

Postulates of Quantum Theory

- The state of a system is defined by a function (usually denoted Ψ and called the wavefunction or state function) that contains all the information that can be known about the system.
- Every physical observable is represented by a linear operator called the “Hermitian” operator.
- Measurement of a physical observable will give a result that is one of the eigenvalues of the corresponding operator for that observable.

Postulate I

$\Psi(x, y, z) \equiv$ stationary state

Often, Ψ is complex: $\Psi = Z(a, b)$

where: $a = \text{Real}(Z)$ & $b = \text{Imag}(Z)$

For example:

$Z \equiv a + ib$ in which

Complex Conjugate $\equiv Z^* = a - ib$

$i =$ complex number unit vector

$$i^2 = -1 ; i^3 = -i ; i^4 = 1 ; \frac{1}{i} = -i .$$

$$Z \cdot Z^* = |Z|^2 = a^2 + b^2$$

Properties of Valid Wave Functions

Boundary conditions

- 1) In order to avoid infinite probabilities, the wave function must be finite everywhere.
 - 2) In order to avoid multiple values of the probability, the wave function must be single valued.
 - 3) For finite potentials, the wave function and its derivative must be continuous. This is required because the second-order derivative term in the wave equation must be single valued. (There are exceptions to this rule when V is infinite.)
 - 4) In order to normalize the wave functions, they must approach zero as x approaches infinity.
- Solutions that do not satisfy these properties do not generally correspond to physically realizable circumstances.

$\Psi^* \cdot \Psi \equiv$ Probability Distribution Function

Criteria of Ψ

- must be single - valued over all space
- must be finite & continuous over all space
- $|\Psi|^2$ must have a finite integral over all space

Normalization of Ψ

$$\int_{\text{all space}} \Psi^* \cdot \Psi \cdot d\tau = 1$$

$$\text{(1D - cartesian) : } d\tau = dx$$

$$\text{(3D - cartesian) : } d\tau = dx \cdot dy \cdot dz$$

$$\text{(3D - spherical) : } d\tau = r^2 \cdot dr \cdot \sin \theta \cdot d\theta \cdot d\phi$$

Probability of finding particle within a certain region of space (1D - x only):

$$\text{Prob}(x_1 \leq x \leq x_2) = \frac{\int_{x_1}^{x_2} |\psi|^2 dx}{\int_{-\infty}^{\infty} |\psi|^2 dx} \xrightarrow{\text{if normalized}} \int_{x_1}^{x_2} |\psi|^2 dx$$

Postulate II

Variable	CM	QM
Position	x, y, z	$\hat{x}, \hat{y}, \hat{z}$
Momentum	mv	$\hat{p}_x = -i\hbar \frac{\partial}{\partial x} = \frac{\hbar}{i} \frac{\partial}{\partial x}$
		$\hat{p}_y = -i\hbar \frac{\partial}{\partial y} = \frac{\hbar}{i} \frac{\partial}{\partial y}$
		$\hat{p}_z = -i\hbar \frac{\partial}{\partial z} = \frac{\hbar}{i} \frac{\partial}{\partial z}$
Potential Energy	V	$\hat{V}(x, y, z) = V(x, y, z)$

Variable	CM	QM
Kinetic Energy	$T = \frac{1}{2}mv^2 = \frac{p^2}{2m}$	$\hat{T} = -\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] = -\frac{\hbar^2}{2m} [\nabla^2]$
Total Energy	$T + V$	$\hat{H} = -\frac{\hbar^2}{2m} \nabla^2 + V(x, y, z)$

$\hat{H} \equiv$ Hamiltonian

Starting Point in all Quantum Mechanical Problems.

Expectation values

- The **expectation value** is the expected result of the average of many measurements of a given quantity. The expectation value of x is denoted by $\langle x \rangle$
- Any measurable quantity for which we can calculate the expectation value is called a **physical observable**. The expectation values of physical observables (for example, position, linear momentum, angular momentum, and energy) must be real, because the experimental results of measurements are real.
- The average value of x is

$$\bar{x} = \frac{N_1x_1 + N_2x_2 + N_2x_2 + N_2x_2 + \dots}{N_1 + N_2 + N_3 + N_4 + \dots} = \frac{\sum_i N_i x_i}{\sum_i N_i}$$

Continuous Expectation values

- We can change from discrete to continuous variables by using the probability $P(x,t)$ of observing the particle at a particular x .
- Using the wave function, the expectation value is:
- The expectation value of any function $g(x)$ for a normalized wave function:

$$\overline{x} = \frac{\int_{-\infty}^{\infty} xP(x) dx}{\int_{-\infty}^{\infty} P(x) dx}$$

$$\langle x \rangle = \frac{\int_{-\infty}^{\infty} x\Psi^*(x,t)\Psi(x,t) dx}{\int_{-\infty}^{\infty} \Psi^*(x,t)\Psi(x,t) dx}$$

$$\langle g(x) \rangle = \int_{-\infty}^{\infty} \Psi^*(x,t)g(x)\Psi(x,t) dx$$

The uncertainty

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

Δx , is often called the *uncertainty* in position

In general for any quantity it given by

$$\Delta Q = \sqrt{\langle Q^2 \rangle - \langle Q \rangle^2}$$

The Schrodinger Wave Equation

- The Schrödinger wave equation in its time-dependent form for a particle of energy E moving in a potential V in one dimension is

$$i\hbar \frac{\partial \Psi(x,t)}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V\Psi(x,t)$$

- The extension into three dimensions is

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \left(\frac{\partial^2 \Psi}{\partial x^2} + \frac{\partial^2 \Psi}{\partial y^2} + \frac{\partial^2 \Psi}{\partial z^2} \right) + V\Psi(x,y,z,t)$$

where $i = \sqrt{-1}$ is an imaginary number.

Time-Independent Schrodinger Wave Equation

- The potential in many cases will not depend explicitly on time.
- The dependence on time and position can then be separated in the Schrödinger wave equation. Let

$$\Psi(x,t) = \psi(x) f(t)$$

which yields:

$$i\hbar \psi(x) \frac{\partial f(t)}{\partial t} = -\frac{\hbar^2 f(t)}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x) \psi(x) f(t)$$

Now divide by the wave function:

$$i\hbar \frac{1}{f(t)} \frac{df(t)}{dt} = -\frac{\hbar^2}{2m} \frac{1}{\psi(x)} \frac{d^2 \psi(x)}{dx^2} + V(x)$$

- The left side of the above Equation depends only on time, and the right side depends only on spatial coordinates. Hence each side must be equal to a constant. The time dependent side is

$$i\hbar \frac{1}{f} \frac{df}{dt} = B$$

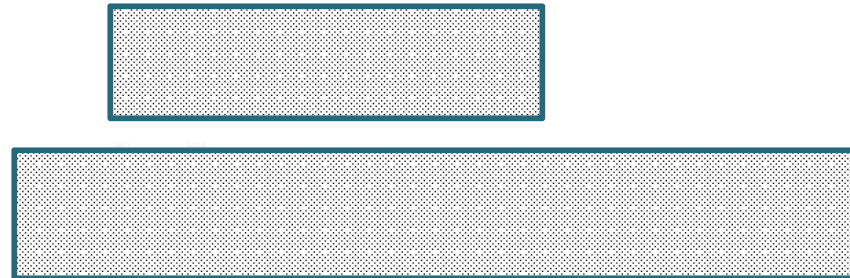
- We integrate both sides and find:
$$i\hbar \int \frac{df}{f} = \int B dt \quad i\hbar \ln f = Bt + C$$

where C is an integration constant that we may choose to be 0. Therefore

$$\ln f = \frac{Bt}{i\hbar}$$

This determines f to be
$$f(t) = e^{Bt/i\hbar} = e^{-iEt/\hbar}$$

$$B = E$$



- This is known as the **time-independent Schrödinger wave equation**, and it is a fundamental equation in quantum mechanics.

Stationary State

- The wave function can be written as: $\Psi(x,t) = \psi(x)e^{-i\omega t}$

$$\omega = \frac{E}{\hbar}$$

- The probability density becomes:

$$\Psi^* \Psi = \psi^2(x)(e^{i\omega t} e^{-i\omega t})$$

$$\Psi^* \Psi = \psi^2(x)$$

- The probability distributions are constant in time. This is a standing wave phenomena that is called the stationary state.