1-cyano-2-arylvinyl)-6-oxo-pyrazolo[1,5-*a*]-quinazolines **6a-j**



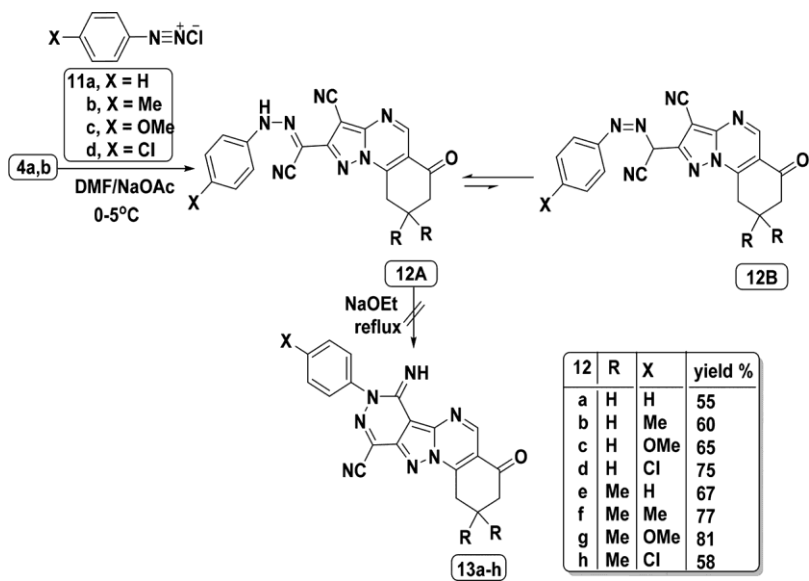
6	R	Ar	yield %	6	R	Ar	yield %
a	H	Ph	65	f	Me	Ph	89
b	H	4-OMeC ₆ H ₄	74	g	Me	4-OMeC ₆ H ₄	91
c	H	4-ClC ₆ H ₄	63	h	Me	4-ClC ₆ H ₄	89
d	H	3,4(OCH ₂ O)C ₆ H ₃	62	i	Me	3,4(OCH ₂ O)C ₆ H ₃	92
e	H	2-furyl	79	j	Me	2-furyl	75

Accepted Manuscript

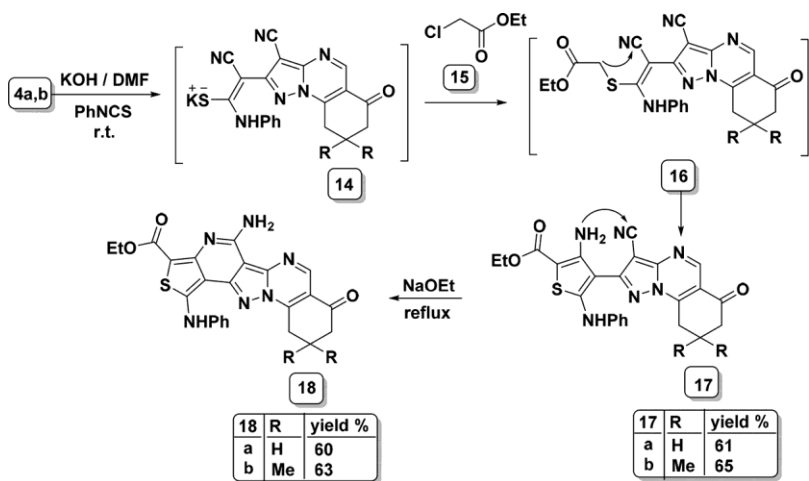
Scheme 3. Synthetic pathway to 2-(2-imino-2*H*-chromen-3-yl)-6-oxo-pyrazolo-[1,5-*a*]quinazolines **9a-d** and fused pyrazolo[1,5-*a*]quinazolines **10a-d**



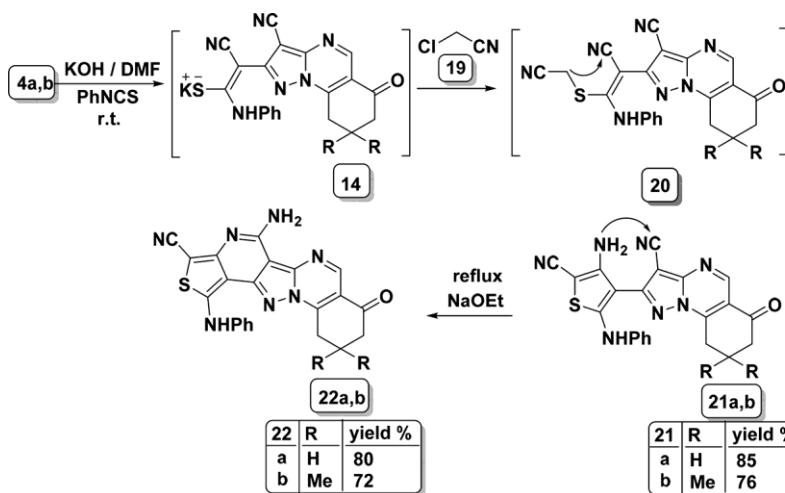
Scheme 4. Synthetic route to 2-arylhydrazono-6-oxo-pyrazolo[1,5-*a*]quinazolines **12a-h**



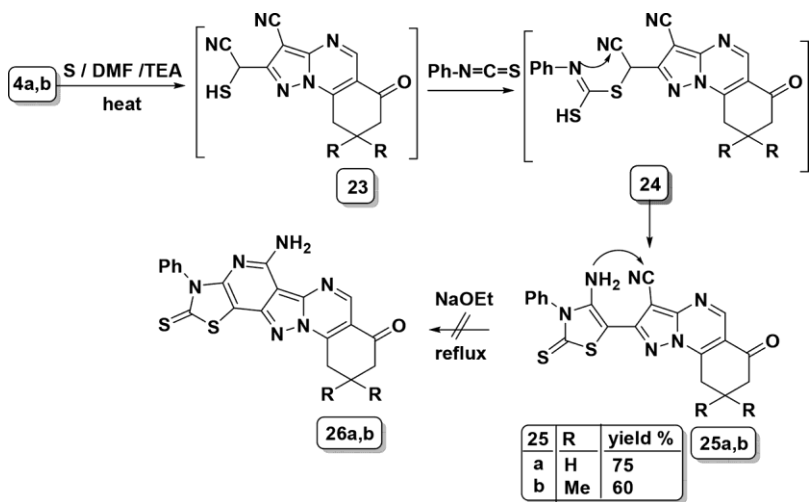
Scheme 5. Synthetic pathway to pyrazolo[1,5-*a*]quinazolines **17** and polyfused heterocyclic compounds **18**



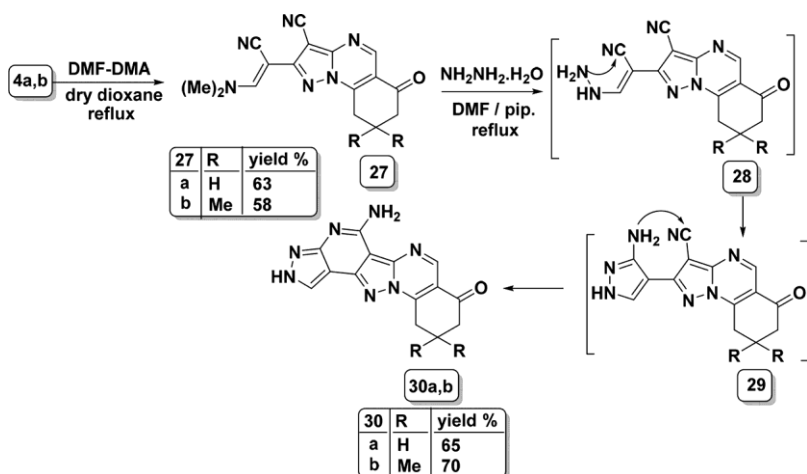
Scheme 6. Synthetic pathway to pyrazolo[1,5-*a*]quinazolines **21** and polyfused heterocyclic compounds **22**



Scheme 7. Synthetic pathway to 2-(4-amino-3-phenyl-2-thioxo-thiazol-5-yl)-6-oxo-pyrazolo[1,5-*a*]quinazoline-3-carbonitriles **25**



Scheme 8. Synthetic pathway to 2-[1-cyano-2-(dimethylamino)vinyl]-6-oxopyrazolo-[1,5-*a*]quinazolines **27a,b** and 5-amino pyrazolo[4'',3'':5',6']pyrido-[4',3':3,4]pyrazolo[1,5-*a*]quinazolin-8-ones **30a,b**



Scheme 9. Synthetic pathway to 3,6-diamino-11,12-dihydropyrimido[5",4":5',6']-pyrido[4',3':3,4]pyrazolo[1,5-*a*]quinazolin-9(10*H*)-ones **33a,b**

